Atomic Physics for X-ray Astronomy at high spectral resolution

Julia C. Lee Harvard University

with thanks to Frits Paerels, Randall Smith for some slides and verbiage!

The Golden Age of X-ray Satellites



Chandra

XMM-Newton



X-ray band: $\sim 0.1 - 100 \text{ keV}$

Extant: Suzaku, RXTE - 2009?, (BeppoSAX - 2003), Planned: Astro-H, IXO (Gamma Ray: Swift, Integral) ²

The Golden Age of X-ray Satellites



In addition to on-board CCD Spectrometers:

- Chandra H(L)ETGS: High (Low) Energy Grating Spectrometers
- XMM-Newton RGS: Reflection Grating Spectrometer

3

Chandra



ISIM

Chandra



HETGS: PI: Canizares (MIT) LETGS: PI: Brinkmann? (Netherlands)

ISIM

Chandra



Spectral Resolution is key



MEG vs. HEG



Example based on lines detected in the Chandra HETGS

X-ray Spectroscopy in the 21st Century era of Chandra and XMM-Newton high spectral resolution instruments

X-ray Spectroscopy
Past and Present

ASCA SIS (R ~ 10 - 50) 5 data/(intrinsic power-law) broad Au edge Fe line feature instrumental) 0.5 OVII/OVIII edges (warm absorber) 0 5 2 Energy (keV) Fabian et al.

Chandra HETGS (R ~ 200 - 1000)



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X-ray Spectroscopy
Past and Present

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2

Energy (keV)

5

0

Fabian et al.

Chandra HETGS (R ~ 200 - 1000)



There are lots of Physics in the lines

Spectrum provides precise information on 'local' conditions in the plasma

element, ionization stage, ionization mechanism, temperature, density, degree of relaxation and past history, velocities (thermal, turbulent, bulk, nonthermal), *E* and *B* fields, ...









no atomic physics @ >10 keV

Fig. from F. Paerels

]	The Elec	ctromag	netic	Spectru	IM
	Wav	e-Particle Dua	ality: E =	$h_{\rm V} = hc/\lambda$	
Wavelength	Light waves = f	luctuation of E&B	fields due to	atomic & molecul	ar interactions
λ	1 m	1 mm	1 µm	1 nm	1 pm
Energy σ ν Ε	33 MHz		2m ⁻¹ 10 ⁴ cm ⁻¹ 33 THz 1.2 eV	10 ⁶ cm ⁻¹ 1.2 keV	
Name	Radio wave Medium Short	Microwave In Far IR	fra-red Near IR	Ultra-violet X-ray VUV XUV	γ∽ ray
Atomic and molecular phenomena	Atomic transistions Molecular rotation Ionic transitions Molecular vibration-rotation Inner shell transitions Mol. electron transitions				
Transparent materials	Air Glass Quartz LiF, MgF ₂ , CaF ₂				

Figure from "Spectrophysics: Principles & Applications" by Thorne, Litzen, Johansson

The Electromagnetic Spectrum										
		Light wa	Wav	e-Partic	e Dua	ality:	E =	$= \mathbf{h}\mathbf{v} = \mathbf{h}\mathbf{c}/\lambda$	cula	r interactions
Wavelen	gth λ	Digite we	1m	1	mm	۳۵۱ Inolas ado ۱ سا	n	1 nm	J	1 pm
Energy	σ ν E	33	MHz	1 cm ⁻¹ 33 GHz	10 ² o	m ⁻¹ 10 ⁴ o 33 THz 1.2	om ^{:1} eV	10 ⁶ cm ⁻¹ 1.2 ke	v	1.2 MeV
Name		Radio Medium	wave Short	Microwave	In Far IR	fra-red Near IR	Visible	Ultra-violet X- VUV XUV	ray	γ− ray
Atomic and molecular phenome	, na			Molecular rot	ation iolecular vi	Atomic tran i bration-rotati Mol. electro	sisti onic on on tr	ons transitions Inner shell transit ansitions	lions	Atomic, Ionic, Inner shell
materials	ent	Glass Glass Quartz LiF, MgF ₂ , CaF ₂								

Figure from "Spectrophysics: Principles & Applications" by Thorne, Litzen, Johansson

The Structure of the Atom Terminology for this lecture



Smallest component of an element having properties of that element

- subatomic particles @ nucleus
 - neutron: n⁰
 - proton: p⁺ = Atomic# Z
- elementary particle surrounding nucleus
 - electron: e⁻

Terminology: The Structure of the Atom



ATOM: smallest component of an element having properties of that element, made up of subatomic particles nucleons = $p^+ + n^0$ surrounded by e^-

- Ionization: physical process by which an e⁻ is removed (or added by capture) to an atom
 - net (+) via absorption
 - net (-) via collisions
- **lon**: atom with:
 - \geq Ie⁻ missing -> net (+)
 - ≥ Ie⁻ added -> net (-) (less common)
- Molecules:
 - > I atom forming bond

Plasma constituents: electrons, ions, photons

Plasma physics Interests: ion interactions with an electron or photon, which result in changes either in the state of the ionization or excitation of the interacting ion



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ions electrons photon (light)



ELECTRONS are accelerated by close approaches to IONS

As the temperature rises, the PHOTON (light) production increases

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Classification of Ionic Processes

(there are other schemes)

- * Spontaneous ion decay
- * Photon-ion Processes
- * Electron-ion interactions
- * Ion-ion interactions (chg xchg)

* Photon-electron processes

Direct Process	Inverse Process	
Spontaneous Decay	Resonant Photoabsorption	
Autoionization	Dielectron Recombination	
Photoionization	Radiative Recombination	
Electron Impact Ionization	3-body recombination (electron impact recombination)	
Electron Impact Excitation	Electron Impact De-excitation	
Bremsstrahlung	Inverse Bremsstralung	

Photon-Ion Interaction

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No particle interaction

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Autoionization	Dielectron Recombination		
Photoionization	Radiative Recombination		
Electron Impact Ionization	3-body recombination (electron impact recombination)		
Electron Impact Excitation	Electron Impact De-excitation		
Bremsstrahlung	Inverse Bremsstralung		

Photon-Electron Interaction (free-free) (Continuum Spectrum - no lines)

Direct Process	Inverse Process	
Spontaneous Decay	Resonant Photoabsorption	
Autoionization	Dielectron Recombination	
Photoionization	Radiative Recombination	
Electron Impact Ionization	3-body recombination (electron impact recombination)	
Electron Impact Excitation	Electron Impact De-excitation	
Bremsstrahlung	Inverse Bremsstralung	

And now for some words on Proton collisions.

In equilibrium, since protons are 1836x more massive than electrons, their speed will be \sim 43x slower than that of electrons. As a result, the collision rate <n σ v> for protons will be much lower. In addition, protons are positively charged, as are ions, so that also means small impact parameter (high momentum transfer) collisions are suppressed.

As a result, proton collisions can *usually* be ignored - except for low-lying transitions within an ion's ground state, which can be excited by proton collisions.



Some Quantum Mechanics Laws



- Strict Laws governing quantization of energy levels
- Arrangement of permitted orbits depends primarily on charge of nucleus which depend on number of protons
 - each element (and their isotopes) have similar orbits
 - different elements have different orbits
 - ionized atoms have different orbits from unionized ones



Emission: light bands







Fermi's Golden Rule

Transition Rate / Cross section = a measure of the number of reactions undergone by one particle per unit time / probability for transition







Atom absorbs photon = $|E_n - E_{n+1}|$: $A_1 \rightarrow A_u$



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Plasmas in Astrophysics of X-ray Interest

External ionizing source



• $k_B T_e \sim Ionization$ energy of plasma ions • k_BT_e << Ionization energy of plasma ions
Collisional (Thermal) Plasma

Astrophysical **collisional** plasmas come in two types:

- Coronal / Nebular: $N_e < 10^{14} 10^{16} \text{ cm}^{-3}$
- Collisional-Radiative: $10^{14} \text{ cm}^{-3} < N_e < 10^{27} \text{ cm}^{-3}$

In the more common Coronal (or Nebular) plasma, collisions excite ions but rarely de-excite them; the decay is radiative. This is also called the "ground-state" approximation, as all ions are assumed to be in the ground-state when collisions occur

In a CR plasma, collisions compete with photons in de-exciting levels; a level with a small oscillator strength (transition rate) value may be collisionally de-excited before it can radiate.

Photoionized Plasma

What happens when an external photon source illuminates the gas?

- The photons ionize the atoms in the gas.
- The photoelectrons created in this way collide with ambient electrons (mostly) and heat the gas n (or P), abundance
- The gas cools by radiation
- The gas temperature adjusts so that the heating and cooling balance

In a photoionized gas the *temperature* is not a free parameter and

The *ionization balance* is determined by the shape and strength of the *radiation field*

Photoionized Plasmas

Parameter definitions:

$$\xi \equiv \frac{L}{n_e R^2} \quad \text{Tarter, Tucker \& Salpeter (1969)}$$
$$U_x \equiv \frac{N_X}{4\pi R^2 nc} \quad \text{Davidson (1974)}$$
$$\Gamma \equiv \frac{L_X(> 13.6 \text{ eV})}{8\pi R^2 nc} \quad \text{Kwan \& Krolik (1981)}$$
$$\Xi \equiv \frac{L}{4\pi n_e ck T R^2} \quad \text{Krolik, McKee \& Tarter (1982)}$$
$$U_1 \equiv \frac{N}{4\pi R^2 nc} \quad \text{Netzer (1994)}$$

where:

$$L \equiv \int_{13.6 \text{ eV}}^{\infty} L(E) dE \quad N \equiv \int_{13.6 \text{ eV}}^{\infty} L(E) \frac{dE}{E} \quad N_X \equiv \int_{100 \text{ eV}}^{\infty} L(E) \frac{dE}{E}$$

Highlights: Coronal vs. Photoionized

- Coronal Gas:
 - need $kT_e \sim \Delta E$ to collisionally excite lines.
- Photoionized Gas:
 - External ionizing source
 - Temperature lower for same ionization than coronal, T~0.1 E_{th}/k
 - temperature not a free parameter; depends on spectral shape
 - Excitation is often by recombination cascade
 - Also get recombination continua (RRCs) due to recombination by cold electrons directly to the ground state. The width of these features is directly proportional to temperature
 - Due to the democratic ionization balance, it is more likely that diverse ions such as N VII, O VIII, Si XIV can coexist and emit efficiently than it would be in a coronal gas
 - Inner shell ionization and fluorescence is also important in gases where the ionization state is low enough to allow ions with filled shells to exist.

Ionization

Ionization is a relatively simple process, with three main channels:

Photoionization: $\gamma + I \rightarrow I^+ + e^-$ Collisional ionization: $e^- + I \rightarrow I^+ + 2e^-$ Inner-shell ionization: γ , $e^- + I \rightarrow I^{*+} + 2e^- \rightarrow I^+ + e^-$, γ

Photoionization and collisional ionization have been measured in the lab and theoretical calculations are relatively straightforward, so little attention is paid to them now.

Plasmas in Astrophysics of X-ray Interest

	Photoionized	Coronal				
Dominant ionization	Photoionization hν+Z _>Z+1	Electron impact e ⁻ +Z -> Z+1				
Examples	Active galaxies(AGN) SNR H II regions	Stellar coronae Supernova remnant Clusters of galaxies				
Spectral signature	Absorption,bound- free, bound-bound Emission: recombination	Emission lines, Δ n= 0,1,2 favored				

Table from R. Smith

There are lots of Physics in the lines

- Line position : ion identification
- Line shifts: dynamics, e.g. inflow, outflow
- Line strengths
 - ionic column : e.g. N_{OVII}, N_{NeX}, N_{FeXXVI} ...
 - degree of ionization : e.g. H-like/He-like
 - temperature, density
- Line widths : e.g. turbulence, GR

There are lots of Physics in the lines

First, a small digression

The isoelectronic sequence (a slight digression on terminology)

- All ions with the same number of electrons have similar internal structure
 - H ~ H-like CVI (+5), OVIII (+7), NeX, MgXII, Si XIV
 - He ~ He-like CV (+4), OVII (+6), NeIX, MgXI, SiXIII ...
 - to 1st order, electronic states differ only by scaling factor:
 - energies larger by Z^{2}_{eff}
 - radii smaller by 1/Z_{eff}

H-like : All transitions of astrophysically abundant metals $(C \rightarrow Ni)$ are in the X-ray band. Ly α /Ly β is a useful temperature diagnostic; Ly α is quite bright.

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H-like : All transitions of astrophysically abundant metals $(C \rightarrow Ni)$ are in the X-ray band. Ly α /Ly β is a useful temperature diagnostic; Ly α is quite bright.

He-like: $\Delta n \ge 1$ transitions are all bright and in X-ray. The $n=2 \rightarrow 1$ transitions have 4 transitions which are useful diagnostics, although R=300 required to separate them.

Ne-like: Primarily Fe XVII; two groups of bright emission lines at 15Å and 17Å; ionization state and density diagnostics, although there are atomic physics problems.

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 - temperature, density
- Line widths : e.g. turbulence, GR

Recall Quantum Mechanics Laws



н			-	-													He
³ Li	⁴ Be											5 B	⁶ C	7 N	⁸ O	9 F	10 Ne
11 Na	12 Mg											13 AI	¹⁴ Si	15 P	16 S	¹⁷ CI	18 Ar
¹⁹ K	20 Ca	21 Sc	22 Ti	²³ V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	³⁸ Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 	54 Xe
55 Cs	56 Ba	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 TI	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111	112	113	114	115	116	117	118
				57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb
				89 Ac	90 Th	91 D a	92	93	94 Du	95 Am	96 Cm	97 Bk	98 Cf	99 Ec	100	101 Md	102

- Strict Laws governing quantization of energy levels
- Arrangement of permitted orbits depends primarily on charge of nucleus which depend on number of protons
 - each element (and their isotopes) have similar orbits
 - different elements have different orbits
 - ionized atoms have different orbits from unionized ones

Ions in Plasma : Line Position

Each element has unique set of permitted orbits depending on # of protons





X-rays Sensitive to log ξ : -4 to 4 !!!

Seyfert 2 NGC 1068 : XMM RGS Kinkhabwala et al. 2002

Seyfert 1 NGC 3783 : Chandra HETGS Kaspi et al. 2000, 2002







For calculations : Behar, Sako, Kahn 2001; & Flexible Atomic Code (FAC) : Gu 2003

Absorption

Unresolved Transition Arrays (UTAs)

- Appears in absorption spectra of AGN, eg. NGC 3783
- Comes from 2p-3s or 2p-3d
 transitions --> requires iron
 less than 9 times ionized
- Potential diagnostic of ionization balance



(Behar, Sako and Kahn 2002)

Innershell OI - OV



Absorption

K shell Photoabsorption (Oxygen)



In theory, could diagnose ionization balance in the ISM or other absorbing material. This data uses semi-empirical corrections to energy levels in the optimization of wavefunctions, based on the experiment, plus multicode approach

Red: Pradhan et al (2003) Green: Verner and Yakovlev (1995) Black: Garcia et al. (2005)

The KLL (1s2s2p) Resonance of O VI



X-ray Fluorescence Lines from photoionized plasmas

Hard X-ray continuum in relatives cool (T~ 10^5 - 10^6 K) matter



Diagram from Liedahl & Torres (2005)

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 - temperature, density
- Line widths : e.g. turbulence, GR

Dynamics : Inflow / Outflow Line (Doppler) Shifts

С





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Line strengths: absorption

R

Instructive example: transmission of continuum thru uniform slab

$$F_{\text{line}} = F_{\text{cont}} \exp(-\tau_{\nu})$$
 Optical depth

$$\tau_{\nu} = N_i \sigma_{\nu} = \frac{\pi e^2}{m_e c} f_{12} \phi_{\nu}; \quad \int \phi_{\nu} d\nu \equiv 1$$
Ion column density (cm⁻²)
(cf. Rybicki and Lightman,
Radiative Processes in
Astrophysics)
Optical depth
Optical dept

Line strengths: absorption

EW as function of column density: 'curve of growth'



Analytic behavior easy to understand:

- τ_{v0} << 1: EW linear in N_i (all atoms absorb)
- τ_{v0} ≈ 1: saturation; EW stalls at the Doppler width
- τ_{v0} >> 1: damping wings become optically thick; EW goes as N_i^{1/2}

Beware of saturation! If unresolved, lines look normal, but are in fact black

He-like Ions in Emission

Two electrons: can have $|\uparrow\uparrow\uparrow\rangle$, $|\downarrow\downarrow\downarrow\rangle$, $|\uparrow\downarrow\downarrow\rangle$, $|\downarrow\uparrow\uparrow\rangle$. But these are not all eigenstates of *J*. These ones are:

 $|\uparrow\uparrow\rangle$, $|\downarrow\downarrow\rangle$, $|\uparrow\downarrow\rangle$ + $|\downarrow\uparrow\rangle$ (symmetric in the spins, total spin 1) 'triplet' $|\uparrow\downarrow\rangle$ - $|\downarrow\uparrow\rangle$ (antisymmetric in the spins, total spin 0) | 'singlet'



- Resonance (w) line
 - allowed electric dipole transitions involving ground state
- Intercombination (xy):
 - triplet (S=1) to singlet (S=0) state transition
 - S=0: no fine structure
 - S=1: 3 possible orientations of spin angular momentum
 - "spin forbidden"; usually seen for large Z, with large spin-orbit interaction; off-diagnoal matrix elements resulting in mixed eigenfunctions
- Forbidden (z)
 - a transition between 2 levels that is not allowed by selection rules for electric dipole radiation, i.e LS forbidden
 - metastable: states for which transitions from upper -> all lower levels are forbidden (transition through E transfer to other particle through collisions)
 - probability w-transitions rapidly increases with Z in hot plasmas
 - observed only when collisional de-excitation probability is low: e.g. astrophysical plasmas in solar corona & nebulae, and in some low density plasma in thermonuclear fusion devices



• Photo-ionization (e.g., AGN, X-ray binaries), out-of-equilibrium (e.g., SNR)



Gabriel & Jordan (1969): He-like Diagnostics

Density: $R(n_e) = Forbidden / Intercombination = z / (x+y)$

Temperature: $G(T_e) = (F + I) / Resonance = z + (x+y) / w$

Density Diagnostics with R = F / I



Helium-like Lines

Density Diagnostics with R = F / I

If n_e is large enough, collisions move electrons from the forbidden to the intercombination and resonance levels.



Ines

like

Helium-J

Temperature Diagnostics with G=(F+I)/R

Why does the G ratio measure temperature and ionization state?


Temperature Diagnostics with G=(F+I)/R

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Temperature Diagnostics with G=(F+I)/RWhy does the G ratio measure temperature and ionization state?



Temperature Diagnostics with G=(F+I)/R



Helium-like Lines

Photoionized + Hybrid Plasmas for OVII



Plasma Diagnostics with He-like ions



Temperature: $G(T_e) = (F + I) / Resonance = z + (x+y) / w$

Be Careful! He Triplet: Not a hard and fast rule ...

Seyfert 2 Galaxy NGC 1068: Kinkhabwala et al. 2002; see also Ogle+ 03, Brinkman+ 03



- Photoexcitation (n=1->n_{i+1}) can mimic the "signatures" of collisionally ionized plasma
 - misleading G since R gets enhanced through recombination
- Solution:
 - Look @ higher order lines
 - photoexcitation will enhance high order lines more than allowed in CI plasma
 - Look for RRC
 - CI plasma: RRC smeared out: E(captured e⁻) < E(ion) so do not contribute much more E to atom
 - PI plasma: DV gives T: E(captured e⁻) >> E(ion) so contribute a lot more E

Line strengths: absorption

Instructive example: transmission of continuum thru uniform slab

$$F_{\text{line}} = F_{\text{cont}} \exp(-\tau_{\nu})$$

$$\tau_{\nu} = N_i \sigma_{\nu} = \frac{\pi e^2}{m_e c} f_{12} \phi_{\nu}; \quad \int \phi_{\nu} d\nu \equiv 1$$
Ion column density (cm⁻²)
Normalized profile
Oscillator strength (classically, f=1)
(cf. Rybicki and Lightman,
Radiative Processes in
Astrophysics)
Get this from classical damped harmonic oscillator!
Absorption/scattering cross section (cm²)

Line strengths: absorption

General profile: convolution of Lorentzian with Gaussian: Voigt Parameter: ratio of damping width to Doppler width



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 - temperature, density
- Line widths : e.g. turbulence, GR

Natural Broadening Due to finite lifetime of ionic/excited states



Arises from the uncertainty in energy of the states involved in the transition

Heisengberg's Uncertainty Principle

Not important in atomic spectra since lifetime of atomic transitions are T ~ 10⁻⁸ s

--> Natural LW ~ 6.6 x 10⁻⁸ eV

Single ion: Lorentzian, with characteristic width given by the lifetime of the upper level involved in the transition ('natural broadening', or 'damping')

$$\phi_{\nu} = \frac{\Gamma/4\pi^2}{(\nu - \nu_0)^2 + (\Gamma/4\pi)^2}$$

and any other process that limits the lifetime (e.g. collisions)

Thermal & Turbulent Broadening



Thermal Broadening: Atoms in constant thermal motion in cloud of gas, which "smear" line in Gaussian profile

Thermal & Turbulent Broadening



Turbulent Broadening: Exists when a cloud is not at rest / flowing smoothly, but churning in eddies and vortices: motion cause (random) Doppler shifting of spectral lines from different parts of cloud that cannot be discerned by telescope

ions

Thermal Broadening: Atoms in constant thermal motion in cloud of gas, which "smear" line in Gaussian profile

Thermal & Turbulent Broadening



Turbulent Broadening: Exists when a cloud is not at rest / flowing smoothly, but churning in eddies and vortices: motion cause (random) Doppler shifting of spectral lines from different parts of cloud that cannot be discerned by telescope

Broadening by radial velocities of ions, e.g. thermal velocities: Gaussian profile

$$\phi_{\nu} = \frac{1}{\Delta \nu_D \sqrt{\pi}} \exp \left(-(\nu - \nu_0)^2 / \Delta \nu_D^2\right)$$

$$\Delta\nu_D/\nu_0 = \left(\frac{2kT}{m_ic^2}\right)^{1/2}$$

and any other process that produces a velocity broadening (e.g. turbulence)

Thermal Doppler width

Line strengths: absorption

Line profile:

Single ion: Lorentzian, with characteristic width given by the lifetime of the upper level involved in the transition ('natural broadening', or 'damping')

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 $\Delta
u_D /
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ight)^{1/2}$

and any other process that produces a velocity broadening (e.g. turbulence)

Thermal Doppler width

How you model your lines will affect your flux and therefore determination of plasma column density.

Line strengths: absorption

General profile: convolution of Lorentzian with Gaussian: Voigt Parameter: ratio of damping width to Doppler width



From Carroll and Ostlie: Introduction to Modern Astrophysics

How you model your lines will affect your flux and therefore determination of plasma column density.

Line strengths: absorption

EW as function of column density: 'curve of growth'



Analytic behavior easy to understand:

- τ_{v0} << 1: EW linear in N_i (all atoms absorb)
- τ_{v0} ≈ 1: saturation; EW stalls at the Doppler width
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Beware of saturation! If unresolved, lines look normal, but are in fact black

Rotational (Doppler) Broadening



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Extending this to accretion disks in the Newtonian limit



http://jilawww.colorado.edu/research/images/accretion.jpg



Fig. from imagine.gsfc.nasa.gov

Rotational Broadening in Accretion Disks



Iron line Profiles near a Black Hole



<u>REVIEWS</u>:

Fabian et al. 2000 Reynolds & Nowak 2003



Relativistic effects imprint characteristic profile on the emission line...



Schwarzchild vs. Kerr Black Holes



Credit : NASA/CXC/M.Weiss; Spectra: NASA/CXC/SAO/J.Miller et al.

Accretion Disk Surrounding a non-rotating (Schwarzschild) black hole

Simulations : P Armitage & C. Reynolds 2003

Turbulent Accretion Disk Surrounding a non-rotating (Schwarzschild) black hole

i = 30 deg

Simulations : P Armitage & C. Reynolds 2003

Turbulent Accretion Disk Surrounding a non-rotating (Schwarzschild) black hole

E --->



Observations of relativistic emission lines

- First seen in 1994 with ASCA observatory
- 5 day observation of Seyfert-1 galaxy MCG-6-30-15
- Needed long observation to collect enough photons to form detailed spectrum



Power-law continuum subtracted ASCA: Tanaka et al. (1995)





Dynamics from line widths: From Atomic motions to General Relativity



Motions of atoms seen in linewidths subject plasma conditions

Turbulence and GR Effects

An X-ray Perspective on Determining Dust Composition JCL with J. Xiang (Harvard), B. Ravel (NRL, ANL, NIST), J. Kortright (LBNL)



The theory behind measuring X-ray Absorption Fine Structure (XAFS) to determine molecular/dust composition

- The photoelectric effect : X-ray photon absorbed by an electron in a tightly bound quantum core level (e.g. 1s or 2p)
 - Bound free continuum absorption \rightarrow edge step
 - Isolated Atoms: Bound-bound process -> ionized lines
 - Molecule : bound-bound process → XAFS

The theory behind measuring X-ray Absorption Fine Structure (XAFS) to determine molecular composition

initial state : an X-ray, a core electron, no photo-election Fermi's Golden Rule : $\mu(E) = \mu_0(E)[1 + \chi(E)] \propto |\langle i | H | f \rangle|^2$ final state : no X-ray, a core hole, a photo-election depends ONLY on Fine-Structure Term : change in photo-electron final state absorbing atom due to back-scattering from neigboring atom $\chi(E) = \frac{\mu(E) - \mu_0(E)}{\Delta \mu_0(E)} \propto \langle i | H | \Delta f \rangle$ H: interaction term - represents changing between 2 energy, momentum states The XAFS equation : represents interaction of forward scattering & backscattering photoelectron $\chi(\mathbf{k}) = \sum_{\mathbf{j}} \frac{\mathbf{N_j S_0^2 f_j(\mathbf{k}) e^{-2\mathbf{R_j}/\lambda(\mathbf{k})} e^{-2\mathbf{k}^2 \sigma_j^2}}}{\mathbf{k R_j}^2} \sin[2\mathbf{k R_j} + \boldsymbol{\delta_j(\mathbf{k})}]$ Single Scattering Approximation

XAFS Theory Bound-bound case for CM (dust+molecules)



Figures from "XAFS" : © 2002 Matt Newville

The amplitude of the back-scattered photo-electron at the absorbing atom will vary with energy --> oscillations in $\mu(E) --> XAFS$

X-ray Absorption Spectra in a Nutshell



Dust will imprint structure near photoelectric edges



For details on theory & astrophysics applications: Lee & Ravel 2005 Lee et al. 2009, in press


Using Quantum Mechanics to do Astrophysics

- Atoms can be identified by their unique set of spectral lines
- Relative line strengths dependent on temperature, density
- Emission/Absorption: dependent on viewing geometry
- Use these properties to figure out dynamics



Gas temperature and viewing angle dictates what we see:

- continuum radiation,
- emission lines, or
- absorption lines

Conclusions

Although moderately complex, there are relatively few processes that dominate X-ray emission; analyzing the observed spectrum from each can reveal the underlying parameters. These processes are:

- Line emission
 - Collisional \Rightarrow temperature, abundance, density
 - Photoionized \Rightarrow photoionization parameter, abundance, density
- Absorption \Rightarrow abundance, density, velocity
- Continuous Spectrum

Synchrotron emission \Rightarrow relativistic electrons, magnetic field Inverse Compton scattering \Rightarrow relativistic electrons Blackbody \Rightarrow temperature, size of emitting region / distance²

Plasma Codes

Understanding a collisional plasma requires a collisional plasma model. Since even a simple model requires considering hundreds of lines, and modern codes track millions, most people select one of the precalculated codes:

Code	Source
Raymond-Smith	ftp://legacy.gsfc.nasa.gov/software/plasma_codes/raymond
SPEX	http://saturn.sron.nłgenerałprojects/spex
Chianti	http://wwwsolar.nrl.navy.miłchianti.html
ATOMDB	http://cxc.harvard.edu/ATOMDB

The calculated spectrum is also known as APEC, and the atomic database is called APED.

From Randall Smith Lecture @ xrayschool.gsfc.nasa.gov

The collisional plasma models available in XSPEC or Sherpa are:

apec	ATOMDB code; good for high-resolution data
raymond	Updated (1993) Raymond-Smith (1977) code
meka	Original Mewe-Kaastra (Mewe et al. 1985) code; outdated
mekal	Mewe-Kaastra-Liedahl code (Kaastra 1992); new Fe L lines
c6mekal	mekal with an polynomial EM distribution
equil	Borkowski update of Hamilton, Sarazin & Chevalier (1983)
nei	Ionizing plasma version of equil
sedov	Sedov (SNR) version of equil
pshock	Plane parallel shock version of equil

Variable abundance versions of all these are available.

Individual line intensities as functions of T, n, etc. are not easily available (yet) in either XSPEC or Sherpa.

Photoionized Plasma Codes

Understanding a photoionized plasma requires a plasma model plus a physical model of the system:

- Illuminated slab of gas
- Torus with central source
- Disk with 'light bulb' above it
- Central source surrounded by small absorbers
- etc...

Code	Website
XSTAR	http://heasarc.nasa.gov/lheasoft/xstar/xstar.html
CLOUDY	http://www.nublado.org/
Titan	http://Vo.obspm.fr:8888/simulation/ (in progress?)
Mocassin	http://hea-www.harvard.edu/~bercolano/

Atomic Codes

HULLAC (Hebrew University / Lawrence Livermore Atomic Code) : Fast, used for many APED calculations, not generally available.

R-Matrix : Slow, used for detailed calculations of smaller systems of lines, available on request but requires months to learn.

FAC (Flexible Atomic Code) : Fast, based on HULLAC and written by Ming Feng Gu

From Randall Smith Lecture @ xrayschool.gsfc.nasa.gov

A few parting words

- There is a lot of Physics in the Lines
 - most probable transitions strongest
 - line blends common
- Think critically about model fits
 - Plasma codes have ~10% error on line strengths
 - Line energies differ depending on databases, especially more difficult transitions
- $\chi^2_n \sim 1$ **not** a useful gauge for a high resolution spectrum -- understand the physics