

Einführung in die Astronomie II

Jörn Wilms

Sommersemester 2007

Büro: Dr. Karl Remeis-Sternwarte, Bamberg Email: joern.wilms@sternwarte.uni-erlangen.de Tel.: (0951) 95222-13

http://pulsar.sternwarte.uni-erlangen.de/wilms/teach/intro2

Friedrich-Alexander-Universität Erlangen-Nürnberg



Astronomie an der FAU

NF im Vordiplom/BA: Gebraucht werden:

- Klausuren Astronomie I und II
- Astronomisches Praktikum (Schein)

Astronomie im Hauptstudium Physik: hängt davon ab...

PWB: 10 SWS weiterführende Vorlesungen Astro-/Teilchenphysik, davon 2 SWS Theorie

nichtphysikalisches Wahlfach: wie NF im Hauptstudium, nur wenn Astronomie nicht im Vordiplom!

Nebenfächler (NF im Hauptstudium für Nichtphysiker):

- Astronomie I und II
- Eine weiterführende Vorlesung (2 SWS)
- Ein physikalischer Praktikumsschein (z.B. Astronomisches Praktikum)

Frühstudium: freiwillig,

möglich sind Klausuren Astronomie I und II (⇒ Scheine)

Preliminaries

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Klausuren?!?

Bologna-Prozess ändert alles: politisch beschlossene Einführung von BA/MA Studiengängen.

Idee: Kumulative Abschlüsse, keine Prüfung am Ende.

An der FAU: Ab WS 2007/2008 BA , aber:

- 1. Wir haben (leider!!!) zu großen Andrang auf's Praktikum
- 2. Wir haben Frühstudierende, die BA studieren werden

3. Wir sollen Klausuren üben, bevor es Ernst wird...

\implies KLAUSUR am 10. Juli 2007

Klausuren werden zu einem *benoteten* Schein führen, auf Wunsch sind bei Bestehen der Klausur auch *unbenotete* Scheine möglich.



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Praktikum

Praktikum wird an der Dr. Karl Remeis-Sternwarte, Bamberg, als Blockpraktikum durchgeführt:

- 10.09.-21.09.2007: NF im Vordiplom/BA, Lehramt
- 24.09.-05.10.2007: NF im Vordiplom/BA, Lehramt

Endgültige Voraussetzung für eine Teilnahme am Praktikum sind normalerweise zwei Scheine aus Astronomie I und II.

⇒ Wir haben zur Zeit eine Warteliste. Zulassungen zum Praktikum werden in der Astronomie I im WS07/08 stattfinden. Die Klausur aus dieser Vorlesung kann als Teil der Zulassungsvoraussetzungen benutzt werden.

Preliminaries

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Textbooks

- KARTUNNEN, KRÖGER & OJA, 2003, *Fundamental Astronomy*, Heidelberg:
 Springer, €64 (softcover), 500 pp.
 Good general overview of astronomy.
 Recommended, especially for exam preparation.
- UNSÖLD & BASCHEK, 2004, *Der neue Kosmos. Einführung in die Astronomie und Astrophysik*, Berlin: Springer, € 50, 577 pp. Very good overview of stellar astronomy, weaker on extragalactic astronomy. Good secondary reading.
- DE PATER & LISSAUER, 2004, *Planetary Sciences*, Cambridge: Cambridge University Press, €93, 544 pp. *The* textbook of planetary science. Good secondary reading.

Literature

Contents

17 Apr Introduction, Reminders, Stellar evolution
24 Apr Stellar Evolution, II
01 May no lecture – May day
8 May End stages of stellar evolution
15 May Milky Way and Galactic Center
22 May Galaxies: Classification, properties
29 May no lecture – Pentecost
5 Jun Galaxies: Distances, Mass
12 Jun AGN, Galaxy clusters
19 Jun Cosmology: Expansion of the Universe
26 Jun Evolution of the Universe
10 Jul EXAM
17 Jul Wrap up

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Stellar Structure and Evolution: Reminder

Stellar Structure

Stellar structure governed by four coupled differential equations:

Mass structure (mass conservation)

$$\frac{\mathrm{d}M}{\mathrm{d}r} = 4\pi r^2 \rho(r)$$

Temperature structure (energy transport)

$$\frac{\mathrm{d}T}{\mathrm{d}r} = -\frac{3}{4ac}\frac{\kappa\rho(r)}{T^3}\frac{L(r)}{4\pi r^2}$$

(hydrostatic equilibrium)

Pressure structure

$$\frac{\mathrm{d}P}{\mathrm{d}r} = -\rho(r)\frac{GM(r)}{r^2}$$

Energy conservation (energy transport)

$$\frac{\mathrm{d}L}{\mathrm{d}r} = 4\pi r^2 \rho(r) \epsilon(r)$$

plus "equation of state" ($P = P(T, \rho)$), energy generation ($\epsilon = \epsilon(T, \rho, Z)$),...

Stellar model: numerical solution of stellar structure equations.

Stellar Structure

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Energy generation: Proton-Proton chain

 $^{1}\text{H} + ^{1}\text{H} \longrightarrow ^{2}\text{H} + e^{+} + \nu_{e}$

 $^{2}\text{H} + ^{1}\text{H} \longrightarrow ^{3}\text{He} + \gamma$

 ${}^{3}\text{He} + {}^{3}\text{He} \longrightarrow {}^{4}\text{He} + 2{}^{1}\text{H}$



Stellar Structure



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(2.1)

(2.2)

(2.3)

Solar Structure

Based on observations of

- Solar Mass: $1 M_{\odot} = 1.997 \times 10^{30} \text{ kg} = 1.997 \times 10^{33} \text{ g}$
- Solar Luminosity: $1 L_{\odot} = 3.127 \times 10^{26} \text{ W} = 3.127 \times 10^{33} \text{ erg s}^{-1}$
- Solar chemical composition (=elemental abundances): 75% H, 24% He, 1% metals (by mass)

it is possible to use the equations of stellar structure to determine a model for the structure of the Sun, i.e., M_r , L_r , $\rho(r)$, T(r), abundances(r). The best models available are called "standard models".

Stellar Structure



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Solar Structure



Solar Structure



Bahcall

SNO (2001): When taking all neutrino flavors into account, i.e., take into account that solar neutrinos change their flavor on the way from the Sun to us, the measured and predicted neutrino fluxes agree.

3 - 1Stars: Evolution

Stellar Evolution

Now that we believe that the solar model is correct: stellar evolution

Principle:

- 1. Construct stellar model by solving equations of stellar structure for given radial abundances.
- 2. Evaluate change in elemental abundances as a function of radius based on the local fusion processes.
- 3. Change abundances appropriately for a time step Δt .
- 4. goto step 1

3 - 2



Characteristic Timescales

Main sequence: Hydrogen burning at the center.

Evolution timescale dominated by the nuclear timescale = timescale needed to use the fuel in the center of the star.

According to simulations, this is \sim 10% of the available Hydrogen. Since 0.7% of $M_{\rm core}c^2$ converted into He, the nuclear timescale is

$$t_{\rm n} = \frac{0.007 \cdot 0.1 M c^2}{L} = \frac{M/M_{\odot}}{L/L_{\odot}} \cdot 10^{10} \,\text{years} \tag{3.1}$$

A second important timescale is the timescale the star would need to radiate its stored thermal energy: thermal timescale. Roughly given as

$$_{\rm t} = \frac{0.5GM^2/R}{L} = \frac{(M/M_{\odot})^2}{(R/R_{\odot})(L/L_{\odot})} \cdot 2 \times 10^7 \,{\rm years}$$
 (3.2)

Evolution of the Sun



 $t(10^9 \, {\rm years}) \longrightarrow$



Evolution of the Sun

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Evolution of the structure of a 1 M_{\odot} star on the main sequence (after Maeder & Meynet, 1989).







Solar Mass Stars: Post Main Sequence, III

Reminder: stars are in hydrostatic equilibrium: inwards gravitational pressure balanced by outwards gas pressure

Since the gas pressure is P = nkT: energy source needed to heat gas (=fusion).

This is a problem for the core during the red giant stage, as virtually no fusion ongoing

 \implies Core gets compressed

 $\implies \rho$ and T increase

BUT:

collapse cannot continue indefinitely!

 \implies once ρ has increased appreciably, there must be a point where quantum mechanical effects become important.

Evolution of the Sun

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Different ways to write the equation of state of an ideal gas

Among the more confusing subjects of thermodynamics are the many different ways N_{mol} : the number of moles of the gas in the volume Vin which the ideal gas equation can be written. P = nkT

The one I prefer for astronomy is

D

where

- P⁻ Pressure (measured in N m⁻²) n: particle density (i.e., number of particles per cubic meter, unit: m⁻³)
- k = 1.38066 × 10⁻²³ J K⁻¹: Boltzmann constant
- T: Temperature (measured in Kelvins)

This equation has the advantage that it counts all particles individually (thus using n). If you know the mass of the gas particles, $\mathit{m}_{\rm gas}$ then another way of writing the ideal gas equation is

$$P = \frac{nm_{gas}}{m_{gas}}kT = \rho kT \frac{1}{m_{gas}}$$

illustrating that for an ideal gas, $P\propto\rho,$ where ρ is the mass density

Another way to write the ideal gas equation is in terms of the total number of gas molecules, N = nV, where V is the volume. The ideal gas equation then is

$$=\frac{N}{V}kT \iff PV = NkT \iff \frac{PV}{T} = Nk$$

This version has the problem, however, that the number of gar molecules is typically rather large (there are 6×10^{23} molecules in a volume of 22.4 liters of gas, this number of particles is called one *mole*). Because working with smaller numbers is generally thought a good idea, chemists prefer to work with moles. Per definition, the unit of particle number here is the Avogadro number $N_{\rm A} = 6.0221 \times 10^{23}$. So, if you want to work with moles, then the above equation becomes

$$PV = \frac{N}{N_{A}}AkT = N_{mol}R$$

where

R = N_Ak8.3145 J mol⁻¹ K⁻¹: the universal gas constant

To summarise, each of these equations has its own uses, and which one you want to use, really depends on the circumstances of the problem you are solving. For your future life as physicists, try to remember one of them, and then understand how you get from this one to the others, instead of memorising all four ones. This approach will need less memory and lead to a better understanding of what is really going on behind the scenes.



QM interlude, I

Quantum mechanics: One of the stranger phenomena in QM is the Pauli exclusion principle:

For particles such as electrons ("Fermions"), at least one of their quantum numbers must be different.

Quantum numbers are, e.g.,

- position (x, y, z),
- momentum (mv_x, mv_y, mv_z) ,
- angular momentum,
- spin (s)

All of these numbers are "quantized", i.e., can only have discrete values (e.g., spin: +1/2, -1/2).

In a typical gas, this is not a problem ("phase space is (almost) empty") once it becomes dense \implies exclusion principle kicks in.

Evolution of the Sun

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Evolution of the structure of a 1 M_{\odot} star to the Helium flash (Maeder & Meynet, 1989).



planetary nebulae: material ejected during AGB phase, photoionized once remaining core of former star has shed enough mass to emit UV photons.





planetary nebulae: material ejected during AGB phase, photoionized once remaining core of former star has shed enough mass to emit UV photons.



NGC 6853/M27 ("Dumbbell Nebula"; ESO VLT/FORS)

planetary nebulae: material ejected during AGB phase, photoionized once remaining core of former star has shed enough mass to emit UV photons.



Hourglass Nebula (HST/Sahai/Trauger

planetary nebulae: material ejected during AGB phase, photoionized once remaining core of former star has shed enough mass to emit UV photons.



IC4406 (ESO VLT)

planetary nebulae: material ejected during AGB phase, photoionized once remaining core of former star has shed enough mass to emit UV photons.