

A Appendix

I Atomic data

Section 2.1 deals with the analysis of absorbed spectra. For the investigation of this absorption, some atomic data is necessary, which is compiled in this section.

A more detailed list can be found in form of S-Lang data structures in the `atomicData`-feature of the module `lineProfile`, which is available online at <http://pulsar.sternwarte.uni-erlangen.de/hanke/diplomathesis/code/>.

I.1 Bound-free transition edge-energies and ISM abundances

Photoionization processes as described in Sect. 2.1.1 can only occur for photon-energies above the ionization threshold of the specific atom. The latter is listed for the first 30 elements in Table A I.1. The K-edge (ionization energy for a $1s$ electron) is given as well as both L-edges (L_1 for the ionization of a $2s$ electron, L_2 for a $2p$ electron). Not every element produces a strong photoabsorption edge, as the optical thickness depends also on the atomic abundance (cf. Eqs. 2.3 and 5.1). To find the astrophysically relevant atoms, the abundances in the interstellar medium (as well as the solar abundances for comparison) are also included:

Table A I.1: Neutral K- and L-edge energies and wavelengths (Verner & Yakovlev, 1995) and relative abundances A_Z^{ISM} in the interstellar medium (Wilms et al., 2000), compared with the solar abundances A_Z^{\odot} (reviewed by Asplund et al., 2005).

$z(\text{element})$	$_1\text{H}$	$_2\text{He}$	$_3\text{Li}$	$_4\text{Be}$	$_5\text{B}$	$_6\text{C}$	$_7\text{N}$	$_8\text{O}$	$_9\text{F}$	$_{10}\text{Ne}$
E_K/keV	0.014	0.025	0.064	0.119	0.194	0.29	0.40	0.54	0.69	0.87
$\lambda_K/\text{\AA}$	912	504	193	104	63.9	42.6	30.6	23.0	17.9	14.25
E_{L_1}/keV			0.005	0.009	0.014	0.019	0.025	0.029	0.038	0.049
$\lambda_{L_1}/\text{\AA}$			2300	1330	882	639	488	435	327	256
E_{L_2}/keV					0.008	0.011	0.015		0.017	0.022
$\lambda_{L_2}/\text{\AA}$					1494	1101	853		712	575
$12 + \log A_Z^{\odot}$	12.0	10.93	1.05	1.38	2.70	8.39	7.78	8.66	4.56	7.84
$12 + \log A_Z^{\text{ISM}}$	12.0	10.99				8.38	7.88	8.69		7.94
$A_Z^{\text{ISM}}/A_{28}^{\text{ISM}}$	891 000	87 100				214	67.6	437		77.6
$z(\text{element})$	$_{11}\text{Na}$	$_{12}\text{Mg}$	$_{13}\text{Al}$	$_{14}\text{Si}$	$_{15}\text{P}$	$_{16}\text{S}$	$_{17}\text{Cl}$	$_{18}\text{Ar}$	$_{19}\text{K}$	$_{20}\text{Ca}$
E_K/keV	1.08	1.31	1.57	1.85	2.15	2.48	2.83	3.20	3.61	4.04
$\lambda_K/\text{\AA}$	11.49	9.46	7.91	6.72	5.76	5.01	4.38	3.87	3.43	3.07
E_{L_1}/keV	0.071	0.094	0.126	0.156	0.194	0.235	0.278	0.326	0.384	0.443
$\lambda_{L_1}/\text{\AA}$	175	132	98.7	79.5	63.9	52.8	44.6	38.0	32.3	28.0
E_{L_2}/keV	0.038	0.055	0.080	0.106	0.140	0.170	0.209	0.249	0.301	0.352
$\lambda_{L_2}/\text{\AA}$	325	226	154	117	88.6	72.9	59.3	49.8	41.1	35.2
$12 + \log A_Z^{\odot}$	6.17	7.53	6.37	7.51	5.36	7.14	5.50	6.18	5.08	6.31
$12 + \log A_Z^{\text{ISM}}$	6.16	7.40	6.33	7.27	5.42	7.09	5.12	6.41		6.20
$A_Z^{\text{ISM}}/A_{28}^{\text{ISM}}$	1.29	22.4	1.91	16.6	0.234	11.0	0.117	2.29		1.41
$z(\text{element})$	$_{21}\text{Sc}$	$_{22}\text{Ti}$	$_{23}\text{V}$	$_{24}\text{Cr}$	$_{25}\text{Mn}$	$_{26}\text{Fe}$	$_{27}\text{Co}$	$_{28}\text{Ni}$	$_{29}\text{Cu}$	$_{30}\text{Zn}$
E_K/keV	4.49	4.97	5.48	6.00	6.55	7.12	7.73	8.35	8.99	9.67
$\lambda_K/\text{\AA}$	2.76	2.49	2.27	2.07	1.89	1.74	1.60	1.49	1.38	1.28
E_{L_1}/keV	0.503	0.569	0.638	0.703	0.782	0.857	0.940	1.02	1.11	1.20
$\lambda_{L_1}/\text{\AA}$	24.6	21.8	19.43	17.64	15.86	14.47	13.19	12.11	11.21	10.31
E_{L_2}/keV	0.405	0.464	0.527	0.585	0.655	0.724	0.800	0.876	0.94	1.04
$\lambda_{L_2}/\text{\AA}$	30.6	26.7	23.5	21.2	18.92	17.13	15.50	14.15	13.09	11.96
$12 + \log A_Z^{\odot}$	3.05	4.90	4.00	5.64	5.39	7.45	4.92	6.23	4.21	4.60
$12 + \log A_Z^{\text{ISM}}$		4.81		5.51	5.34	7.43	4.92	6.05		
$A_Z^{\text{ISM}}/A_{28}^{\text{ISM}}$		0.0575		0.288	0.195	24.0	0.0741	1.00		

I.2 Bound-bound line-transition wavelengths

The identification of absorption lines requires a large database of transition wavelengths. This section quotes the most important lines, which were also used in this analysis. It can never be a replacement for the numerous complete tables: All quoted wavelengths are either from the atomic database ATOMDB (see also Table A I.7) or from the table of Verner et al. (1996), except of the Na x triplet, which was only found in Mewe et al. (1985).

H-like ions

The strongest lines of H-like ions belong to the Lyman series (Table A I.2). They are also most important in photoionized plasmas, as they start at the ground state. The energies are relatively large, the spin-orbit coupling is therefore usually not resolved – in contrast to the lines of the Balmer series (Table A I.2), which comprises the transitions from the first excited state ($n = 2$) with higher states, where the energy differences are more easily noticeable.

Table A I.2: Wavelengths [in Å] of H-like ions' transitions from the ground state $1s$ ($^2S_{1/2}$)

trans.	name	O	Ne	Na	Mg	Al	Si	S	Ar	Ca	Fe	Ni
$1s \rightarrow np$		VIII	X	XI	XII	XIII	XIV	XVI	XVIII	XX	XXVI	XXVIII
$1s \rightarrow 2p$	Ly α	18.97	12.13	10.03	8.42	7.17	6.18	4.73	3.73	3.02	1.78	1.53
$1s \rightarrow 3p$	Ly β	16.01	10.24	8.46	7.11	6.05	5.22	3.99	3.15	2.55	1.50	1.29
$1s \rightarrow 4p$	Ly γ	15.18	9.71	8.02	6.74	5.74	4.95	3.78	2.99	2.42	1.42	1.23
$1s \rightarrow 5p$	Ly δ	14.82	9.48	7.83	6.58	5.60	4.83	3.70	2.92	2.36	1.39	1.20

Table A I.3: Wavelengths [in Å] of H-like ions' transitions from the first excited state ($n = 2$)

transition	O	Ne	Na	Mg	Al	Si	S	Ar	Ca	Fe	Ni
	VIII	X	XI	XII	XIII	XIV	XVI	XVIII	XX	XXVI	XXVIII
$2s$ ($^2S_{1/2}$) \rightarrow $3p$ ($^2P_{1/2}$)	102.40	65.49	45.44	33.35	25.51	20.13	16.28	9.58	8.25		
$2s$ ($^2S_{1/2}$) \rightarrow $3p$ ($^2P_{3/2}$)	102.36	65.45	45.40	33.31	25.46	20.08	16.23	9.54	8.20		
$2p$ ($^2P_{1/2}$) \rightarrow $3s$ ($^2S_{1/2}$)	102.39	65.49	45.44	33.35	25.50	20.12	16.27	9.58	8.25		
$2p$ ($^2P_{3/2}$) \rightarrow $3s$ ($^2S_{1/2}$)	102.55	65.64	45.59	33.51	25.66	20.28	16.43	9.74	8.40		
$2s$ ($^2S_{1/2}$) \rightarrow $4p$ ($^2P_{1/2}$)	75.86	48.52	33.66	24.71	18.90	14.91	12.06	7.10	6.11		
$2s$ ($^2S_{1/2}$) \rightarrow $4p$ ($^2P_{3/2}$)	75.84	48.50	33.65	24.70	18.89	14.90	12.05	7.09	6.10		
$2p$ ($^2P_{1/2}$) \rightarrow $4s$ ($^2S_{1/2}$)	75.85	48.51	33.66	24.71	18.89	14.91	12.06	7.10	6.11		
$2p$ ($^2P_{1/2}$) \rightarrow $4s$ ($^2S_{1/2}$)	75.94	48.60	33.75	24.79	18.98	15.00	12.15	7.19	6.20		

He-like ions

As described in Sect. 2.1.2, He-like triplets – resonance (r), intercombination (i) and forbidden (f) line – can be very important for the diagnostics of an optically thin plasma. Table A I.4 lists their wavelengths. As the triplets connect, however, only $n = 1$ and $n = 2$ levels, see Table A I.5 for further transitions from the ground state $1s^2$ (1S_0) to $1snp$ (1P_1) states.

Table A I.4: Wavelengths [in Å] of He-like ions' triplet transitions (from the $1s^2$ (1S_0) state)

upper level	O	Ne	Na	Mg	Al	Si	S	Ar	Ca	Fe	Ni
	VII	IX	X	XI	XII	XIII	XV	XVII	XIX	XXV	XXVII
r $1s2p$ (1P_1)	21.60	13.45	11.00	9.17	7.76	6.65	5.04	3.95	3.18	1.85	1.59
i $1s2p$ ($^3P_{1,2}$)	21.80	13.55	11.08	9.23	7.80	6.69	5.07	3.97	3.19	1.86	1.60
f $1s2s$ (3S_1)	22.10	13.70	11.19	9.31	7.87	6.74	5.10	3.99	3.21	1.87	

Table A I.5: Wavelengths [in Å] of He-like ions' transitions from the $1s^2$ (1S_0) ground state

upper level	O	Ne	Na	Mg	Al	Si	S	Ar	Ca	Fe	Ni
	VII	IX	X	XI	XII	XIII	XV	XVII	XIX	XXV	XXVII
$1s2p$ (1P_1)	21.60	13.45	11.00	9.17	7.76	6.65	5.04	3.95	3.18	1.85	1.59
$1s3p$ (1P_1)	18.63	11.54	9.43	7.85	6.63	5.68	4.30	3.37	2.71	1.57	1.35
$1s4p$ (1P_1)	17.77	11.00	8.98	7.47	6.31	5.40	4.09	3.20	2.57	1.50	1.28
$1s5p$ (1P_1)	17.40	10.77	8.79	7.31	6.18	5.29	4.00	3.13	2.51	1.46	1.25
$1s6p$ (1P_1)	17.20	10.64	8.69	7.22	6.10	5.22	3.95	3.10			
$1s7p$ (1P_1)	17.09	10.57	8.63	7.17	6.06	5.19	3.92				
$1s8p$ (1P_1)	17.01	10.51	8.59	7.14	6.03	5.16	3.90				

Li-like ions

The ground state of the alkali metal Lithium is $[1s^2] 2s ({}^2S_{1/2})$. The strongest transitions lead therefore (similar to the Balmer series) to $[1s^2] np ({}^2P)$ states. Table A I.6 lists the wavelength of these transitions and includes in case of $n = 3$ both the ${}^2P_{1/2}$ and the ${}^2P_{3/2}$ state, as the difference due to spin-orbit coupling might be resolvable.

Table A I.6: Wavelengths [in Å] of Li-like ions' transitions from the ground state $[1s^2] 2s ({}^2S_{1/2})$

upper level	O VI	Ne VIII	Na IX	Mg X	Al XI	Si XII	S XIV	Ar XVI	Ca XVIII	Fe XXIV	Ni XXVI
$[1s^2] 3p ({}^2P_{1/2})$	150.09	88.08	70.65	57.88	48.34	40.95	30.47	23.59	18.73	10.66	9.10
$[1s^2] 3p ({}^2P_{3/2})$	150.12	88.12	70.61	57.92	48.30	40.91	30.43	23.55	18.69	10.62	9.06
$[1s^2] 4p$	115.8	67.4	53.9	44.1	36.7	31.0	23.0	17.74	14.09	8.00	6.82
$[1s^2] 5p$	104.8	60.8	48.6	39.7	33.0	27.9	20.7	15.93	12.64	7.17	6.11
$[1s^2] 6p$									11.99	6.79	
$[1s^2] 7p$									11.62		

Level numbers in the ATOMDB

For convenient access the atomic database ATOMDB, e.g., via the `trans(Z,ion,up,low)`-function¹ in ISIS or the web-guide (<http://cxc.harvard.edu/atomdb/WebGUIDE/>), the meaning of the level numbers for H-like, He-like and Li-like ions have been compiled in Table A I.7.

Table A I.7: Quantum states assigned to the first 25 level numbers in the ATOMDB

level #	H-like	He-like	Li-like
25	$5g ({}^2G_{9/2})$	$1s4d ({}^3D_2)$	$[1s^2] 6s ({}^2S_{1/2})$
24	$5g ({}^2G_{7/2})$	$1s4d ({}^3D_1)$	$[1s^2] 5g ({}^2G_{9/2})$
23	$5f ({}^2F_{7/2})$	$1s4p ({}^1P_1)$	$[1s^2] 5g ({}^2G_{7/2})$
22	$5f ({}^2F_{5/2})$	$1s4p ({}^3P_2)$	$[1s^2] 5f ({}^2F_{7/2})$
21	$5d ({}^2D_{3/2})$	$1s4p ({}^3P_1)$	$[1s^2] 5f ({}^2F_{7/2})$
20	$5d ({}^2D_{3/2})$	$1s4p ({}^3P_0)$	$[1s^2] 5d ({}^2D_{5/2})$
19	$5p ({}^2P_{3/2})$	$1s4s ({}^1S_0)$	$[1s^2] 5d ({}^2D_{3/2})$
18	$5p ({}^2P_{1/2})$	$1s4s ({}^3S_1)$	$[1s^2] 5p ({}^2P_{3/2})$
17	$5s ({}^2S_{1/2})$	$1s3d ({}^1D_2)$	$[1s^2] 5p ({}^2P_{1/2})$
16	$4f ({}^2F_{7/2})$	$1s3d ({}^3D_3)$	$[1s^2] 5s ({}^2S_{1/2})$
15	$4f ({}^2F_{5/2})$	$1s3d ({}^3D_2)$	$[1s^2] 4f ({}^2F_{7/2})$
14	$4d ({}^2D_{5/2})$	$1s3d ({}^3D_1)$	$[1s^2] 4f ({}^2F_{5/2})$
13	$4d ({}^2D_{3/2})$	$1s3p ({}^1P_1)$	$[1s^2] 4d ({}^2D_{5/2})$
12	$4p ({}^2P_{3/2})$	$1s3p ({}^3P_2)$	$[1s^2] 4d ({}^2D_{3/2})$
11	$4p ({}^2P_{1/2})$	$1s3p ({}^3P_1)$	$[1s^2] 4p ({}^2P_{3/2})$
10	$4s ({}^2S_{1/2})$	$1s3p ({}^3P_0)$	$[1s^2] 4p ({}^2P_{1/2})$
9	$3d ({}^2D_{5/2})$	$1s3s ({}^1S_0)$	$[1s^2] 4s ({}^2S_{1/2})$
8	$3d ({}^2D_{3/2})$	$1s3s ({}^3S_1)$	$[1s^2] 3d ({}^2D_{5/2})$
7	$3p ({}^2P_{3/2})$	$1s2p ({}^1P_1)$	$[1s^2] 3d ({}^2D_{3/2})$
6	$3p ({}^2P_{1/2})$	$1s2p ({}^3P_2)$	$[1s^2] 3p ({}^2P_{3/2})$
5	$3s ({}^2S_{1/2})$	$1s2p ({}^3P_1)$	$[1s^2] 3p ({}^2P_{1/2})$
4	$2p ({}^2P_{3/2})$	$1s2p ({}^3P_0)$	$[1s^2] 3s ({}^2S_{1/2})$
3	$2p ({}^2P_{1/2})$	$1s2s ({}^1S_0)$	$[1s^2] 2p ({}^2P_{3/2})$
2	$2s ({}^2S_{1/2})$	$1s2s ({}^3S_1)$	$[1s^2] 2p ({}^2P_{1/2})$
1	$1s ({}^2S_{1/2})$	$1s^2 ({}^1S_0)$	$[1s^2] 2s ({}^2S_{1/2})$

¹ The function `trans` returns a boolean array telling for every line-id whether the transition matches or not. The list of matching line-ids can be obtained by `where(trans(Z,ion,up,low))`.

Further iron ions

Continuing with further less-ionized ions of all atoms is not useful due to the limited energy-range and resolution of a *Chandra*/HETGS observation. Therefore, only further iron ions will be discussed in the rest of this section. The excitation of Be-like and B-like ions can, to some extent, still be treated in a systematic way similar to the series as above for very highly ionized ions. This is, however, hardly possible for the overwhelming number of L-shell transitions of lower ionized iron in the range between $\approx 7 \text{ \AA}$ and $\approx 17 \text{ \AA}$. Therefore, the following tables only present the strongest transitions, which is even not well defined, as several weaker transitions blend in many cases and may thus effectively again produce stronger features.

Table A I.8: Further iron lines

The quoted wavelengths rely on the ATOMDB, as the table of Verner et al. (1996) is not complete and its combination of several transitions into multiplets is not so clear.

(a) Fe xxiii (Be-like ion)			(b) Fe xxii (B-like ion)		
transition from	#	λ [\AA]	transition from	#	λ [\AA]
$[1s^2] 2s^2 ({}^1S_0)$	1		$[1s^2 2s^2] 2p ({}^2P_{1/2})$	1	
$\rightarrow [1s^2] 2s3p ({}^1P_1)$	15	10.98	$\rightarrow [1s^2 2s^2] 3s ({}^2S_{1/2})$	16	12.25
$\rightarrow [1s^2] 2s4p ({}^1P_1)$	52	8.30	$\rightarrow [1s^2 2s^2] 4s ({}^2S_{1/2})$	69	9.06
$\rightarrow [1s^2] 2s5p ({}^1P_1)$	104	7.47	$\rightarrow [1s^2 2s^2] 5s ({}^2S_{1/2})$	148	8.11
$\rightarrow [1s^2] 2s3p ({}^3P_1)$	13	11.02	$\rightarrow [1s^2 2s^2] 3d ({}^2D_{3/2})$	21	11.77
$\rightarrow [1s^2] 2s4p ({}^3P_1)$	50	8.32	$\rightarrow [1s^2 2s^2] 4d ({}^2D_{3/2})$	72	8.97
			$\rightarrow [1s^2 2s^2] 5d ({}^2D_{3/2})$	151	8.09
			$\rightarrow [1s^2] 2s2p 3p_{3/2}$	30	11.49
			$\rightarrow [1s^2] 2s2p 3p_{3/2}$	32	11.43

(c) Fe xxi (C-like ion)			
transition from	#	λ	A
$[1s^2 2s^2] 2p^2 ({}^3P_0)$	1	[\AA]	[$10^{12}/\text{s}$]
$\rightarrow [1s^2 2s^2] 2p 3d ({}^3D_0)$	40	12.28	18.2
$\rightarrow [1s^2] 2s 2p_{1/2}^2 3p_{3/2}$	58	11.97	3.09
$\rightarrow [1s^2] 2s 2p_{1/2} 2p_{3/2} 3p_{3/2}$	60	11.95	1.82
$\rightarrow [1s^2 2s^2] 2p 4d ({}^3P_1)$	248	9.48	6.12
$\rightarrow [1s^2] 2s 2p_{1/2}^2 4p_{3/2}$	283	9.19	2.88
$\rightarrow [1s^2 2s^2] 2p 5d_{3/2}$	460	8.57	2.85

(d) Fe xx (N-like ion)			
transition from	#	λ	A
$[1s^2 2s^2] 2p^3 ({}^4S_{3/2})$	1	[\AA]	[$10^{12}/\text{s}$]
$\rightarrow [1s^2 2s^2] 2p^2 3s ({}^4P_{1/2})$	16	13.96	1.19
$\rightarrow [1s^2 2s^2] 2p^2 3s ({}^4P_{3/2})$	17	13.84	1.00
$\rightarrow [1s^2 2s^2] 2p^2 3s ({}^4P_{5/2})$	19	13.77	1.02
	42	13.06	2.62
	45	12.99	2.01
	47,48	12.97...12.96	0.66 + 3.46
$\rightarrow [1s^2 2s^2] 2p^2 3d (\dots)$	50,51	12.92...12.91	0.74 + 4.91
	56,58	12.86...12.85	12.1 + 19.2
	59,60	12.83...12.82	4.90 + 17.1
	62,63	12.76...12.75	0.25 + 1.44
$\rightarrow [1s^2] 2s 2p^3 3p (\dots)$	72,73	12.58	1.44 + 4.39
	75	12.53	4.23
	285,286	10.13...10.12	0.39 + 2.12
$\rightarrow [1s^2 2s^2] 2p^2 4d (\dots)$	297,299-302	10.04...10.06	1.16 + 0.48 + 2.80 + 0.64 + 0.63
	305,306,309,313	9.99...10.01	3.01 + 5.80 + 6.56 + 0.81
$\rightarrow [1s^2] 2s 2p^3 4p (\dots)$	363,364,365	9.73...9.72	2.42 + 2.42 + 2.47
	518,526	9.20...9.19	1.04 + 1.43
$\rightarrow [1s^2 2s^2] 2p^2 5d (\dots)$	555,556,559,564	9.11...9.10	0.67 + 0.29 + 1.46 + 0.36
	590,592,594	9.07...9.06	1.16 + 2.51 + 3.21
$\rightarrow \dots$	700-702	8.82	1.07 + 1.27 + 1.37

Table A I.8b includes only the strongest transitions of Fe XXII with $A > 5 \times 10^{12}/s$. Table A I.8e lists the strongest ($A > 5 \times 10^{12}/s$) lines of Fe XIX and Table A I.8f lists all lines of Fe XVIII with $A > 9 \times 10^{11}/s$.

(e) Fe XIX (O-like ion)

transition from $[1s^2 2s^2] 2p^4 (^3P_2)$	# 1	λ [Å]	A [$10^{12}/s$]
$\rightarrow [1s^2 2s^2] 2p_{1/2} 2p_{3/2}^2 3d_{5/2}$	53	13.79	5.35
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 3d (^3F_3)$	57	13.64	2.43
$\rightarrow [1s^2 2s^2] 2p^3 3d_{5/2}$	65,67	13.55	4.44 + 2.25
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 3d (^3D_3)$	68	13.52	18.7
$\rightarrow [1s^2 2s^2] 2p_{1/2} 2p_{3/2}^2 3d_{3/2}$	71	13.50	12.9
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 3d (^3S_1 =$	74	13.46	14.1
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 3d (^1F_3)$	76	13.42	5.01
$\rightarrow [1s^2] 2s 2p_{1/2} 2p_{3/2}^3 3p_{3/2}$	104	12.95	3.11
$\rightarrow [1s^2] 2s 2p_{1/2}^2 2p_{3/2}^2 3p_{3/2}$	106	12.93	3.37
$\rightarrow [1s^2 2s^2] 2p^3 (^4S) 4d (^3D_3)$	243	10.82	5.65
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 4d (^3F_3)$	276	10.68	2.28
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 4d (^3D_3)$	286	10.65	3.74
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 4d (^3P_2)$	288	10.64	5.20
$\rightarrow [1s^2 2s^2] 2p^3 (^2D) 4d (^3S_1)$	292	10.63	4.78
$\rightarrow [1s^2 2s^2] 2p^3 (^4S) 5d (^3D_3)$	432	9.86	3.59
$\rightarrow [1s^2 2s^2] 2p_{1/2} 2p_{3/2}^2 5d_{3/2}$	532,536	9.69	2.56 + 2.18

(f) Fe XVIII (F-like ion)

transition from $[1s^2 2s^2] 2p^5 (^2P_{3/2})$	# 1	λ [Å]	A [$10^{12}/s$]
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 3s (^2P_{3/2})$	5	16.00	1.36
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 3s (^2P_{1/2})$	8	15.76	1.06
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 3d (^4P_{1/2})$	39	14.60	2.50
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 3d (^4P_{3/2})$	40	14.57	3.09
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 3d (^2F_{5/2})$	41	14.53	4.05
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 3d (^2D_{5/2})$	49	14.37	6.75
$\rightarrow [1s^2 2s^2] 2p^4 (^1D) 3d (\dots)$	52,53	14.26	12.9 + 1.29
$\rightarrow [1s^2 2s^2] 2p^4 3d (\dots)$	55,56	14.21	17.9 + 19.4
$\rightarrow [1s^2 2s^2] 2p^4 (^1D) 3d (^2D_{3/2})$	57	14.16	4.03
$\rightarrow [1s^2 2s^2] 2p^4 (^1D) 3d (^2P_{1/2})$	58	14.14	4.57
$\rightarrow [1s^2 2s^2] 2p^4 (^1S) 3d (^2D_{5/2})$	59	13.95	1.04
$\rightarrow [1s^2] 2s 2p_{1/2}^2 2p_{3/2}^3 3p_{3/2}$	69	13.41	1.09
$\rightarrow [1s^2 2s 2p^5 (^3P) 3p (^2D_{5/2})$	70	13.39	1.64
$\rightarrow [1s^2 2s 2p^5 (^3P) 3p (^2P_{3/2})$	72	13.36	2.31
$\rightarrow [1s^2] 2s 2p_{1/2} 2p_{3/2}^4 3p (\dots)$	73,74	13.32	3.59 + 1.17
$\rightarrow [1s^2 2s 2p^5 (^1P) 3p (^2S_{1/2})$	80	13.18	1.18
$\rightarrow [1s^2 2s^2] 2p_{1/2}^2 2p_{3/2}^2 4d_{5/2}$	136	11.57	1.53
$\rightarrow [1s^2 2s^2] 2p^4 4d (\dots)$	137,138	11.53	3.55 + 4.22
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 4d (^2F_{5/2})$	164	11.42	4.75
$\rightarrow [1s^2 2s^2] 2p^4 (^1D) 4d (\dots)$	176,178,180	11.33	4.82 + 4.48 + 3.26
$\rightarrow [1s^2 2s^2] 2p_{1/2} 2p_{3/2}^3 4d_{5/2}$	182	11.31	1.09
$\rightarrow [1s^2 2s^2] 2p^4 (^1D) 4d (^2D_{3/2})$	181	11.29	1.28
$\rightarrow [1s^2] 2s 2p_{1/2}^2 2p_{3/2}^3 4p_{3/2}$	220	10.57	1.39
$\rightarrow [1s^2 2s 1s^2 2p_{1/2}^2 2p_{3/2}^3 4p_{3/2}$	221	10.56	1.58
$\rightarrow [1s^2 2s^2] 2p^4 5d (\dots)$	228,231	10.54	1.22 + 2.25 + 2.60
$\rightarrow [1s^2 2s^2] 2p^4 (^3P) 5d (^2D_{5/2})$	276	10.45	2.09
$\rightarrow [1s^2 2s^2] 2p_{1/2} 2p_{3/2}^3 5d (\dots)$	323,326,328	10.36	2.36 + 1.93 + 1.25

(g) Fe xvii (Ne-like ion)

transition from [$1s^2 2s^2 2p^6$] ($1S_0$)	# 1	λ [Å]	A [$10^{12}/s$]
→ [$1s^2 2s^2$] $2p^5(^2P) 3s(^3P_1)$	3	17.05	1.00
→ [$1s^2 2s^2$] $2p^5(^2P) 3s(^1P_1)$	5	16.78	0.90
→ [$1s^2 2s^2$] $2p^5(^2P) 3d(^3D_1)$	23	15.26	5.87
→ [$1s^2 2s^2$] $2p^5(^2P) 3d(^1P_1)$	27	15.01	27.0
→ [$1s^2$] $2s 2p^6 3p(^1P_1)$	33	13.82	3.40
→ [$1s^2 2s^2$] $2p^5(^2P) 4d(^3D_1)$	59	12.27	4.21
→ [$1s^2 2s^2$] $2p^5(^2P) 4d(^1P_1)$	71	12.12	4.83
→ [$1s^2 2s^2$] $2p^5(^2P_2) 5d(^3D_1)$	93	11.25	2.87
→ [$1s^2 2s^2$] $2p^5(^2P) 5d(^1P_1)$	118	11.13	2.26
→ [$1s^2$] $2s 2p^6 4p(^1P_1)$	131	11.03	1.75
→ [$1s^2 2s^2$] $2p^5(^2P) 6d(^3D_1)$	155	10.77	1.90
→ [$1s^2 2s^2$] $2p^5(^2P) 6d(^1P_1)$	181	10.66	1.15
→ [$1s^2 2s^2$] $2p^5(^2P) 7d(^3D_1)$	205	10.50	1.39
→ [$1s^2$] $2s 2p^6 5p(^1P_1)$	245	10.12	0.99

Table A I.8: Further iron lines (end)

II Data files from the CXC

It was explained in Section 3.1 how data from the *Chandra* X-ray observatory is organized and is reduced with CIAO. Primary and secondary data files can be downloaded from <http://cda.harvard.edu:9011/chaser/>. This appendix contains a file structure listing of the most important involved files and is thought to serve as a reference during the data reduction.

All data files from the Chandra X-ray Center (CXC) are organized in the FITS (flexible image transport system) format. They are fully consistent with the OGIP standards. OGIP is “*is a division of the Laboratory for High Energy Astrophysics at Goddard Space Flight Center. They oversee the activities of the HEASARC FITS Working Group (HFWG), which makes sure that FITS formats and keywords conform to the current standards and conventions.*” (CIAO online dictionary, <http://cxc.harvard.edu/ciao/dictionary/ogip.html>)

II.1 Primary and secondary data files

While primary data products are considered (by CXC) “to be sufficient for most analyses”, all secondary (level 1) data products are needed for data reprocessing and thus essential for this work. (Primary products from standard data processing may not always be reliable.)

Table A II.1: Contents of a level 1-event file

block	#	column	unit	description
BLOCK EVENTS	1	time	s	S/C TT corresponding to mid-exposure
	2	ccd_id		CCD reporting event
	3	node_id		CCD serial readout amplifier node
	4	expno		Exposure number of CCD frame containing event
	5	chip(chipx,chipy)	pixel	Chip coords
	6	tdet(tdetx,tdety)	pixel	ACIS tiled detector coordinates
	7	det(detx,dety)	pixel	ACIS detector coordinates
	8	sky(x,y)	pixel	sky coordinates
	9	pha	adu	total pulse height of event
	10	pha_r	adu	total read-out pulse height of event
	11	corn_pha		mean of event corner pixel PHA
	12	energy	eV	nominal energy of event (eV)
	13	pi	chan	pulse invariant energy of event
	14	fltgrade		event grade, flight system
	15	grade		binned event grade
	16	status[4]		event status bits
BLOCK GTI _n	1	start	s	S/C TT corresponding to mid-exposure
	2	stop	s	S/C TT corresponding to mid-exposure

Table A II.2: Contents of a aspect/PCAD file

block	#	column	unit	description
BLOCK ASPSOL	1	time	s	Time
	2	ra	deg	RA of MNC frame (x-axis)
	3	dec	deg	DEC of MNC frame (x-axis)
	4	roll	deg	ROLL of MNC frame
	5	ra_err	deg	Uncertainty in RA
	6	dec_err	deg	Uncertainty in DEC
	7	roll_err	deg	Uncertainty in ROLL
	8	dy	mm	dY of STF frame - FC frame
	9	dz	mm	dZ of STF frame - FC frame
	10	dtheta	deg	dTHETA of STF frame - FC frame
	11	dy_err	mm	Uncertainty in dY
	12	dz_err	mm	Uncertainty in dZ
	13	dtheta_err	deg	Uncertainty in dTHETA
	14	q_att[4]		S/C attitude quaternion
	15	roll_bias	deg/s	Roll bias rate
	16	pitch_bias	deg/s	Pitch bias rate
	17	yaw_bias	deg/s	Yaw bias rate
	18	roll_bias_err	deg/s	Roll bias rate error
	19	pitch_bias_err	deg/s	Pitch bias rate error
	20	yaw_bias_err	deg/s	Yaw bias rate error

Table A II.3: Contents of a parameter block file

block	#	column	unit	description
BLOCK PBK	1	ccd_id		CCD ID
	2	fep_id		Front End Processor ID
	3	vidresp		CCD video chain response selection, 0 for 1:1
	4	evt_thr[4]	adu	Event thresholds for nodes A-D (TLMIN=-4096)
	5	spl_thr[4]	adu	Split thresholds for output nodes A-D
	6	bcmpslot		Slot identifier for bias map compression tab
	7	biasalg		Bias algorithm is. 1:whole frame; 2:strip
	8	biasarg0		Bias argument 0 (TLMIN=-32768)
	9	biasarg1		Bias argument 1 (TLMIN=-32768)
	10	biasarg2		Bias argument 2 (TLMIN=-32768)
	11	biasarg3		Bias argument 3 (TLMIN=-32768)
	12	biasarg4		Bias argument 4 (TLMIN=-32768)
	13	vid_off[4]		Video offsets for CCD output nodes A-D

Table A II.4: Contents of a bias file

block	#	column	unit	description
BLOCK BIAS	1	bias[1024,1024]		

Table A II.5: Contents of a filter file

block	#	column	unit	description
BLOCK GTI _n	1	start	s	S/C TT corresponding to mid-exposure
	2	stop	s	S/C TT corresponding to mid-exposure

Table A II.6: Contents of a mask file

block	#	column	unit	description
BLOCK MASK _n	1	shape		region shape
	2	component		Component index
	3	chip(chipx,chipy)[2]	pixel	CHIP position
	4	samp_cyc		sampling cycle
	5	phamin	adu	minimum pulse height
	6	phamax	adu	maximum pulse height

Table A II.7: Contents of a bad pixel file

block	#	column	unit	description
BLOCK BADPIX _n	1	shape		region shape
	2	component		Component number
	3	chip(chipx,chipy)[2]	pixel	CHIP location
	4	time	s	Time pixel went bad
	5	time_stop	s	Time pixel went bad
	6	status[4]		Badpixel status code

II.2 High level data files

High level data products are the ones obtained by the data reduction. They are finally used for scientific analysis. It is possible that they have to be reprocessed to apply a new calibration.

Table A II.8: Contents of a level 1.5-event file

block	#	column	unit	description
BLOCK EVENTS	1	time	s	time tag of data record
	2	expno		
	3	rd(tg_r,tg_d)	deg	Grating angular coords
	4	chip(chipx,chipy)	pixel	Chip coords
	5	tdet(tdetx,tdety)	pixel	Tdet coords
	6	det(detx,dety)	pixel	Det coords
	7	sky(x,y)	pixel	Sky coords
	8	ccd_id		
	9	pha		
	10	pi		
	11	energy		
	12	grade		
	13	fltgrade		
	14	node_id		
	15	tg_m		Diffraction order (m)
	16	tg_lam	angstrom	wavelength (λ)
	17	tg_mlamp	angstrom	Order times wavelength ($m * \lambda$)
	18	tg_srcid		source ID, index from detect table
	19	tg_part		component index (HEG, MEG, LEG, HESF regions)
	20	tg_smap		source map; flags for up to 10 sources
	21	status[4]		event status bits
BLOCK REGION	1	source		Source Number
	2	shape		Shape of the region
	3	sky(x,y)	pixel	Sky coords
	4	r[2]	pixel	Radius Vector for SHAPE
	5	rotang	deg	Rotation angle for SHAPE
	6	grating		Applicable grating. hetg or letg
	7	tg_part		TG_PART
	8	component		Component number

Table A II.9: Contents of a light curve file

block	#	column	unit	description
BLOCK LIGHTCURVE	1	time_bin	channel	time tag of data record
	2	time_min	s	Minimum Value in Bin
	3	time	s	time tag of data record
	4	time_max	s	Maximum Value in Bin
	5	counts	count	Counts
	6	stat_err	count	Statistical error
	7	count_rate	count/s	Rate
	8	count_rate_err	count/s	Rate Error
	9	exposure	s	Time per interval
BLOCK GTI _n	1	start	s	time tag of data record
	2	stop	s	time tag of data record

Table A II.10: Contents of a spectra (pha2) file

block	#	column	unit	description
BLOCK SPECTRUM	1	spec_num		Spectrum Number
	2	tg_m		Diffraction order (m)
	3	tg_part		Spectral component (HEG, MEG, LEG, HESF parts)
	4	tg_srcid		Source ID, output by detect
	5	x	pixel	X sky coord of source
	6	y	pixel	Y sky coord of source
	7	channel[8192]		Vector of spectral bin numbers.
	8	counts[8192]	count	Counts array (a spectrum)
	9	stat_err[8192]	count	Statistical uncertainty (error) on counts column
	10	background_up[8192]	count	Background count vector
	11	background_down[8192]	count	Background count vector
	12	bin_lo[8192]	angstrom	Bin boundary, left edge
	13	bin_hi[8192]	angstrom	Bin boundary, right edge
BLOCK REGION	1	spec_num		Spectrum number, which points to the row in the
	2	rowid		Source or a background region?
	3	shape		Shape of region
	4	wavpos(tg_lam, tg_d)		Wavelength(angstrom), Cross Dispersion(degrees)
	5	r[2]	(angstrom, degrees)	Radius vector for SHAPE
	6	rotang	degrees	Rotation angle for SHAPE
	7	tg_part		Grating part index (HEG=1, MEG=2, LEG=3)
	8	tg_srcid		Source identification number
	9	tg_m		Diffraction order
	10	component		Component number

Table A II.11: Contents of a background (bkg2) file

block	#	column	unit	description
BLOCK SPECTRUM	1	spec_num		Spectrum Number
	2	tg_m		Diffraction order (m)
	3	tg_part		Spectral component (HEG, MEG, LEG, HESF parts)
	4	tg_srcid		Source ID, output by detect
	5	x	pixel	X sky coord of source
	6	y	pixel	Y sky coord of source
	7	channel[8192]		Vector of spectral bin numbers.
	8	bin_lo[8192]	angstrom	Bin boundary, left edge
	9	bin_hi[8192]	angstrom	Bin boundary, right edge
	10	counts[8192]		User defined column

Table A II.12: Contents of a grating redistribution matrix function (gRMF) file

	block	#	column	unit	description
BLOCK MATRIX		1	energ_lo	keV	
		2	energ_hi	keV	
		3	n_grp		
		4	f_chan		
		5	n_chan		
		6	matrix[103]		
BLOCK EBOUNDS		1	channel	channel	
		2	e_min	keV	
		3	e_max	keV	

Table A II.13: Contents of a grating ancillary response function (gARF) file

	block	#	column	unit	description
BLOCK SPECRESP		1	energ_lo	keV	Energy
		2	energ_hi	keV	Energy
		3	specresp	cm**2	Effective Area
		4	bin_lo	angstrom	
		5	bin_hi	angstrom	
		6	fracexpo		

III The *Chandra* observation # 3814

The analysis of the *Chandra* observation # 3814 was the main achievement of this thesis. It was discussed in great detail in chapter 4. This appendix contains completing plots and detailed tables, which were removed from the main part to keep the overview on the main procedures.

III.1 Flux-ratios of the sub-spectra

The consistency of the ‘non-dip’ sub-spectra with the whole ‘non-dip’ spectrum was investigated in Section 4.2.1 (page 57). Figs. A III.1 and A III.2 show the flux-ratios of the ‘non-dip 1’ and the ‘non-dip 3’ sub-spectra to the average ‘non-dip’ spectrum. (The same is not shown explicitly for the ‘non-dip 2’ spectrum again, which is complementary to the ‘non-dip 1’ and ‘non-dip 3’ spectra regarding the ratio to the total ‘non-dip’ spectrum. As there are almost no deviations, it is clear that the ‘non-dip 2’ spectrum would show the same behavior.)

Colors were used to visualize the consistency of the fluxratio with 1: The gray lines show bins, where the value 1 is in the middle third of the error bar ($|r - 1| \leq \Delta r/3$), red indicates positive deviations, blue negative deviations. Darker colors are used when the ratio is not consistent with 1 at all ($|r - 1| > \Delta r$). Each plot shows in its first 4 panels the individual ratios of the MEG \pm 1 and HEG \pm 1 spectra, while the last panel shows the average.

The same analysis was repeated for the ‘dip’ sub-spectra with respect to the whole ‘dip’ spectrum, as described in Section 4.3.1 (page 88): Figs. A III.3 – A III.5 show the analogous flux-ratios of the ‘dip 1’, ‘dip 2’ and ‘dip 3’ sub-spectra to the ‘dip’ spectrum. The trend that ‘dip 1’ is less strongly absorbed than the average ‘dip’, while ‘dip 2’ is more absorbed, can also be directly seen from the flux-corrected spectra themselves (Fig. 4.16).

III.2 Spectral analysis of the ‘non-dip’ spectrum

The detailed results for the description of lines in the ‘non-dip’ spectrum (Sect. 4.2.3) are shown on the following pages:

Table 4.10 (Sect. 4.2.3, page 73) gave the fit-parameters of all trustably identified lines. Table A III.1, which starts on page 129, however, lists the parameters of all the lines included to describe the spectrum. (It was stated that many of the (unidentified) lines might not be real, but just describe calibration uncertainties or statistical fluctuation.) The parameters are – as in Table 4.10: their position, full width at half maximum (with colored background if the confidence interval or even the width itself did not converge properly), equivalent width, improvement on the χ^2 -statistics (cf. comments to Table 4.4, page 61) and possible identifications (gray backgrounds mark emission lines): ion, electronic states of the transition (where ground states are underlined), theoretical wavelength (from CXCC’s atomic database ATOMDB 1.3.1) and Einstein coefficient as a measure for the expected strength of the lines. Identifications in brackets just mean that the theoretical wavelengths are not within the range of the fitted positions (which is, however due to Doppler shifts, no reason that the identification is not correct). Trustable identifications (cf. Sect. 4.2.3) are marked by \leftarrow arrows.

The plots following this long list of lines (Fig. A III.6a-A III.6h, on pages 137-144) show the spectrum with all fitted lines in each 2 Å-wavelength-intervals. The upper panel shows the residuals $\Delta\chi$ of the continuum, i.e., before including the fitted lines in the model. Then, the combined count rates of each the MEG \pm 1 and HEG \pm 1 spectra are shown. This is only for clearer visualization; all spectra have been fitted independently (though simultaneously). Trusted line identifications are labeled.

III.3 Dependencies of the continuum-parameters

The next section starts at page 145.

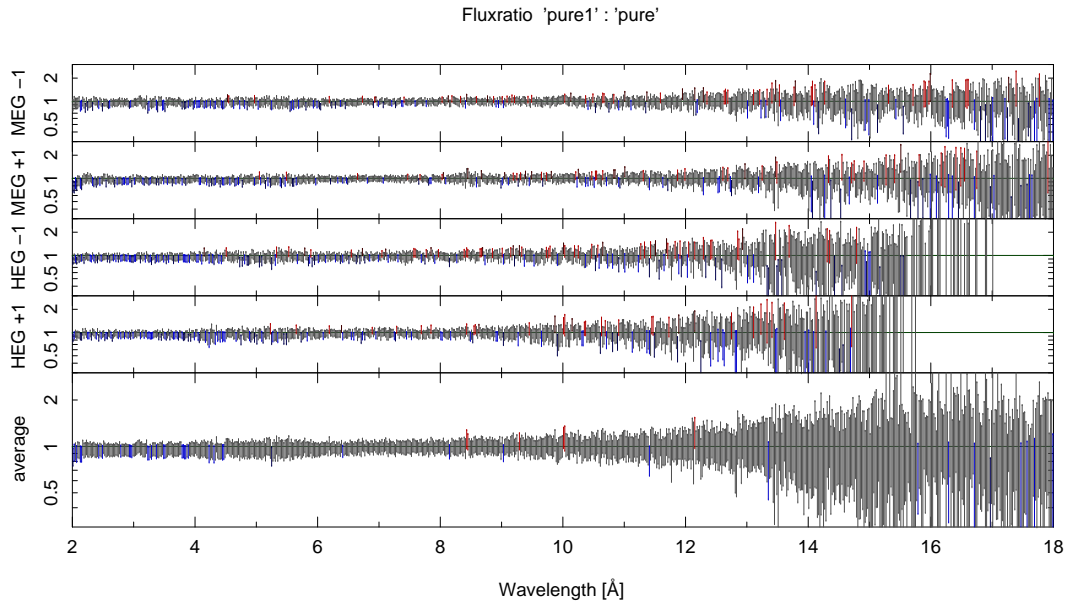
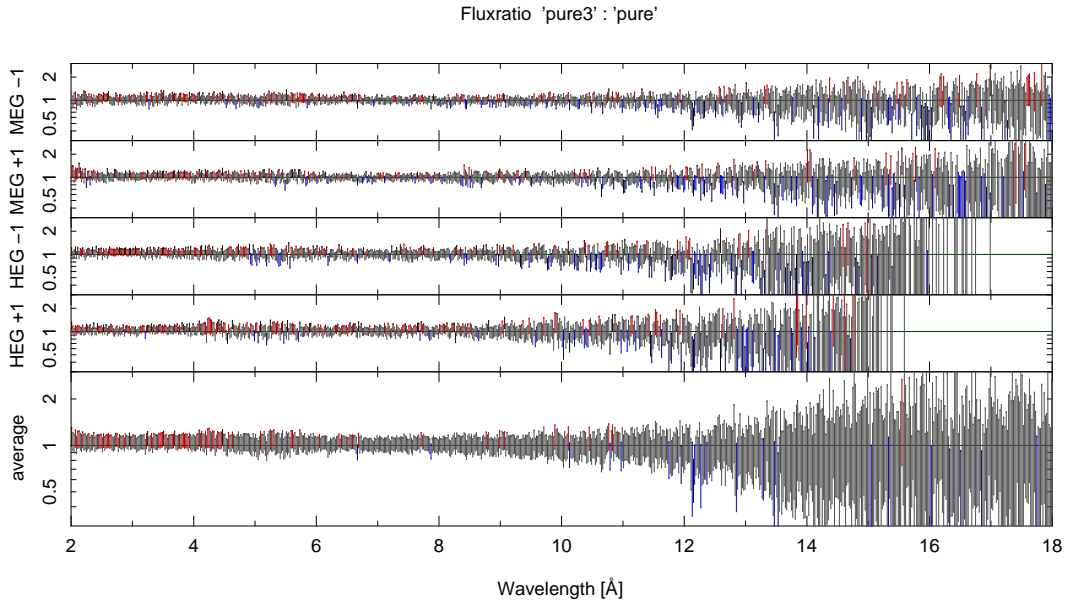
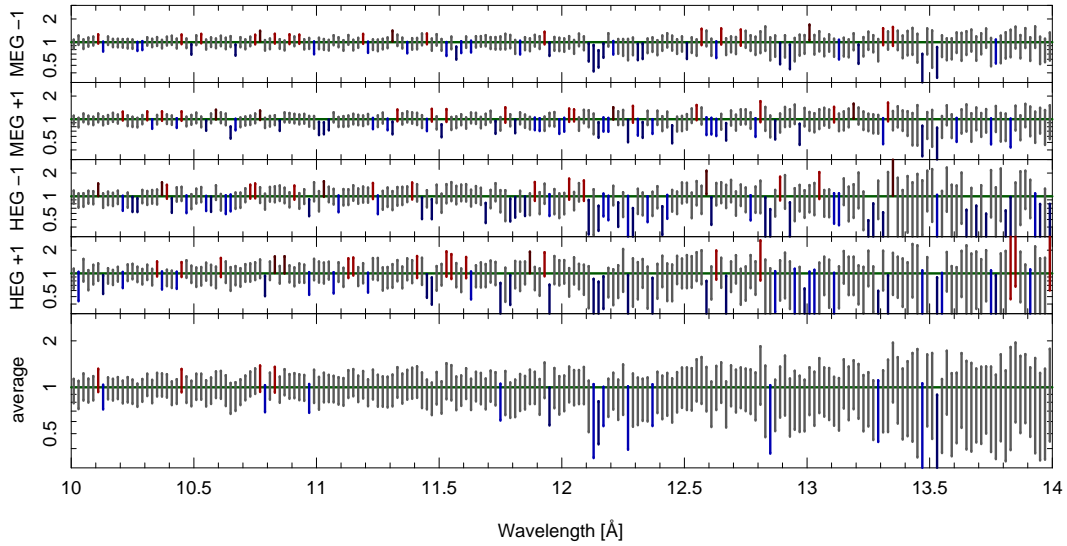


Figure A III.1: Ratio of 'non-dip 1' and 'non-dip' flux-spectrum.



(a) total spectrum



(b) zoom of the 10...14 Å range

Figure A III.2: Ratio of 'non-dip 3' and 'non-dip' flux-spectrum.

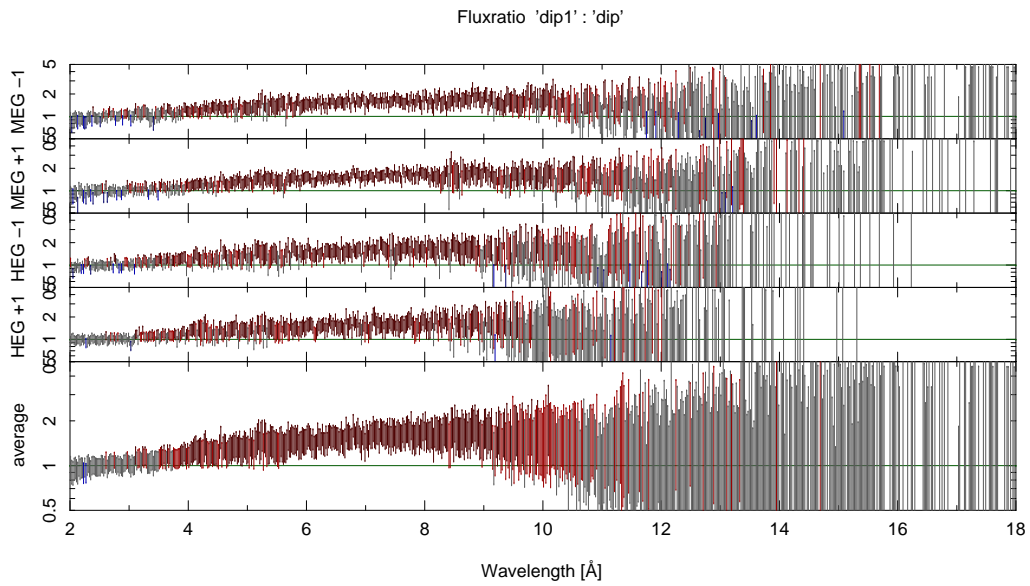


Figure A III.3: Ratio of 'dip 1' and 'dip' flux-spectrum.

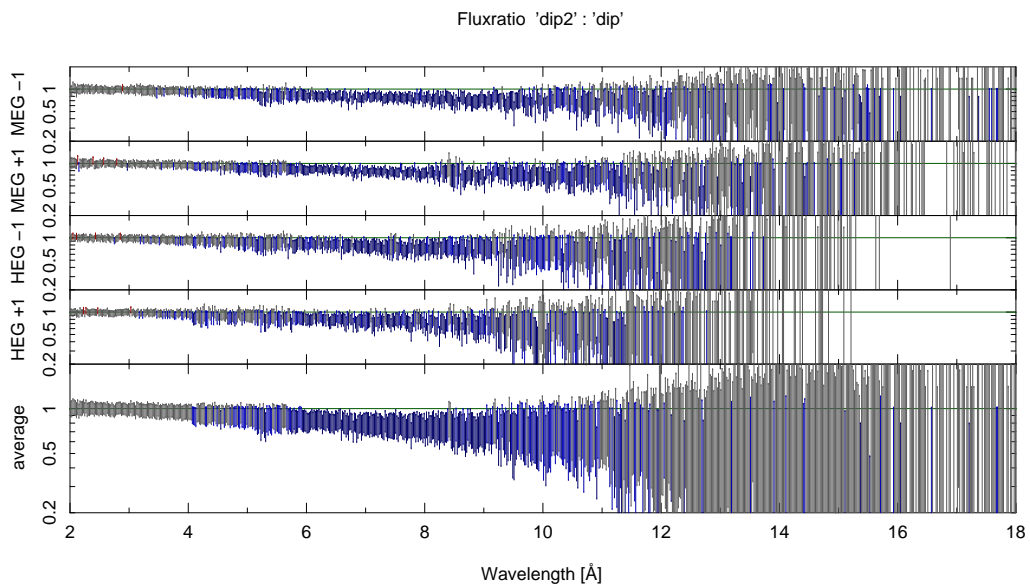


Figure A III.4: Ratio of 'dip 2' and 'dip' flux-spectrum.

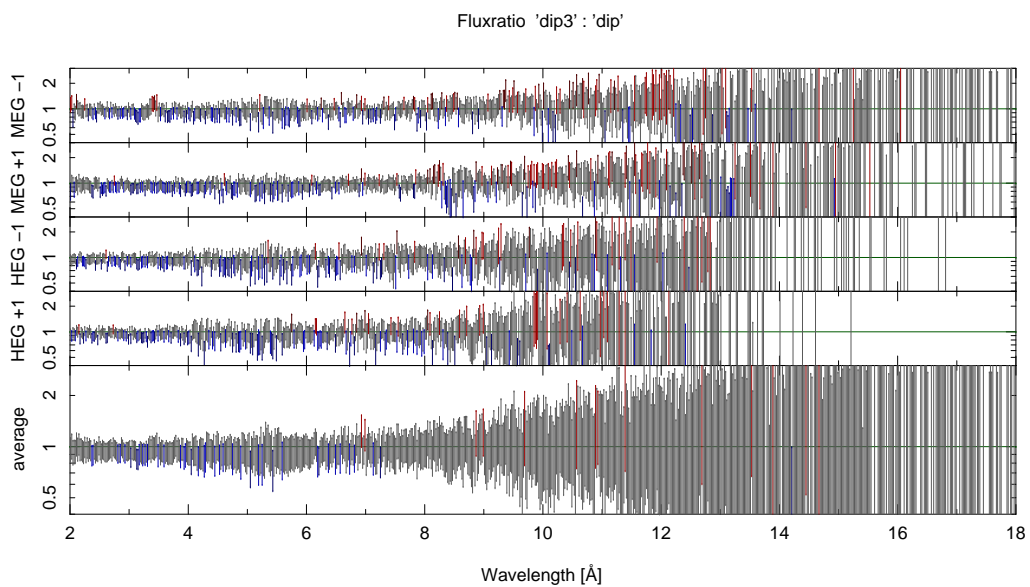


Figure A III.5: Ratio of 'dip 3' and 'dip' flux-spectrum.

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength

λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å][10 ¹² s ⁻¹]	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]	
1.2146 ^{+0.0070} _{-0.0070}	27.92 ^{+2.50} _{-15.81}	-19.84 ^{+8.25} _{-7.53}	0.0	(Ni xxviii 1s 4p)	1.2268	41.2	←	-2984 ⁺¹⁷²² ₋₁₇₁₁
				(Ni xxviii 1s 4p)	1.2272	41.2		-3089 ⁺¹⁷²¹ ₋₁₇₁₀
1.4475 ^{+0.0025} _{-0.0031}	0.11 ^{+19.89} _{-0.11}	-4.50 ^{+1.66} _{-3.90}	19.3	(Fe xxv 1s ² 1s5p)	1.4610	28.3		-2777 ⁺⁵⁰³ ₋₆₄₀
1.4670 ^{+0.0030} _{-0.0028}	1.78 ^{+11.14} _{-1.78}	-4.43 ^{+1.63} _{-2.07}	19.9	(Fe xxv 1s ² 1s5p)	1.4610	28.3		1238 ⁺⁶⁰⁷ ₋₅₆₆
1.4914 ^{+0.0066} _{-0.0041}	0.72 ^{+23.63} _{-0.72}	-2.51 ^{+1.61} _{-2.28}	6.5	Fe xxv 1s ² 1s4p	1.4950	56.3	←	-713 ⁺¹³¹⁷ ₋₈₁₇
1.5180 ^{+0.0028} _{-0.0027}	0.18 ^{+10.71} _{-0.18}	-3.54 ^{+1.49} _{-1.49}	14.7	(Ni xxviii 1s 2p)	1.5304	379	←	-2422 ⁺⁵⁵⁵ ₋₅₃₅
				(Ni xxviii 1s 2p)	1.5356	378		-3428 ⁺⁵⁵³ ₋₅₃₃
1.5526 ^{+0.0033} _{-0.0025}	0.33 ^{+11.99} _{-0.33}	-3.02 ^{+1.48} _{-1.58}	10.9	(Fe xxv 1s ² 1s3p)	1.5731	137		-3912 ⁺⁶³⁵ ₋₄₈₆
1.8499 ^{+0.0029} _{-0.0038}	5.42 ^{+14.13} _{-5.42}	-2.11 ^{+0.90} _{-1.25}	18.6	Fe xxv 1s ² 1s2p	1.8504	503	←	-87 ⁺⁴⁷⁸ ₋₆₂₂
1.9395 ^{+0.0021} _{-0.0020}	11.51 ^{+6.13} _{-5.28}	5.00 ^{+1.24} _{-1.18}	58.8	(Fe Kα)	1.9370		←	385 ⁺³³¹ ₋₃₁₆
1.9678 ^{+0.0086} _{-0.0043}	0.04 ^{+25.29} _{-0.04}	0.93 ^{+0.86} _{-0.82}	3.8					
1.9922 ^{+0.0051} _{-0.0046}	13.69 ^{+15.92} _{-13.69}	2.28 ^{+1.13} _{-1.22}	13.9					
2.0328 ^{+0.0022} _{-0.0028}	0.00 ^{+16.47} _{-0.00}	1.77 ^{+0.87} _{-0.83}	13.1					
2.2900 ^{+0.0026} _{-0.0001}	0.01 ^{+5.73} _{-0.01}	1.81 ^{+0.80} _{-0.76}	16.0					
2.3204 ^{+0.0200} _{-0.0090}	10.07 ^{+65.36} _{-10.07}	1.10 ^{+2.13} _{-0.82}	5.4					
2.3476 ^{+0.0025} _{-0.0026}	0.02 ^{+74.98} _{-0.02}	1.48 ^{+0.77} _{-0.72}	12.1					
2.4407 ^{+0.0206} _{-0.0035}	0.12 ^{+74.88} _{-0.12}	0.82 ^{+0.71} _{-0.66}	4.3					
2.4618 ^{+0.0099} _{-0.0107}	15.24 ^{+59.76} _{-15.24}	1.47 ^{+1.53} _{-0.94}	8.4					
2.6944 ^{+0.0200} _{-0.0200}	13.48 ^{+61.94} _{-13.48}	1.32 ^{+1.96} _{-0.85}	7.9	(Ca xix 1s ² 1s3p)	2.7050	46.3		-1171 ⁺⁷²³ ₋₂₂₁₇
2.7000 ^{+0.0025} _{-0.0000}	0.00 ^{+24.05} _{-0.00}	-1.64 ^{+0.53} _{-0.50}	23.9	(Ca xix 1s ² 1s3p)	2.7050	46.3	←	-554 ⁺²⁷⁷ ₋₀
2.9776 ^{+0.0238} _{-0.0162}	0.01 ^{+49.99} _{-0.01}	-0.53 ^{+0.53} _{-0.55}	2.6	Ar xviii 1s 4p	2.9873	7.04	←	-974 ⁺²³⁸⁹ ₋₁₆₂₅
				Ar xviii 1s 4p	2.9878	7.03		-1017 ⁺²³⁸⁹ ₋₁₆₂₅
3.0201 ^{+0.0175} _{-0.0026}	0.01 ^{+74.99} _{-0.01}	-0.60 ^{+0.56} _{-0.54}	3.1	Ca xx 1s 2p	3.0185	98.6	←	155 ⁺¹⁷³⁴ ₋₂₅₈
				Ca xx 1s 2p	3.0239	98.5		-383 ⁺¹⁷³¹ ₋₂₅₇
3.0748 ^{+0.0002} _{-0.0048}	0.00 ^{+8.77} _{-0.00}	-0.85 ^{+0.56} _{-0.53}	6.1	(Ar xvii 1s ² 1s5p)	3.1280	6.20		-5101 ⁺²³ ₋₄₅₈
3.1427 ^{+0.0213} _{-0.0187}	0.16 ^{+49.84} _{-0.16}	-0.44 ^{+0.44} _{-0.54}	1.4	Ar xviii 1s 3p	3.1502	17.3	←	-713 ⁺²⁰²³ ₋₁₇₈₄
				Ar xviii 1s 3p	3.1514	17.2		-822 ⁺²⁰²² ₋₁₇₈₃
3.1878 ^{+0.0163} _{-0.0306}	0.00 ^{+50.42} _{-0.00}	-0.55 ^{+0.55} _{-0.56}	2.2	Ca xix 1s ² 1s2p	3.1772	170	←	1001 ⁺¹⁵³⁸ ₋₂₈₉₂
3.1942 ^{+0.0033} _{-0.0018}	0.00 ^{+17.52} _{-0.00}	1.01 ^{+0.71} _{-0.68}	6.2	(Ca xix 1s ² 1s2p)	3.189	0.001	←	484 ⁺³⁰⁶ ₋₁₆₅
				(Ca xix 1s ² 1s2p)	3.192	4.85		141 ⁺³⁰⁶ ₋₁₆₄
3.2075 ^{+0.0025} _{-0.0050}	0.00 ^{+27.57} _{-0.00}	-0.93 ^{+0.51} _{-0.49}	8.0	(Ar xvii 1s ² 1s4p)	3.2000	12.3	←	702 ⁺²³⁵ ₋₄₆₈
3.3667 ^{+0.0158} _{-0.0067}	0.00 ^{+0.00} _{-0.00}	-0.60 ^{+0.59} _{-0.57}	2.8	Ar xvii 1s ² 1s3p	3.3650	30.0	←	154 ⁺¹⁴⁰⁵ ₋₅₉₉
3.7002 ^{+0.0202} _{-0.0202}	0.00 ^{+75.00} _{-0.00}	-0.41 ^{+0.41} _{-0.59}	1.2	S xvi 1s 5p	3.6958	2.22	←	355 ⁺¹⁶³⁷ ₋₁₆₃₇
				S xvi 1s 5p	3.6960	2.21		337 ⁺¹⁶³⁷ ₋₁₆₃₇
3.7289 ^{+0.0211} _{-0.0103}	12.88 ^{+37.12} _{-12.88}	-0.94 ^{+0.79} _{-1.63}	4.0	Ar xviii 1s 2p	3.7311	64.7	←	-179 ⁺¹⁶⁹⁷ ₋₈₃₁
				Ar xviii 1s 2p	3.7365	64.6		-614 ⁺¹⁶⁹⁴ ₋₈₃₀
3.7801 ^{+0.0061} _{-0.0051}	0.04 ^{+49.96} _{-0.04}	-0.64 ^{+0.58} _{-0.83}	3.3	S xvi 1s 4p	3.7843	4.40	←	-336 ⁺⁴⁸⁶ ₋₄₀₄
				S xvi 1s 4p	3.7848	4.39		-370 ⁺⁴⁸⁶ ₋₄₀₄
3.9250 ^{+0.0041} _{-0.0040}	8.53 ^{+11.89} _{-8.53}	-1.54 ^{+0.73} _{-0.82}	13.0					
3.9475 ^{+0.0026} _{-0.0026}	0.03 ^{+18.90} _{-0.03}	-1.11 ^{+0.54} _{-0.57}	9.1	Ar xvii 1s ² 1s2p	3.9491	109	←	-118 ⁺¹⁹⁴ ₋₂₀₁
3.9860 ^{+0.0214} _{-0.0181}	37.50 ^{+37.50} _{-37.50}	-1.73 ^{+1.43} _{-1.49}	5.9	S xvi 1s 3p	3.9908	10.8	←	-363 ⁺¹⁶⁰⁵ ₋₁₃₆₀
				S xvi 1s 3p	3.9920	10.8		-449 ⁺¹⁶⁰⁴ ₋₁₃₅₉
4.0984 ^{+0.0088} _{-0.0093}	20.58 ^{+31.95} _{-20.58}	-1.88 ^{+1.18} _{-1.24}	9.1	(S xv 1s ² 1s4p)	4.0883	7.53	←	738 ⁺⁶⁴⁸ ₋₆₈₂
4.3019 ^{+0.0211} _{-0.0211}	74.98 ^{+0.44} _{-64.42}	-3.88 ^{+2.47} _{-1.82}	12.4	S xv 1s ² 1s3p	4.2990	18.3	←	204 ⁺¹⁴⁶⁸ ₋₁₄₆₈
4.3876 ^{+0.0049} _{-0.0048}	0.14 ^{+16.57} _{-0.14}	-1.09 ^{+0.68} _{-0.84}	6.3	S xv 1s ² 2p (autoion.)	4.3910			-235 ⁺³³⁷ ₋₃₃₀
4.4150 ^{+0.0075} _{-0.0075}	0.00 ^{+75.00} _{-0.00}	-0.66 ^{+0.66} _{-0.71}	2.7	S xv 1s ² 2p (autoion.)	4.4149			8 ⁺⁵¹¹ ₋₅₁₁
4.7285 ^{+0.0015} _{-0.0017}	7.78 ^{+9.06} _{-5.76}	-4.13 ^{+0.92} _{-1.18}	93.6	S xvi 1s 2p	4.7274	40.4	←	71 ⁺⁹³ ₋₁₀₉
				(S xvi 1s 2p)	4.7328	40.3		-272 ⁺⁹³ ₋₁₀₉
4.9518 ^{+0.0032} _{-0.0018}	0.01 ^{+13.26} _{-0.01}	-1.42 ^{+0.69} _{-0.73}	11.2	(Si xiv 1s 4p)	4.9468	2.58	←	305 ⁺¹⁹⁶ ₋₁₁₂
				(Si xiv 1s 4p)	4.9472	2.57		279 ⁺¹⁹⁶ ₋₁₁₂
5.0397 ^{+0.0020} _{-0.0020}	9.26 ^{+7.70} _{-5.73}	-3.80 ^{+0.91} _{-1.05}	62.7	S xv 1s ² 1s2p	5.0387	66.7	←	61 ⁺¹²⁰ ₋₁₂₁
5.0656 ^{+0.0244} _{-0.0156}	5.27 ^{+44.73} _{-5.27}	0.77 ^{+0.93} _{-0.77}	2.2	S xv 1s ² 1s2p	5.063	0.000	←	143 ⁺¹⁴⁴⁷ ₋₉₂₁
				S xv 1s ² 1s2p	5.066	0.59		-55 ⁺¹⁴⁴⁶ ₋₉₂₁
5.1004 ^{+0.0196} _{-0.0204}	0.22 ^{+49.78} _{-0.22}	0.34 ^{+0.84} _{-0.34}	0.6	S xv 1s ² 1s2s	5.101	0.000	←	-64 ⁺¹¹⁵¹ ₋₁₁₉₉
5.2193 ^{+0.0077} _{-0.0048}	22.15 ^{+20.59} _{-11.94}	-4.03 ^{+1.41} _{-2.18}	33.5	Si xiv 1s 3p	5.2168	6.32	←	145 ⁺⁴⁴⁴ ₋₂₇₄
				Si xiv 1s 3p	5.2180	6.31		79 ⁺⁴⁴⁴ ₋₂₇₄
5.3750 ^{+0.0050} _{-0.0050}	0.00 ^{+37.24} _{-0.00}	-1.42 ^{+0.84} _{-0.76}	6.9					
5.4033 ^{+0.0043} _{-0.0008}	0.01 ^{+25.54} _{-0.01}	-1.42 ^{+0.77} _{-1.05}	8.7	Si xiii 1s ² 1s4p	5.4045	4.30	←	-68 ⁺²³⁷ ₋₄₅
5.5306 ^{+0.0044} _{-0.0007}	0.01 ^{+13.42} _{-0.01}	-1.44 ^{+0.80} _{-0.78}	8.5	(Si xiii 1s ² 2s (autoion.))	5.5424			-636 ⁺²³⁷ ₋₃₈
				(Si xiii 1s ² 2s (autoion.))	5.5425			-641 ⁺²³⁷ ₋₃₈

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å] $[10^{10}\text{s}^{-1}]$	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
5.5750 ^{+0.0050} _{-0.0025}	0.00 ^{+41.16} _{-0.00}	-1.41 ^{+0.68} _{-0.74}	10.7	(Si XIII (Si XIII	$1s^2 2p$ $1s^2 2p$	(autoion.) (autoion.)	5.5618 5.5627))	713 ⁺²⁶⁸ ₋₁₃₇ 665 ⁺²⁶⁸ ₋₁₃₇
5.6809 ^{+0.0037} _{-0.0036}	22.09 ^{+14.30} _{-10.29}	-4.64 ^{+1.29} _{-1.47}	58.5	Si XIII	$1s^2$	$1s3p$	5.6805	10.4 ←	23 ⁺¹⁹⁵ ₋₁₉₀
5.8572 ^{+0.0003} _{-0.0047}	0.00 ^{+26.23} _{-0.00}	-1.11 ^{+0.63} _{-0.63}	8.0	(Ni XXV (Ni XXV	$1s^2 2s^2$ $1s^2 2s^2$	$1s^2 2s7p$ $1s^2 2s7p$	5.8598 5.8584	0.19) 1.25)	-133 ⁺¹³ ₋₂₄₃ -60 ⁺¹³ ₋₂₄₃
6.0602 ^{+0.0121} _{-0.0093}	14.48 ^{+5.52} _{-14.48}	-1.28 ^{+0.87} _{-1.05}	7.0	Al XIII Al XIII	$1s$ $1s$	$3p$ $3p$	6.0526 6.0537	4.70 ← 4.69	376 ⁺⁵⁹⁸ ₋₄₅₉ 321 ⁺⁵⁹⁸ ₋₄₅₉
6.1810 ^{+0.0005} _{-0.0009}	13.80 ^{+2.31} _{-1.75}	-10.31 ^{+0.78} _{-0.85}	888.9	Si XIV (Si XIV	$1s$ $1s$	$2p$ $2p$	6.1804 6.1858	23.7 ← 23.6)	29 ⁺²⁵ ₋₄₂ -233 ⁺²⁵ ₋₄₂
6.3150 ^{+0.0150} _{-0.0050}	0.00 ^{+50.00} _{-0.00}	-0.37 ^{+0.37} _{-0.65}	1.3	Al XII	$1s^2$	$1s4p$	6.3140	3.14 ←	48 ⁺⁷¹¹ ₋₂₃₈
6.6346 ^{+0.0045} _{-0.0024}	0.88 ^{+10.91} _{-0.88}	-1.81 ^{+1.10} _{-2.12}	0.0	Al XII	$1s^2$	$1s3p$	6.6350	7.63 ←	-18 ⁺²⁰¹ ₋₁₀₉
6.6468 ^{+0.0009} _{-0.0010}	9.50 ^{+2.50} _{-2.18}	-6.62 ^{+1.01} _{-1.07}	215.5	(Si XIII	$1s^2$	$1s2p$	6.6479	37.7) ←	-51 ⁺⁴² ₋₄₄
6.7031 ^{+0.0036} _{-0.0040}	18.94 ^{+16.40} _{-12.14}	-3.67 ^{+1.57} _{-1.30}	45.4	Ni XXIV (Si XIII (Si XIII	$1s^2 2s^2 2p$ $1s^2$ $1s^2$	$1s^2 2s2p5p$ $1s2p$ $1s2p$	6.7029 6.6850 6.6882	0.17 0.000) 0.16)	9 ⁺¹⁶¹ ₋₁₈₁ 810 ⁺¹⁶¹ ₋₁₈₁ 666 ⁺¹⁶¹ ₋₁₈₁
6.7223 ^{+0.0033} _{-0.0034}	14.39 ^{+10.73} _{-6.92}	-2.96 ^{+0.95} _{-1.07}	37.2	Ni XXIV	$1s^2 2s^2 2p$	$1s^2 2s^2 6d$	6.7190	2.11	148 ⁺¹⁴⁶ ₋₁₅₁
6.7445 ^{+0.0023} _{-0.0019}	7.50 ^{+6.33} _{-7.50}	2.77 ^{+0.94} _{-0.85}	36.1	(Si XIII	$1s^2$	$1s2s$	6.740	0.000) ←	189 ⁺¹⁰² ₋₈₆
6.7613 ^{+0.0041} _{-0.0047}	6.57 ^{+15.96} _{-6.57}	1.35 ^{+0.89} _{-0.72}	9.9	Ni XXIV Ni XXIV	$1s^2 2s2p^2$ $1s^2 2s2p^2$	$1s^2 2s2p6d$ $1s^2 2s2p6d$	6.759 6.759	2.86 1.58	95 ⁺¹⁸² ₋₂₁₀ 64 ⁺¹⁸² ₋₂₁₀
6.7800 ^{+0.0024} _{-0.0021}	0.03 ^{+14.37} _{-0.03}	-1.66 ^{+0.39} _{-0.66}	28.7	Ni XXV (Fe XXV (Fe XXV (Fe XXV	$1s^2 2p^2$ $1s2s$ $1s2s$ $1s2s$	$1s^2 2p5d$ $1s5p$ $1s5p$ $1s5p$	6.7796 6.7880 6.7880 6.7880	3.87 1.98) 1.95) 1.94)	19 ⁺¹⁰⁸ ₋₉₅ -353 ⁺¹⁰⁷ ₋₉₅ -353 ⁺¹⁰⁷ ₋₉₅ -353 ⁺¹⁰⁷ ₋₉₅
7.1050 ^{+0.0025} _{-0.0000}	0.01 ^{+5.89} _{-0.01}	-2.31 ^{+0.43} _{-0.72}	49.6	Mg XII Mg XII	$1s$ $1s$	$3p$ $3p$	7.1058 7.1069	3.41 ← 3.41	-33 ⁺¹⁰⁵ ₋₁ -81 ⁺¹⁰⁵ ₋₁
7.1687 ^{+0.0027} _{-0.0020}	9.91 ^{+7.80} _{-9.90}	-3.04 ^{+0.79} _{-0.99}	46.2	Al XIII (Al XIII	$1s$ $1s$	$2p$ $2p$	7.1710 7.1764	17.6 ← 17.6)	-98 ⁺¹¹³ ₋₈₃ -323 ⁺¹¹³ ₋₈₃
7.2777 ^{+0.0116} _{-0.0077}	20.00 ^{+32.33} _{-13.07}	2.06 ^{+1.48} _{-1.29}	11.6	Ni XXIV	$1s^2 2s2p^2$	$1s^2 2s2p5d$	7.278	6.10	-46 ⁺⁴⁷⁹ ₋₃₁₅
7.3168 ^{+0.0200} _{-0.0147}	6.36 ^{+43.64} _{-6.36}	1.17 ^{+1.94} _{-0.82}	6.0	Mg XI	$1s^2$	$1s5p$	7.310	1.13	274 ⁺⁸²⁰ ₋₆₀₃
7.3525 ^{+0.0050} _{-0.0075}	0.00 ^{+75.00} _{-0.00}	1.02 ^{+0.84} _{-0.67}	6.2	(Ni XXV (Ni XXV	$1s^2 2p^2$ $1s^2 2p^2$	$1s^2 2p4d$ $1s^2 2p4d$	7.345 7.359	9.01 8.55)	289 ⁺²⁰⁴ ₋₃₀₆ -269 ⁺²⁰⁴ ₋₃₀₅
7.4774 ^{+0.0001} _{-0.0049}	0.00 ^{+11.26} _{-0.00}	-1.80 ^{+0.56} _{-0.52}	24.9	Mg XI (Fe XXIII	$1s^2$ $1s^2 2s^2$	$1s4p$ $1s^2 2s5p$	7.4730 7.4780	2.24 ← 2.51)	175 ⁺⁵ ₋₁₉₆ -25 ⁺⁵ ₋₁₉₅
7.6245 ^{+0.0043} _{-0.0104}	14.77 ^{+35.42} _{-11.65}	2.69 ^{+1.82} _{-1.03}	22.9	Ni XXIII Ni XXIII (Ni XXI	$1s^2 2s2p^3$ $1s^2 2s2p^3$ $2s^2 2p^4$	$1s^2 2s2p2p5$ $1s^2 2s2p2p5$ $2s2p^2 2p^2 5p$	7.625 7.628 7.629	0.85 1.61 0.97)	-43 ⁺¹⁶⁸ ₋₄₀₈ -148 ⁺¹⁶⁸ ₋₄₀₈ -210 ⁺¹⁶⁸ ₋₄₀₈
7.7532 ^{+0.0056} _{-0.0090}	0.41 ^{+34.00} _{-0.41}	-1.09 ^{+0.75} _{-1.49}	5.6	Al XII	$1s^2$	$1s2p$	7.7573	27.5 ←	-158 ⁺²¹⁷ ₋₃₄₇
7.7676 ^{+0.0049} _{-0.0026}	0.00 ^{+18.45} _{-0.00}	-1.15 ^{+0.78} _{-0.70}	5.9	(Al XII	$1s^2$	$1s2p$	7.7573	27.5) ←	398 ⁺¹⁹⁰ ₋₁₀₁
7.7908 ^{+0.0067} _{-0.0033}	0.00 ^{+39.61} _{-0.00}	1.46 ^{+0.97} _{-0.94}	6.7	(Al XII (Al XII	$1s^2$ $1s^2$	$1s2p$ $1s2p$	7.807 7.803	0.082) ← 0.000)	-620 ⁺²⁵⁷ ₋₁₂₇ -501 ⁺²⁵⁷ ₋₁₂₇
7.8150 ^{+0.0092} _{-0.0211}	7.03 ^{+42.97} _{-7.03}	1.26 ^{+1.74} _{-1.04}	4.1	Al XII Al XII	$1s^2$ $1s^2$	$1s2p$ $1s2p$	7.807 7.803	0.082 ← 0.000	310 ⁺³⁵² ₋₈₀₉ 430 ⁺³⁵² ₋₈₁₀
7.8482 ^{+0.0022} _{-0.0025}	14.35 ^{+8.95} _{-6.36}	-4.78 ^{+1.30} _{-1.55}	62.4	Mg XI	$1s^2$	$1s3p$	7.8503	5.43 ←	-81 ⁺⁸⁵ ₋₉₅
7.8751 ^{+0.0200} _{-0.0200}	0.02 ^{+49.98} _{-0.02}	0.47 ^{+1.03} _{-0.47}	1.2	Al XII	$1s^2$	$1s2s$	7.872	0.000 ←	112 ⁺⁷⁶² ₋₇₆₂
7.9040 ^{+0.0072} _{-0.0143}	13.39 ^{+38.70} _{-13.39}	-1.30 ^{+0.88} _{-1.27}	6.7	(Fe XXII (Fe XXII (Ni XXIV (Ni XXIV (Ni XXIV (Fe XXII (Ni XXII	$1s^2 2s^2 2p$ $1s^2 2s^2 2p$ $1s^2 2s2p^2$ $1s^2 2s2p^2$ $1s^2 2s2p^2$ $1s^2 2s^2 2p$ $2s^2 2p^3$	$1s^2 2s2p5p$ $1s^2 2s2p5p$ $1s^2 2s2p4d$ $1s^2 2s2p4d$ $1s^2 2s2p4d$ $1s^2 2s2p5p$ $2s^2 2p^2 5d$	7.8806 7.8838 7.8844 7.8851 7.8872 7.8883 7.8892	1.82) 1.49) 7.31) 5.30) 2.73) 1.20) 2.70)	888 ⁺²⁷⁴ ₋₅₄₄ 765 ⁺²⁷⁴ ₋₅₄₄ 745 ⁺²⁷⁴ ₋₅₄₄ 717 ⁺²⁷⁴ ₋₅₄₄ 638 ⁺²⁷⁴ ₋₅₄₄ 597 ⁺²⁷⁴ ₋₅₄₄ 560 ⁺²⁷⁴ ₋₅₄₄
7.9250 ^{+0.0050} _{-0.0025}	0.00 ^{+26.40} _{-0.00}	-0.82 ^{+0.71} _{-0.57}	3.7	(Ni XXII (Ni XXII (Ni XXII (Ni XXII (Ni XXII (Ni XXII	$2s^2 2p^3$ $2s^2 2p^3$ $2s^2 2p^3$ $2s^2 2p^3$ $2s^2 2p^3$ $2s^2 2p^3$	$2s^2 2p2p5d$ $2s^2 2p2p5d$ $2s^2 2p^2 5d$ $2s2p^2 2p5p$ $2s^2 2p^2 5d$ $2s^2 2p^2 5d$	7.9065 7.9076 7.9097 7.9137 7.9144 7.9146	4.31) 3.38) 1.56) 1.27) 2.24) 1.22)	701 ⁺¹⁹⁰ ₋₉₅ 660 ⁺¹⁹⁰ ₋₉₅ 580 ⁺¹⁹⁰ ₋₉₅ 429 ⁺¹⁸⁹ ₋₉₅ 401 ⁺¹⁸⁹ ₋₉₅ 393 ⁺¹⁸⁹ ₋₉₅
7.9607 ^{+0.0052} _{-0.0040}	6.29 ^{+15.26} _{-6.29}	1.65 ^{+1.26} _{-0.80}	12.9	(Ni XXII Ni XXIII	$2s2p^4$ $1s^2 2s^2 2p^2$	$2s2p2p^2 5d$ $1s^2 2s2p^2 4p$	7.952 7.960	4.34) 3.66	295 ⁺¹⁹⁵ ₋₁₅₂ 22 ⁺¹⁹⁵ ₋₁₅₂
7.9745 ^{+0.0005} _{-0.0020}	0.00 ^{+12.13} _{-0.00}	2.04 ^{+0.90} _{-0.84}	20.3	Ni XXIV	$1s^2 2s2p^2$	$1s^2 2s2p4d$	7.972	4.32	74 ⁺¹⁸ ₋₇₆

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion $i j$	transition λ_0 [Å] $[10^{12}\text{s}^{-1}]$	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]	
7.9900 ^{+0.0024} _{-0.0032}	0.15 ^{+16.77} _{-0.15}	-1.77 ^{+0.68} _{-0.92}	21.2	(Fe XXIV	$\frac{1s^2 2s}{1s^2 2s}$	$1s^2 4p$	7.9857 3.24) ←	160 ⁺⁸⁹ ₋₁₂₂
				(Fe XXIV	$\frac{1s^2 2s}{1s^2 2s}$	$1s^2 4p$	7.9960 3.30)	-226 ⁺⁸⁹ ₋₁₂₁
8.0291 ^{+0.0077} _{-0.0044}	0.14 ^{+37.27} _{-0.14}	0.91 ^{+0.80} _{-0.72}	4.4	Ni XXII	$\frac{2s2p^4}{2s^2 2p^3}$	$\frac{2s2p^2 2p5d}{2p2p^3 4d}$	8.032 [!] 1.61	-116 ⁺²⁸⁷ ₋₁₆₃
				Ni XXII	$\frac{2s^2 2p^3}{2s^2 2p^2 5d}$		8.034 [!] 1.43	-189 ⁺²⁸⁷ ₋₁₆₃
				Ni XXII	$\frac{2s^2 2p^3}{2s^2 2p^2 5d}$		8.034 [!] 3.89	-210 ⁺²⁸⁷ ₋₁₆₃
8.0485 ^{+0.0056} _{-0.0044}	2.93 ^{+26.44} _{-2.93}	1.58 ^{+0.71} _{-0.89}	10.2	(Ni XXIV	$\frac{1s^2 2s2p^2}{1s^2 2s2p^2}$	$\frac{1s^2 2s2p4d}{1s^2 2s2p4d}$	8.043 [!] 4.10)	178 ⁺²⁰⁷ ₋₁₆₃
				Ni XXIV	$\frac{1s^2 2s2p^2}{1s^2 2s2p^2}$	$\frac{1s^2 2s2p4d}{1s^2 2s2p4d}$	8.048 [!] 3.83	1 ⁺²⁰⁷ ₋₁₆₃
				Ni XXII	$\frac{2s2p^4}{2s2p^3 5d}$		8.049 [!] 4.98	-33 ⁺²⁰⁷ ₋₁₆₃
				Ni XXIII	$\frac{1s^2 2s^2 2p^2}{1s^2 2s2p2p4}$		8.049 [!] 3.52	-40 ⁺²⁰⁷ ₋₁₆₃
				Ni XXIV	$\frac{1s^2 2s2p^2}{1s^2 2s2p4d}$		8.053 [!] 9.77	-177 ⁺²⁰⁷ ₋₁₆₃
8.1000 ^{+0.0200} _{-0.0200}	0.00 ^{+75.42} _{-0.00}	0.61 ^{+0.84} _{-0.61}	2.2	Fe XXII	$\frac{1s^2 2s2p^2}{1s^2 2s2p^2}$	$\frac{1s^2 2s2p5d}{1s^2 2s2p5d}$	8.106 [!] 3.83	-232 ⁺⁷⁴⁰ ₋₇₄₀
				Fe XXII	$\frac{1s^2 2s2p^2}{1s^2 2s2p^2}$	$\frac{1s^2 2s2p5d}{1s^2 2s2p5d}$	8.107 [!] 2.19	-284 ⁺⁷⁴⁰ ₋₇₄₀
8.1325 ^{+0.0399} _{-0.0399}	0.00 ^{+50.00} _{-0.00}	0.44 ^{+0.78} _{-0.44}	1.0	Ni XXI	$\frac{2s2p^5}{2s2p^5}$	$\frac{2s2p^4 5d}{2s2p^4 5d}$	8.115 [!] 3.28	621 ⁺¹⁴⁷⁵ ₋₁₄₇₅
				Ni XXI	$\frac{2s2p^5}{2s2p^5}$	$\frac{2s2p^4 5d}{2s2p^4 5d}$	8.117 [!] 1.65	555 ⁺¹⁴⁷⁵ ₋₁₄₇₅
8.1625 ^{+0.0050} _{-0.0025}	0.00 ^{+14.67} _{-0.00}	0.95 ^{+0.78} _{-0.81}	3.9	(Ni XXII	$\frac{2s2p^4}{2p^6}$	$\frac{2s2p^3 5d}{2p2p^4 5d}$	8.166 [!] 3.16)	-159 ⁺¹⁸² ₋₉₃
				(Ni XXI	$\frac{2p^6}{2p2p^4 5d}$		8.168 [!] 5.24)	-205 ⁺¹⁸² ₋₉₃
				(Fe XXII	$\frac{1s^2 2s^2 2p}{1s^2 2s^2 5d}$		8.168 [!] 2.88)	-215 ⁺¹⁸² ₋₉₃
8.2201 ^{+0.0050} _{-0.0026}	0.00 ^{+22.48} _{-0.00}	1.13 ^{+0.87} _{-0.84}	5.1	(Ni XXIII	$\frac{1s^2 2s2p^3}{1s^2 2s2p^2 4d}$		8.220 [!] 1.21)	-3 ⁺¹⁸¹ ₋₉₃
				(Ni XXII	$\frac{2s2p^4}{2p2p^3 4p}$		8.226 [!] 1.95)	-217 ⁺¹⁸⁰ ₋₉₃
				(Ni XXIII	$\frac{1s^2 2s2p^3}{1s^2 2s2p^2 4d}$		8.227 [!] 1.50)	-259 ⁺¹⁸⁰ ₋₉₃
				(Ni XXII	$\frac{2s^2 2p^3}{2s2p2p^2 4p}$		8.228 [!] 1.09)	-294 ⁺¹⁸⁰ ₋₉₃
8.2726 ^{+0.0074} _{-0.0026}	0.01 ^{+16.66} _{-0.01}	1.10 ^{+0.83} _{-0.85}	4.7	Ni XXIII	$\frac{1s^2 2s2p^3}{1s^2 2s2p2p4}$		8.276 [!] 11.4	-139 ⁺²⁶⁸ ₋₉₄
				Fe XXII	$\frac{1s^2 2s2p^2}{1s^2 2s2p^2}$	$\frac{1s^2 2s2p5d}{1s^2 2s2p5d}$	8.274 [!] 4.53	-51 ⁺²⁶⁸ ₋₉₄
				Ni XXII	$\frac{2s2p^4}{2s2p^2 2p5d}$		8.272 [!] 3.65	-12 ⁺²⁶⁸ ₋₉₄
8.3062 ^{+0.0038} _{-0.0014}	0.02 ^{+19.98} _{-0.02}	-1.72 ^{+0.66} _{-0.60}	17.6	(Fe XXIII	$\frac{1s^2 2s^2}{1s^2 2s4p}$		8.3038 4.66) ←	88 ⁺¹³⁵ ₋₄₉
8.3808 ^{+0.0045} _{-0.0100}	0.08 ^{+50.32} _{-0.08}	1.53 ^{+0.87} _{-0.87}	9.1	(Ni XXIII	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p4d}$		8.384 [!] 5.08)	-140 ⁺¹⁶⁰ ₋₃₅₈
				(Ni XXIII	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p4d}$		8.389 [!] 14.0)	-313 ⁺¹⁶⁰ ₋₃₅₇
8.4001 ^{+0.0049} _{-0.0053}	0.10 ^{+32.23} _{-0.10}	-0.86 ^{+0.61} _{-0.95}	5.0	(Ni XXI	$\frac{2s^2 2p^4}{2s2p2p^3 4p}$		8.3958 0.11)	154 ⁺¹⁷⁷ ₋₁₈₇
				(Ni XXIII	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p4d}$		8.4051 2.65)	-179 ⁺¹⁷⁶ ₋₁₈₇
				(Ni XXIII	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p4d}$		8.3896 14.0)	375 ⁺¹⁷⁷ ₋₁₈₈
8.4203 ^{+0.0007} _{-0.0008}	15.23 ^{+2.60} _{-1.66}	-11.98 ^{+0.90} _{-1.12}	780.0	(Mg XII	$\frac{1s}{2p}$		8.4192 12.8) ←	39 ⁺²⁶ ₋₂₉
				(Mg XII	$\frac{1s}{2p}$		8.4246 12.8)	-154 ⁺²⁶ ₋₂₉
8.5748 ^{+0.0061} _{-0.0051}	10.91 ^{+18.46} _{-10.91}	-1.75 ^{+0.90} _{-1.14}	11.0	Fe XXI	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p^2}$	$\frac{1s^2 2s^2 2p5d}{1s^2 2s^2 2p5d}$	8.5740 2.85 ←	28 ⁺²¹⁴ ₋₁₈₀
				Fe XXI	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p^2}$	$\frac{1s^2 2s^2 2p5d}{1s^2 2s^2 2p5d}$	8.5740 2.43	28 ⁺²¹⁴ ₋₁₈₀
				Fe XXI	$\frac{1s^2 2s^2 2p^2}{1s^2 2s^2 2p^2}$	$\frac{1s^2 2s^2 2p5d}{1s^2 2s^2 2p5d}$	8.5740 1.55	28 ⁺²¹⁴ ₋₁₈₀
8.5895 ^{+0.0040} _{-0.0041}	6.40 ^{+11.03} _{-6.40}	2.11 ^{+1.08} _{-0.93}	14.9	Fe XXI	$\frac{1s^2 2s2p^3}{1s^2 2s2p^3}$	$\frac{1s^2 2s2p^2 5d}{1s^2 2s2p^2 5d}$	8.589 [!] 1.67	5 ⁺¹³⁸ ₋₁₄₅
				Fe XXI	$\frac{1s^2 2s2p^3}{1s^2 2s2p^3}$	$\frac{1s^2 2s2p^2 5d}{1s^2 2s2p^2 5d}$	8.591 [!] 1.31	-66 ⁺¹³⁸ ₋₁₄₅
8.6200 ^{+0.0050} _{-0.0025}	0.00 ^{+16.42} _{-0.00}	1.32 ^{+1.20} _{-0.68}	8.6	Ni XXIII	$\frac{1s^2 2s2p^3}{1s^2 2s2p2p4}$		8.617 [!] 3.42	84 ⁺¹⁷⁴ ₋₈₇
				Ni XXIII	$\frac{1s^2 2s2p^3}{1s^2 2s2p2p4}$		8.620 [!] 4.89	-14 ⁺¹⁷⁴ ₋₈₇
				Ni XXIII	$\frac{1s^2 2s2p^3}{1s^2 2s2p2p4}$		8.623 [!] 4.29	-122 ⁺¹⁷⁴ ₋₈₇
				(Fe XXIII	$\frac{1s^2 2s2p}{1s^2 2s4d}$		8.617 [!] 7.04)	98 ⁺¹⁷⁴ ₋₈₇
8.7141 ^{+0.0128} _{-0.0192}	21.69 ^{+28.31} _{-21.69}	-1.77 ^{+1.49} _{-0.98}	5.1	(Ni XXVII	$\frac{1s2p}{1s3d}$		8.7331 2.2e+05)	-654 ⁺⁴³⁹ ₋₆₅₈
				Ni XXVII	$\frac{1s2p}{1s3d}$		8.7069 1.7e+05	245 ⁺⁴⁴⁰ ₋₆₆₀
				(Ni XXVII	$\frac{1s2p}{1s3d}$		8.7331 1.1e+05)	-654 ⁺⁴³⁹ ₋₆₅₈
				Ni XXVII	$\frac{1s2p}{1s3d}$		8.7135 1.5e+04	21 ⁺⁴⁴⁰ ₋₆₆₀
				Ni XXVII	$\frac{1s2s}{1s3p}$		8.7265 10.2	-429 ⁺⁴³⁹ ₋₆₅₉
				Ni XXII	$\frac{2s^2 2p^3}{2s^2 2p2p4d}$		8.7204 8.82	-217 ⁺⁴⁴⁰ ₋₆₅₉
				Ni XXII	$\frac{2s^2 2p^3}{2s^2 2p^2 4d}$		8.7227 7.80	-296 ⁺⁴⁴⁰ ₋₆₅₉
8.7403 ^{+0.0046} _{-0.0029}	0.00 ^{+17.16} _{-0.00}	-1.11 ^{+0.76} _{-0.63}	6.0	(Fe XXII	$\frac{1s^2 2s^2 2p}{1s^2 2s2p4p}$		8.7254 3.54) ←	513 ⁺¹⁶⁰ ₋₉₈
				(Fe XXII	$\frac{1s^2 2s^2 2p}{1s^2 2s2p4p}$		8.7360 1.31)	149 ⁺¹⁶⁰ ₋₉₈
8.7862 ^{+0.0115} _{-0.0084}	0.09 ^{+74.91} _{-0.09}	0.87 ^{+0.88} _{-0.81}	3.0	Ni XXII	$\frac{2s2p^4}{2s2p^3 4d}$		8.778 [!] 4.38	254 ⁺³⁹³ ₋₂₈₈
				Ni XXII	$\frac{2s2p^4}{2s2p^3 4d}$		8.782 [!] 1.42	144 ⁺³⁹³ ₋₂₈₈
				Ni XX	$\frac{2s2p^6}{2s2p^2 2p^3 5d}$		8.782 [!] 5.42	121 ⁺³⁹² ₋₂₈₈
				Ni XXII	$\frac{2s^2 2p^3}{2s^2 2p^2 4d}$		8.783 [!] 1.08	86 ⁺³⁹² ₋₂₈₈
				Fe XXI	$\frac{1s^2 2s2p^3}{1s^2 2s2p^2 5d}$		8.784 [!] 1.58	49 ⁺³⁹² ₋₂₈₈
				Ni XX	$\frac{2s2p^6}{2s2p^2 2p^3 5d}$		8.789 [!] 2.22	-122 ⁺³⁹² ₋₂₈₈
				Ni XXII	$\frac{2s2p^4}{2s2p2p^2 4d}$		8.790 [!] 1.00	-141 ⁺³⁹² ₋₂₈₈
8.8924 ^{+0.0040} _{-0.0074}	2.97 ^{+14.38} _{-2.97}	1.48 ^{+0.99} _{-0.85}	8.2	Ni XXII	$\frac{2s2p^4}{2s2p2p^2 4d}$		8.886 [!] 9.27	204 ⁺¹³⁵ ₋₂₅₁
				Ni XXII	$\frac{2s2p^4}{2s2p2p^2 4d}$		8.890 [!] 5.52	57 ⁺¹³⁵ ₋₂₅₁
				Ni XXII	$\frac{2p^5}{2p2p^3 4d}$		8.891 [!] 6.91	42 ⁺¹³⁵ ₋₂₅₁

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å] 10^{12}s^{-1}	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
8.9125 ^{+0.0150} _{-0.0100}	0.00 ^{+75.00} _{-0.00}	0.96 ^{+0.79} _{-0.95}	2.8	Ni XXI	2s2p ⁵	2s2p ⁴ 4d	8.905	8.11	235 ⁺⁵⁰⁶ ₋₃₃₆
				Ni XXII	2s ² 2p ³	2s ² 2p ² 4d	8.906	9.54	200 ⁺⁵⁰⁶ ₋₃₃₆
				Ni XXII	2s2p ⁴	2s2p ³ 4d	8.908	6.78	132 ⁺⁵⁰⁵ ₋₃₃₆
				Ni XXI	2s ² 2p ⁴	2s ² 2p ³ 4d	8.920	7.75	-277 ⁺⁵⁰⁵ ₋₃₃₅
8.9275 ^{+0.0025} _{-0.0075}	0.00 ^{+52.16} _{-0.00}	1.40 ^{+0.89} _{-0.87}	7.3	Ni XXI	2s ² 2p ⁴	2s ² 2p2p ² 4d	8.921	3.18	189 ⁺⁸⁵ ₋₂₅₁
				Fe XXIII	1s ² 2p ²	1s ² 2p4d	8.929	6.22	-67 ⁺⁸⁵ ₋₂₅₁
8.9775 ^{+0.0002} _{-0.0030}	0.05 ^{+21.73} _{-0.05}	-2.73 ^{+0.67} _{-0.22}	41.6	Fe XXII	1s ² 2s ² 2p	1s ² 2s ² 4d	8.9748	5.30	← 89 ⁺⁵ ₋₁₀₀
9.0478 ^{+0.0075} _{-0.0079}	9.45 ^{+18.09} _{-9.45}	1.29 ^{+1.17} _{-0.90}	5.4	Ni XXII	2s2p ⁴	2s2p ² 2p4d	9.040	8.33	229 ⁺²⁵⁰ ₋₂₆₁
				Ni XXI	2s ² 2p ⁴	2s ² 2p2p ² 4d	9.041	8.23	200 ⁺²⁵⁰ ₋₂₆₁
				Ni XXI	2s ² 2p ⁴	2s ² 2p ³ 4d	9.044	7.60	99 ⁺²⁵⁰ ₋₂₆₁
				Ni XXI	2s ² 2p ⁴	2s ² 2p2p ² 4d	9.046	6.43	31 ⁺²⁵⁰ ₋₂₆₁
				Ni XXI	2s ² 2p ⁴	2s ² 2p ³ 4d	9.053	12.8	-179 ⁺²⁵⁰ ₋₂₆₁
9.1672 ^{+0.0011} _{-0.0016}	17.02 ^{+4.41} _{-3.44}	-8.97 ^{+1.14} _{-1.18}	293.7	(Mg XI	1s ²	1s2p	9.1687	19.5	← -52 ⁺³⁷ ₋₅₃
9.1918 ^{+0.0032} _{-0.0048}	13.21 ^{+8.85} _{-6.83}	-3.35 ^{+1.28} _{-1.02}	35.3	Fe XXI	1s ² 2s ² 2p ²	1s ² 2s2p ² 4p	9.1944	2.88	← -85 ⁺¹⁰⁵ ₋₁₅₆
				(Fe XX	2s ² 2p ³	2s2p2p ² 4p	9.1979	1.04	-200 ⁺¹⁰⁵ ₋₁₅₆
				(Fe XX	2s ² 2p ³	2s ² 2p ² 5d	9.1979	0.49	-197 ⁺¹⁰⁵ ₋₁₅₆
9.2350 ^{+0.0000} _{-0.0050}	0.00 ^{+12.46} _{-0.00}	2.02 ^{+1.00} _{-0.95}	13.2	(Mg XI	1s ²	1s2p	9.228	0.000	← 222 ⁺⁰ ₋₁₆₂
				Mg XI	1s ²	1s2p	9.231	0.034	123 ⁺⁰ ₋₁₆₂
9.2600 ^{+0.0028} _{-0.0031}	0.09 ^{+14.38} _{-0.09}	1.85 ^{+0.88} _{-0.94}	10.8	Ni XX	2s2p ⁶	2s2p2p ⁴ 4d	9.261	7.19	-57 ⁺⁸⁹ ₋₁₀₁
				(Fe XXII	1s ² 2s2p ²	1s ² 2s2p4d	9.263	5.69	-95 ⁺⁸⁹ ₋₁₀₁
				(Ni XX	2s2p ⁶	2s2p ² 2p ³ 4d	9.264	4.87	-143 ⁺⁸⁹ ₋₁₀₁
				(Ni XXV	1s ² 2s2p	1s ² 2p3p	9.268	8.79	-261 ⁺⁸⁹ ₋₁₀₁
9.2784 ^{+0.0038} _{-0.0049}	5.57 ^{+21.39} _{-5.57}	-1.94 ^{+0.83} _{-1.28}	15.0	Ni XXI	2s ² 2p ⁴	2s ² 2p ² 2p4s	9.2763	0.86	68 ⁺¹²⁴ ₋₁₅₇
				Fe XX	2s2p ⁴	2s2p2p ² 5d	9.2788	2.21	-11 ⁺¹²⁴ ₋₁₅₇
				Fe XX	2s2p ⁴	2s2p2p ² 5d	9.2792	2.08	-24 ⁺¹²⁴ ₋₁₅₇
				Fe XX	2s2p ⁴	2s2p ³ 5d	9.2812	2.42	-90 ⁺¹²⁴ ₋₁₅₇
9.3100 ^{+0.0050} _{-0.0000}	0.00 ^{+16.24} _{-0.00}	2.45 ^{+1.05} _{-0.99}	17.9	Mg XI	1s ²	1s2s	9.314	0.000	-140 ⁺¹⁶¹ ₋₀
9.3311 ^{+0.0073} _{-0.0087}	13.00 ^{+34.72} _{-13.00}	1.94 ^{+1.46} _{-1.12}	9.0	Ni XXV	1s ² 2s2p	1s ² 2p3p	9.323	1.36	238 ⁺²³⁵ ₋₂₈₀
				Fe XX	2s2p ⁴	2p2p ³ 4p	9.331	1.33	4 ⁺²³⁵ ₋₂₈₀
				Fe XX	2s2p ⁴	2p2p ³ 4p	9.324	0.88	207 ⁺²³⁵ ₋₂₈₀
				(Fe XX	2s ² 2p ³	2s2p ³ 4p	9.321	0.87	297 ⁺²³⁵ ₋₂₈₀
				Fe XXI	1s ² 2s2p ³	1s ² 2s2p ² 4d	9.323	0.83	252 ⁺²³⁵ ₋₂₈₀
9.3676 ^{+0.0062} _{-0.0062}	22.07 ^{+21.35} _{-12.07}	-3.18 ^{+1.26} _{-1.50}	21.0						
9.3956 ^{+0.0083} _{-0.0038}	2.39 ^{+30.98} _{-2.39}	-1.69 ^{+0.87} _{-1.50}	11.0	(Ni XXV	1s ² 2s ²	1s ² 2s3p	9.3900	6.30) 178 ⁺²⁶⁷ ₋₁₂₁
9.4048 ^{+0.0004} _{-0.0073}	0.03 ^{+19.97} _{-0.03}	1.46 ^{+0.97} _{-0.93}	6.9	Ni XXV	1s ² 2s2p	1s ² 2p3p	9.399	9.21	178 ⁺¹⁴ ₋₂₃₂
				Fe XX	2s2p ⁴	2s2p ² 2p5d	9.404	3.65	19 ⁺¹⁴ ₋₂₃₂
				Fe XXI	1s ² 2s2p ³	1s ² 2p2p ² 4p	9.402	2.09	72 ⁺¹⁴ ₋₂₃₂
9.4745 ^{+0.0023} _{-0.0012}	11.02 ^{+4.74} _{-8.19}	-6.05 ^{+1.12} _{-1.11}	131.3	(Ne x	1s	5p	9.4807	0.34	← -195 ⁺⁷³ ₋₃₈
				(Ne x	1s	5p	9.4809	0.34	-202 ⁺⁷³ ₋₃₈
				(Fe XXI	1s ² 2s ² 2p ²	1s ² 2s ² 2p4d	9.4797	6.12	-164 ⁺⁷³ ₋₃₈
9.5116 ^{+0.0043} _{-0.0046}	12.29 ^{+17.79} _{-12.29}	3.14 ^{+1.46} _{-1.25}	19.9	(Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.517	4.39	-194 ⁺¹³⁶ ₋₁₄₆
				Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.512	4.02	-13 ⁺¹³⁶ ₋₁₄₆
				Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.514	2.45	-93 ⁺¹³⁶ ₋₁₄₆
				Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.514	2.33	-75 ⁺¹³⁶ ₋₁₄₆
9.5380 ^{+0.0050} _{-0.0005}	0.01 ^{+27.19} _{-0.01}	2.46 ^{+1.11} _{-1.05}	16.6	(Fe XXVI	2s	3p	9.536	10.1) 53 ⁺¹⁵⁶ ₋₁₆
9.7080 ^{+0.0028} _{-0.0020}	23.47 ^{+4.99} _{-5.70}	-8.57 ^{+1.38} _{-1.29}	153.7	Ne x	1s	4p	9.7080	0.67	← -1 ⁺⁸⁵ ₋₆₂
				Ne x	1s	4p	9.7085	0.67	-15 ⁺⁸⁵ ₋₆₂
9.8161 ^{+0.0043} _{-0.0043}	9.07 ^{+15.05} _{-9.07}	2.77 ^{+1.26} _{-1.25}	15.6	Fe XXI	1s ² 2s2p ³	1s ² 2s2p ² 4d	9.818	4.90	-82 ⁺¹³¹ ₋₁₃₂
				Fe XXI	1s ² 2s2p ³	1s ² 2s2p ² 4d	9.819	2.45	-96 ⁺¹³¹ ₋₁₃₂
				Fe XX	2s2p ⁴	2p ² 2p ² 4p	9.812	0.87	108 ⁺¹³² ₋₁₃₂
9.8739 ^{+0.0200} _{-0.0200}	0.10 ^{+49.90} _{-0.10}	0.93 ^{+1.08} _{-0.93}	2.3	Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.871	3.19	70 ⁺⁶⁰⁷ ₋₆₀₇
				Ni XXV	1s ² 2p ²	1s ² 2p3d	9.873	16.5	27 ⁺⁶⁰⁷ ₋₆₀₇
9.9066 ^{+0.0040} _{-0.0037}	2.75 ^{+21.58} _{-2.75}	2.48 ^{+1.23} _{-1.16}	12.7	(Ni XXV	1s ² 2p ²	1s ² 2p3d	9.924	9.00) -526 ⁺¹²² ₋₁₁₁
				(Ni XXV	1s ² 2p ²	1s ² 2p3d	9.938	8.84	-948 ⁺¹²² ₋₁₁₁
9.9361 ^{+0.0063} _{-0.0168}	22.66 ^{+27.34} _{-15.63}	2.36 ^{+1.78} _{-1.43}	7.4	(Ni XXV	1s ² 2p ²	1s ² 2p3d	9.907	1.86) 862 ⁺¹⁹² ₋₅₀₉
				(Ni XXIII	1s ² 2s ² 2p ²	1s ² 2s2p ² 3p	9.906	0.69	906 ⁺¹⁹² ₋₅₀₉
				(Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.911	0.57	737 ⁺¹⁹² ₋₅₀₉
				(Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p4	9.908	0.54	835 ⁺¹⁹² ₋₅₀₉
9.9724 ^{+0.0400} _{-0.0000}	21.56 ^{+18.44} _{-21.56}	-0.19 ^{+0.19} _{-1.48}	0.0	(Ni XXV	1s ² 2p ²	1s ² 2p3d	9.9240	9.00) 1461 ⁺¹²⁰⁸ ₋₀
				(Ni XXV	1s ² 2p ²	1s ² 2p3d	9.9380	8.84	1036 ⁺¹²⁰⁷ ₋₀

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion i	transition j	λ_0 [Å] $[10^{12}\text{s}^{-1}]$	A_{ji} [10^{12}s^{-1}]	$\Delta\lambda/\lambda \cdot c$ [km/s]		
9.9924 ^{+0.0041} _{-0.0030}	7.10 ^{+10.08} _{-7.10}	-2.76 ^{+1.05} _{-1.00}	20.4	(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 4d$	9.9977	6.56	←	-161 ⁺¹²⁴ ₋₈₉
				(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 4d$	10.0004	5.80		-241 ⁺¹²³ ₋₈₉
				Fe XX	$2s^2 2p^3$	$2s^2 2p^2 4d$	9.9935	0.81		-33 ⁺¹²⁴ ₋₈₉
				(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 4d$	10.0054	3.01		-391 ⁺¹²³ ₋₈₉
10.0335 ^{+0.0030} _{-0.0029}	21.88 ^{+6.75} _{-5.11}	-6.85 ^{+1.33} _{-1.43}	86.7							
10.1249 ^{+0.0054} _{-0.0200}	4.37 ^{+33.85} _{-4.37}	-1.77 ^{+0.99} _{-2.05}	8.9	Fe XX	$2s^2 2p^3$	$2s^2 2p^2 4d$	10.1203	2.12	←	137 ⁺¹⁶⁰ ₋₅₉₂
				(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 4d$	10.1322	0.39		-216 ⁺¹⁶⁰ ₋₅₉₂
10.2377 ^{+0.0023} _{-0.0004}	0.26 ^{+2.86} _{-0.26}	-4.80 ^{+0.68} _{-0.48}	108.5	Ne X	$1s$	$3p$	10.2385	1.65	←	-24 ⁺⁶⁸ ₋₁₁
				Ne X	$1s$	$3p$	10.2396	1.64		-56 ⁺⁶⁸ ₋₁₁
10.2799 ^{+0.0025} _{-0.0105}	0.23 ^{+40.24} _{-0.23}	2.86 ^{+1.20} _{-1.16}	19.7	Ni XXIV	$1s^2 2s^2 2p$	$1s^2 2s^2 3d$	10.2770	21.9		83 ⁺⁷⁴ ₋₃₀₇
				Fe XX	$2p^5$	$2p^2 2p^3 4d$	10.2690	5.14		295 ⁺⁷⁴ ₋₃₀₇
				(Fe XX	$2p^5$	$2s^2 2p^2 2p^2 5f$	10.2640	1.83		453 ⁺⁷⁴ ₋₃₀₇
				(Fe XX	$2p^5$	$2s^2 2p^2 2p^2 5f$	10.2640	1.44		453 ⁺⁷⁴ ₋₃₀₇
10.3025 ^{+0.0034} _{-0.0025}	0.08 ^{+11.95} _{-0.08}	2.55 ^{+1.47} _{-1.02}	18.0	(Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2p^2 2p^2 3p$	10.2990	3.26		102 ⁺⁹⁸ ₋₇₃
				Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^2 3p$	10.3010	1.01		33 ⁺⁹⁸ ₋₇₃
10.3201 ^{+0.0024} _{-0.0001}	0.01 ^{+5.83} _{-0.01}	2.83 ^{+1.22} _{-1.17}	18.0	(Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2p^3 3p$	10.3080	8.10		325 ⁺⁷¹ ₋₄
				(Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 5d$	10.3100	2.14		276 ⁺⁷¹ ₋₄
10.3375 ^{+0.0050} _{-0.0025}	0.00 ^{+12.39} _{-0.00}	1.94 ^{+1.05} _{-1.22}	8.1	Ni XXIV	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.3370	5.45		6 ⁺¹⁴⁵ ₋₇₂
				Fe XX	$2p^5$	$2p^4 4s$	10.3380	0.81		-23 ⁺¹⁴⁵ ₋₇₂
10.4825 ^{+0.0049} _{-0.0006}	0.01 ^{+15.33} _{-0.07}	3.49 ^{+0.87} _{-0.95}	28.6	Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^2 3p$	10.4840	5.77		-64 ⁺¹⁴⁰ ₋₁₉
				Ni XXII	$2s^2 2p^4$	$2p^2 2p^2 3p$	10.4830	3.43		-20 ⁺¹⁴⁰ ₋₁₉
10.6207 ^{+0.0010} _{-0.0013}	5.19 ^{+3.91} _{-5.18}	-6.65 ^{+0.87} _{-1.03}	164.0	(Fe XXIV	$1s^2 2s$	$1s^2 3p$	10.6190	7.19	←	47 ⁺³⁰ ₋₃₇
10.6389 ^{+0.0025} _{-0.0027}	5.01 ^{+11.16} _{-5.01}	-3.83 ^{+1.05} _{-1.47}	42.7							
10.6613 ^{+0.0037} _{-0.0035}	19.99 ^{+10.15} _{-8.78}	-5.63 ^{+1.51} _{-1.60}	51.4	Fe XXIV	$1s^2 2s$	$1s^2 3p$	10.6630	7.41	←	-47 ⁺¹⁰⁵ ₋₉₉
				(Fe XVII	$2s^2 2p^6$	$2s^2 2p^5 6d$	10.6570	1.15		122 ⁺¹⁰⁵ ₋₉₉
10.7071 ^{+0.0064} _{-0.0051}	21.87 ^{+13.61} _{-15.62}	4.76 ^{+1.95} _{-1.79}	20.7	Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.7070	17.6		-22 ⁺¹⁷⁹ ₋₁₄₄
				Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.7090	14.3		-54 ⁺¹⁷⁹ ₋₁₄₄
				Ni XXIV	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.7010	8.86		144 ⁺¹⁷⁹ ₋₁₄₄
				Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	10.7130	5.64		-171 ⁺¹⁷⁹ ₋₁₄₄
				(Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	10.7000	5.18		188 ⁺¹⁷⁹ ₋₁₄₄
10.7223 ^{+0.0027} _{-0.0023}	0.35 ^{+6.94} _{-0.35}	3.52 ^{+1.41} _{-1.34}	21.3	Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.7210	24.5		24 ⁺⁷⁶ ₋₆₃
				Ni XXII	$2s^2 2p^3$	$2s^2 2p^3 3p$	10.7200	4.58		41 ⁺⁷⁶ ₋₆₃
				Ni XXII	$2s^2 2p^4$	$2p^2 2p^3 3p$	10.7200	2.76		57 ⁺⁷⁶ ₋₆₃
10.7407 ^{+0.0143} _{-0.0032}	0.17 ^{+29.25} _{-0.17}	2.15 ^{+1.25} _{-1.21}	8.8	Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3p$	10.7430	34.9		-87 ⁺³⁹⁸ ₋₈₉
				Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	10.7520	29.9		-326 ⁺³⁹⁸ ₋₈₉
				(Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.7580	10.6		-483 ⁺³⁹⁷ ₋₈₉
				(Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	10.7360	8.86		117 ⁺³⁹⁸ ₋₈₉
10.7600 ^{+0.0050} _{-0.0025}	0.00 ^{+12.06} _{-0.00}	-0.87 ^{+0.57} _{-1.80}	4.2	(Ne IX	$1s^2$	$1s^5 p$	10.7650	0.52	←	-138 ⁺¹³⁸ ₋₇₁
10.8175 ^{+0.0075} _{-0.0025}	0.00 ^{+11.04} _{-0.00}	-1.59 ^{+0.90} _{-1.09}	7.5	Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	10.8197	18.3		-60 ⁺²⁰⁷ ₋₇₀
				Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	10.8215	11.0		-110 ⁺²⁰⁷ ₋₇₀
				Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.8156	9.49		53 ⁺²⁰⁷ ₋₇₀
				Fe XIX	$2s^2 2p^4$	$2s^2 2p^2 2p^2 4d$	10.8160	6.12		42 ⁺²⁰⁷ ₋₇₀
				Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.8160	5.65		42 ⁺²⁰⁷ ₋₇₀
10.8859 ^{+0.0037} _{-0.0072}	5.09 ^{+15.34} _{-5.09}	3.34 ^{+1.37} _{-1.31}	18.0	Ni XXIII	$1s^2 2p^4$	$1s^2 2p^3 3d$	10.8880	18.5		-76 ⁺¹⁰² ₋₁₉₉
				(Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3p$	10.8930	9.34		-201 ⁺¹⁰² ₋₁₉₉
				Ni XXIV	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 s$	10.8800	6.30		138 ⁺¹⁰² ₋₁₉₉
10.9150 ^{+0.0025} _{-0.0027}	0.05 ^{+19.52} _{-0.05}	3.66 ^{+1.58} _{-1.10}	24.7	(Ni XXIII	$1s^2 2p^4$	$1s^2 2p^2 2p^2 3d$	10.9200	11.7		-143 ⁺⁶⁸ ₋₇₄
				(Fe XIX	$2p^6$	$2p^2 2p^4 4d$	10.9230	8.25		-222 ⁺⁶⁸ ₋₇₄
10.9450 ^{+0.0000} _{-0.0050}	0.00 ^{+27.55} _{-0.00}	3.82 ^{+1.42} _{-1.32}	23.8	Ni XXIII	$1s^2 2p^4$	$1s^2 2p^2 2p^2 3d$	10.9400	28.8		130 ⁺⁰ ₋₁₃₇
				Ni XXIII	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p^3 d$	10.9430	8.04		49 ⁺⁰ ₋₁₃₇
10.9871 ^{+0.0019} _{-0.0021}	13.68 ^{+5.10} _{-4.50}	-7.05 ^{+1.30} _{-1.33}	104.5	(Fe XXIII	$1s^2 2s^2$	$1s^2 2s^3 p$	10.9810	7.56	←	167 ⁺⁵¹ ₋₅₈
				(Fe XXII	$1s^2 2s^2 2p$	$1s^2 2s^2 2p^3 p$	10.9935	1.35		-174 ⁺⁵¹ ₋₅₈
11.0071 ^{+0.0032} _{-0.0031}	17.86 ^{+8.15} _{-6.54}	-5.69 ^{+1.40} _{-1.49}	56.1	(Ne IX	$1s^2$	$1s^4 p$	11.0010	1.03	←	165 ⁺⁸⁶ ₋₈₅
11.0225 ^{+0.0033} _{-0.0025}	0.03 ^{+11.44} _{-0.03}	-2.24 ^{+0.83} _{-1.05}	16.6	(Fe XXIII	$1s^2 2s^2$	$1s^2 2s^3 p$	11.0190	4.68	←	96 ⁺⁹⁰ ₋₆₈
				(Fe XVII	$2s^2 2p^6$	$2s^2 2p^6 4p$	11.0260	1.75		-95 ⁺⁹⁰ ₋₆₈
11.0886 ^{+0.0070} _{-0.0085}	40.42 ^{+0.00} _{-14.98}	8.28 ^{+2.18} _{-2.17}	42.1	Ni XXIII	$1s^2 2p^4$	$1s^2 2p^2 2p^2 3d$	11.0890	13.3		-13 ⁺¹⁸⁹ ₋₂₂₉
				Ni XXIII	$1s^2 2p^4$	$1s^2 2p^2 2p^2 3d$	11.0950	12.9		-175 ⁺¹⁸⁹ ₋₂₂₈
				Ni XXIII	$1s^2 2s^2 2p^3$	$1s^2 2s^2 2p^2 3d$	11.0890	12.3		-20 ⁺¹⁸⁹ ₋₂₂₉
				Ni XXI	$2s^2 2p^5$	$2s^2 2p^2 2p^3 3d$	11.0950	10.8		-177 ⁺¹⁸⁹ ₋₂₂₈

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

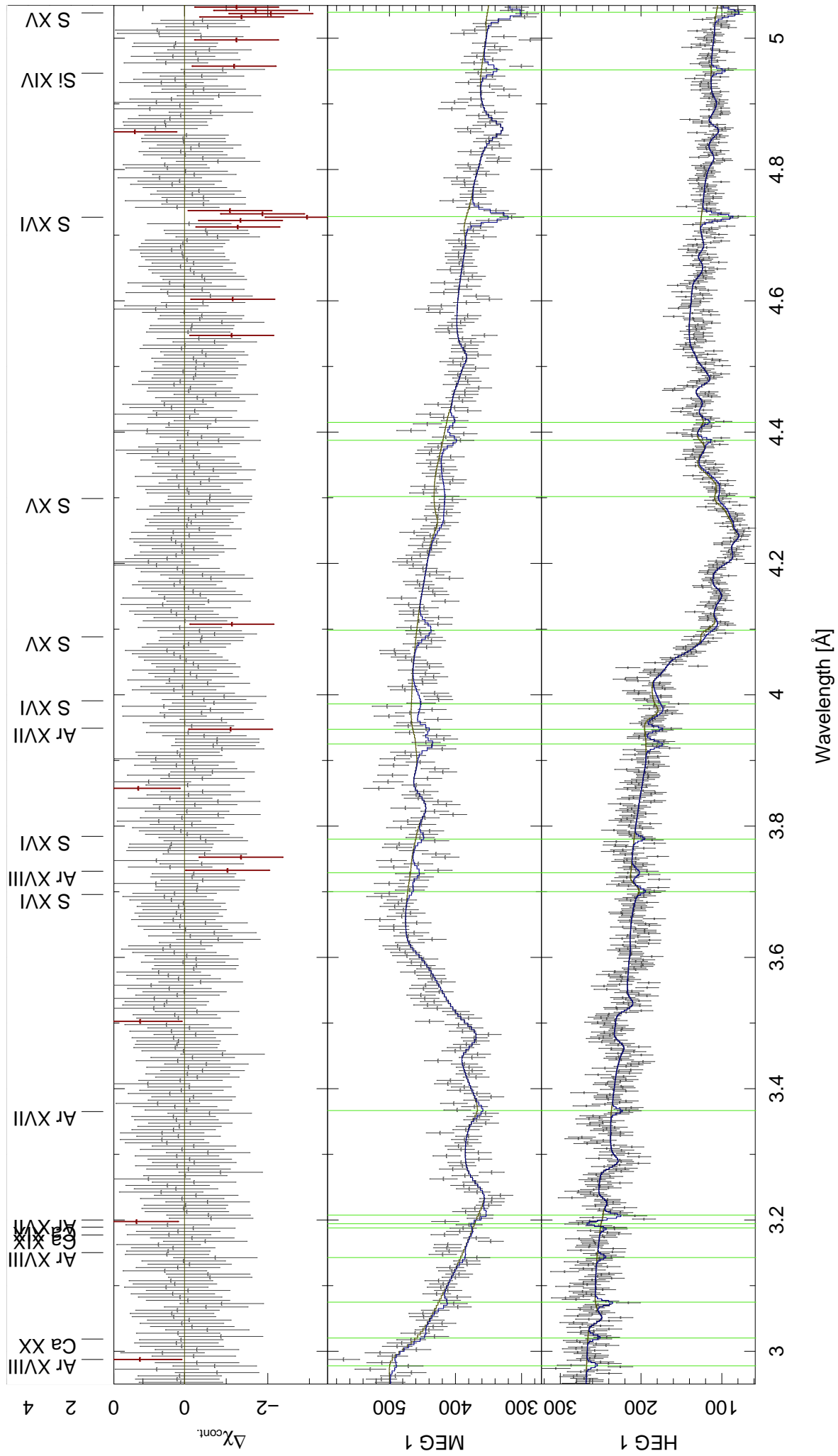
	λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition <i>i j</i>	λ_0 [Å] 10^{12}s^{-1}	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
11.1212 ^{+0.0188} _{-0.0212}	6.25 ^{+43.75} _{-6.25}	1.20 ^{+2.66} _{-1.20}	3.1	Ni XXIII	1s ² 2p ⁴	1s ² 2p ³ 3d	11.117	37.9		112 ⁺⁵⁰⁶ ₋₅₇₃	
				Ni XXIII	1s ² 2p ⁴	1s ² 2p ² 2p ² 3d	11.115	16.5	143 ⁺⁵⁰⁶ ₋₅₇₃		
				Ni XXIII	1s ² 2p ⁴	1s ² 2p ² 2p ² 3d	11.125	16.5	-106 ⁺⁵⁰⁶ ₋₅₇₂		
				Ni XXI	2s2p ⁵	2s2p2p ³ 3d	11.128	9.20	-191 ⁺⁵⁰⁵ ₋₅₇₂		
11.1516 ^{+0.0084} _{-0.0079}	24.59 ^{+25.41} _{-24.59}	4.26 ^{+2.24} _{-2.55}	15.8	Ni XXII	2s ² 2p ³	2s ² 2p ² 3d	11.146	5.26		126 ⁺²²⁷ ₋₂₁₃	
				Ni XXIII	1s ² 2s2p ³	1s ² 2s2p2p3	11.149	4.51	63 ⁺²²⁷ ₋₂₁₃		
				Ni XXIII	1s ² 2s2p ³	1s ² 2s2p2p3	11.147	4.46	117 ⁺²²⁷ ₋₂₁₃		
11.1790 ^{+0.0053} _{-0.0057}	22.21 ^{+18.06} _{-10.95}	5.78 ^{+2.22} _{-2.27}	27.2	Ni XXII	2s ² 2p ³	2s ² 2p2p3d	11.181	23.0		-75 ⁺¹⁴¹ ₋₁₅₃	
				Fe XXIV	1s ² 2p	1s ² 3d	11.176	21.5	80 ⁺¹⁴¹ ₋₁₅₃		
11.2140 ^{+0.0053} _{-0.0057}	21.20 ^{+18.30} _{-21.20}	5.55 ^{+2.31} _{-2.50}	26.4	Ni XXII	2s ² 2p ³	2s ² 2p2p3d	11.211	16.3		59 ⁺¹⁴¹ ₋₁₅₂	
				Ni XXII	2s ² 2p ³	2s ² 2p ² 3d	11.218	12.8	-110 ⁺¹⁴⁰ ₋₁₅₁		
				Ni XXII	2s2p ⁴	2s2p ³ 3d	11.210	11.0	108 ⁺¹⁴¹ ₋₁₅₂		
11.3153 ^{+0.0025} _{-0.0005}	0.08 ^{+9.91} _{-0.08}	-3.10 ^{+1.04} _{-1.02}	23.2	(Fe XVIII	2s ² 2p ⁵	2s ² 2p ⁴ 4d	11.3260	4.82	←	-283 ⁺⁶⁶ ₋₁₃	
				(Fe XVIII	2s ² 2p ⁵	2s ² 2p ⁴ 4d	11.3260	4.48	←	-283 ⁺⁶⁶ ₋₁₃	
				(Fe XVIII	2s ² 2p ⁵	2s ² 2p ⁴ 4d	11.3260	3.26	←	-283 ⁺⁶⁶ ₋₁₃	
11.3710 ^{+0.0065} _{-0.0025}	0.32 ^{+43.78} _{-0.32}	3.25 ^{+1.65} _{-1.41}	14.8	Ni XXI	2s2p ⁵	2s2p ² 2p ² 3d	11.375	15.5		-107 ⁺¹⁷² ₋₆₅	
				Ni XXI	2s ² 2p ⁴	2s ² 2p2p ² 3d	11.369	10.9	47 ⁺¹⁷² ₋₆₅		
				(Fe XXIII	1s ² 2s2p	1s ² 2s3d	11.366	9.35	131 ⁺¹⁷² ₋₆₅		
				Ni XXII	2s2p ⁴	2s2p2p ² 3d	11.372	8.88	-35 ⁺¹⁷² ₋₆₅		
11.4264 ^{+0.0036} _{-0.0029}	7.40 ^{+7.17} _{-7.40}	-3.96 ^{+1.15} _{-1.40}	29.9	Fe XXII	1s ² 2s ² 2p	1s ² 2s2p3p	11.4270	5.85	←	-17 ⁺⁷⁵ ₋₇₆	
11.4789 ^{+0.0059} _{-0.0025}	4.73 ^{+8.91} _{-4.73}	-3.16 ^{+1.47} _{-1.24}	15.4	(Fe XXII	1s ² 2s ² 2p	1s ² 2s2p3p	11.4900	6.40	←	-289 ⁺¹⁵⁵ ₋₆₆	
				(Fe XXII	1s ² 2s ² 2p	1s ² 2s2p3p	11.4900	1.68	←	-289 ⁺¹⁵⁵ ₋₆₆	
11.5312 ^{+0.0013} _{-0.0037}	0.00 ^{+10.14} _{-0.00}	-3.48 ^{+1.02} _{-1.01}	34.7	(Fe XVIII	2s ² 2p ⁵	2s ² 2p ² 2p ² 4d	11.5270	3.55	←	110 ⁺³³ ₋₉₇	
				(Fe XVIII	2s ² 2p ⁵	2s ² 2p ⁴ 4d	11.5270	4.22	←	110 ⁺³³ ₋₉₇	
11.5426 ^{+0.0024} _{-0.0001}	0.01 ^{+7.31} _{-0.01}	-3.47 ^{+1.05} _{-1.03}	28.4	Ne IX	1s ²	1s3p	11.5440	2.48	←	-37 ⁺⁶³ ₋₃	
11.5942 ^{+0.0049} _{-0.0051}	21.03 ^{+19.40} _{-14.99}	6.56 ^{+2.62} _{-2.42}	28.9	(Ni XXII	2s2p ⁴	2s2p2p ² 3d	11.599	7.34		-131 ⁺¹²⁸ ₋₁₃₂	
				Ni XXII	2s2p ⁴	2s2p ³ 3d	11.598	7.32	-119 ⁺¹²⁸ ₋₁₃₂		
				Fe XXIII	1s ² 2p ²	1s ² 2p3d	11.590	4.29	107 ⁺¹²⁸ ₋₁₃₂		
				Fe XXI	1s ² 2s2p ³	1s ² 2p ² 2p3p	11.596	3.75	-56 ⁺¹²⁸ ₋₁₃₂		
				Fe XXI	1s ² 2s2p ³	1s ² 2p2p ² 3p	11.594	3.33	-7 ⁺¹²⁸ ₋₁₃₂		
11.6256 ^{+0.0082} _{-0.0082}	30.39 ^{+10.04} _{-11.24}	6.30 ^{+2.98} _{-2.43}	20.9	(Fe XXIII	1s ² 2p ²	1s ² 2p3d	11.616	12.3		246 ⁺²¹³ ₋₂₁₃	
				(Ni XXII	2s2p ⁴	2s2p2p ² 3d	11.615	9.54	268 ⁺²¹³ ₋₂₁₃		
				Ni XXII	2s2p ⁴	2s2p ³ 3d	11.619	8.82	161 ⁺²¹³ ₋₂₁₃		
11.6609 ^{+0.0052} _{-0.0051}	28.21 ^{+12.21} _{-10.80}	8.85 ^{+2.83} _{-2.58}	39.7	Ni XXII	2s2p ⁴	2s2p ³ 3d	11.662	10.8		-40 ⁺¹³⁴ ₋₁₃₂	
				(Ni XXII	2s2p ⁴	2s2p2p ² 3d	11.652	2.52	229 ⁺¹³⁴ ₋₁₃₂		
11.6952 ^{+0.0078} _{-0.0090}	40.42 ^{+0.00} _{-9.78}	8.77 ^{+2.75} _{-2.36}	34.1	(Fe XXIII	1s ² 2p ²	1s ² 2p3d	11.684	6.69		275 ⁺²⁰⁰ ₋₂₃₂	
				Ni XXI	2s2p ⁵	2s2p2p ³ 3d	11.689	2.99	141 ⁺²⁰⁰ ₋₂₃₂		
11.7698 ^{+0.0012} _{-0.0014}	8.77 ^{+3.89} _{-4.92}	-10.01 ^{+0.98} _{-1.94}	248.9	Fe XXII	1s ² 2s ² 2p	1s ² 2s ² 3d	11.7700	16.3	←	-6 ⁺³¹ ₋₃₅	
				(Fe XX	2s ² 2p ³	2s2p ² 2p3p	11.7620	1.66	←	198 ⁺³¹ ₋₃₆	
11.8435 ^{+0.0040} _{-0.0035}	0.00 ^{+18.05} _{-0.00}	2.71 ^{+1.65} _{-1.60}	7.3	Ni XX	2s ² 2p ⁵	2s ² 2p2p ³ 3d	11.846	24.1		-64 ⁺¹⁰² ₋₈₈	
				Fe XXII	1s ² 2s2p ²	1s ² 2s2p3d	11.844	11.2	-16 ⁺¹⁰² ₋₈₈		
11.8887 ^{+0.0057} _{-0.0053}	29.52 ^{+10.90} _{-10.27}	8.81 ^{+2.78} _{-2.62}	37.3	(Fe XXII	1s ² 2s2p ²	1s ² 2s2p3d	11.881	12.5		193 ⁺¹⁴³ ₋₁₃₃	
				Fe XXI	1s ² 2s ² 2p ²	1s ² 2s2p2p3p	11.894	3.73	-142 ⁺¹⁴³ ₋₁₃₃		
11.9700 ^{+0.0050} _{-0.0000}	0.00 ^{+9.96} _{-0.00}	-3.06 ^{+1.24} _{-1.21}	16.1	(Fe XXI	1s ² 2s ² 2p ²	1s ² 2s2p ² 3p	11.9750	3.09	←	-125 ⁺¹²⁵ ₋₀	
12.0440 ^{+0.0012} _{-0.0040}	0.03 ^{+1.74} _{-0.03}	3.71 ^{+1.74} _{-1.70}	13.0	(Ca XX	2p	4d	12.048	2.70		-121 ⁺³⁰ ₋₁₀₀	
				(Ca XX	2s	4p	12.051	1.52	←	-188 ⁺³⁰ ₋₁₀₀	
12.0695 ^{+0.0060} _{-0.0059}	18.34 ^{+13.45} _{-9.07}	5.23 ^{+2.31} _{-2.12}	17.0	Fe XXII	1s ² 2s2p ²	1s ² 2s2p3d	12.075	9.55		-141 ⁺¹⁴⁹ ₋₁₄₆	
				(Fe XXI	1s ² 2s2p ³	1s ² 2s2p ² 3d	12.076	3.49	←	-171 ⁺¹⁴⁹ ₋₁₄₆	
				Fe XX	2s2p ⁴	2p2p ³ 3p	12.071	2.58	-37 ⁺¹⁴⁹ ₋₁₄₆		
12.1141 ^{+0.0014} _{-0.0024}	1.72 ^{+9.08} _{-1.72}	-4.09 ^{+1.08} _{-1.07}	45.2	(Fe XVII	2s ² 2p ⁶	2s ² 2p ⁵ 4d	12.1240	4.83	←	-244 ⁺³⁵ ₋₅₉	
				(Ne X	1s	2p	12.1321	6.16	←	-84 ⁺³⁰ ₋₂₃	
12.1287 ^{+0.0012} _{-0.0009}	19.38 ^{+3.03} _{-2.59}	-18.90 ^{+1.59} _{-1.85}	736.5	(Ne X	1s	2p	12.1375	6.16	←	-218 ⁺³⁰ ₋₂₃	
				(Ne X	1s	2p	12.1375	6.16	←	-218 ⁺³⁰ ₋₂₃	
12.1540 ^{+0.0045} _{-0.0048}	12.60 ^{+11.63} _{-12.60}	5.40 ^{+2.16} _{-2.12}	21.4	Ni XX	2s ² 2p ⁵	2s ² 2p ⁴ 3d	12.156	3.37		-64 ⁺¹¹² ₋₁₁₇	
				Fe XX	2s2p ⁴	2p2p ³ 3p	12.150	1.98	81 ⁺¹¹² ₋₁₁₇		
12.1845 ^{+0.0034} _{-0.0037}	2.15 ^{+11.41} _{-2.15}	3.15 ^{+1.72} _{-1.78}	9.9	(Fe XXI	1s ² 2s2p ³	1s ² 2s2p ² 3d	12.191	6.59		-176 ⁺⁸³ ₋₉₂	
				(Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p3p	12.193	3.99	←	-228 ⁺⁸³ ₋₉₂	
12.2144 ^{+0.0058} _{-0.0061}	15.87 ^{+16.52} _{-10.32}	5.07 ^{+2.41} _{-2.25}	16.2	(Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p3p	12.204	8.54		256 ⁺¹⁴³ ₋₁₄₉	
				Fe XXII	1s ² 2s2p ³	1s ² 2s2p3d	12.210	8.24	109 ⁺¹⁴³ ₋₁₄₉		
				Fe XXI	1s ² 2s2p ³	1s ² 2s2p2p3p	12.209	6.40	131 ⁺¹⁴³ ₋₁₄₉		
12.2523 ^{+0.0027} _{-0.0023}	0.07 ^{+13.90} _{-0.07}	-4.27 ^{+1.25} _{-1.21}	26.3	Fe XXII	1s ² 2s ² 2p	1s ² 2s ² 3s	12.2519	0.91	←	11 ⁺⁶⁵ ₋₅₇	
12.2650 ^{+0.0025} _{-0.0011}	0.32 ^{+7.58} _{-0.32}	-5.38 ^{+1.04} _{-1.23}	61.8	Fe XVII	2s ² 2p ⁶	2s ² 2p ⁵ 4d	12.2660	4.21	←	-24 ⁺⁶⁰ ₋₂₆	
12.2821 ^{+0.0012} _{-0.0008}	12.49 ^{+2.02} _{-4.46}	-14.47 ^{+1.39} _{-0.98}	467.7	(Fe XXI	1s ² 2s ² 2p ²	1s ² 2s ² 2p3d	12.2840	18.2	←	-46 ⁺³⁰ ₋₂₀	

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

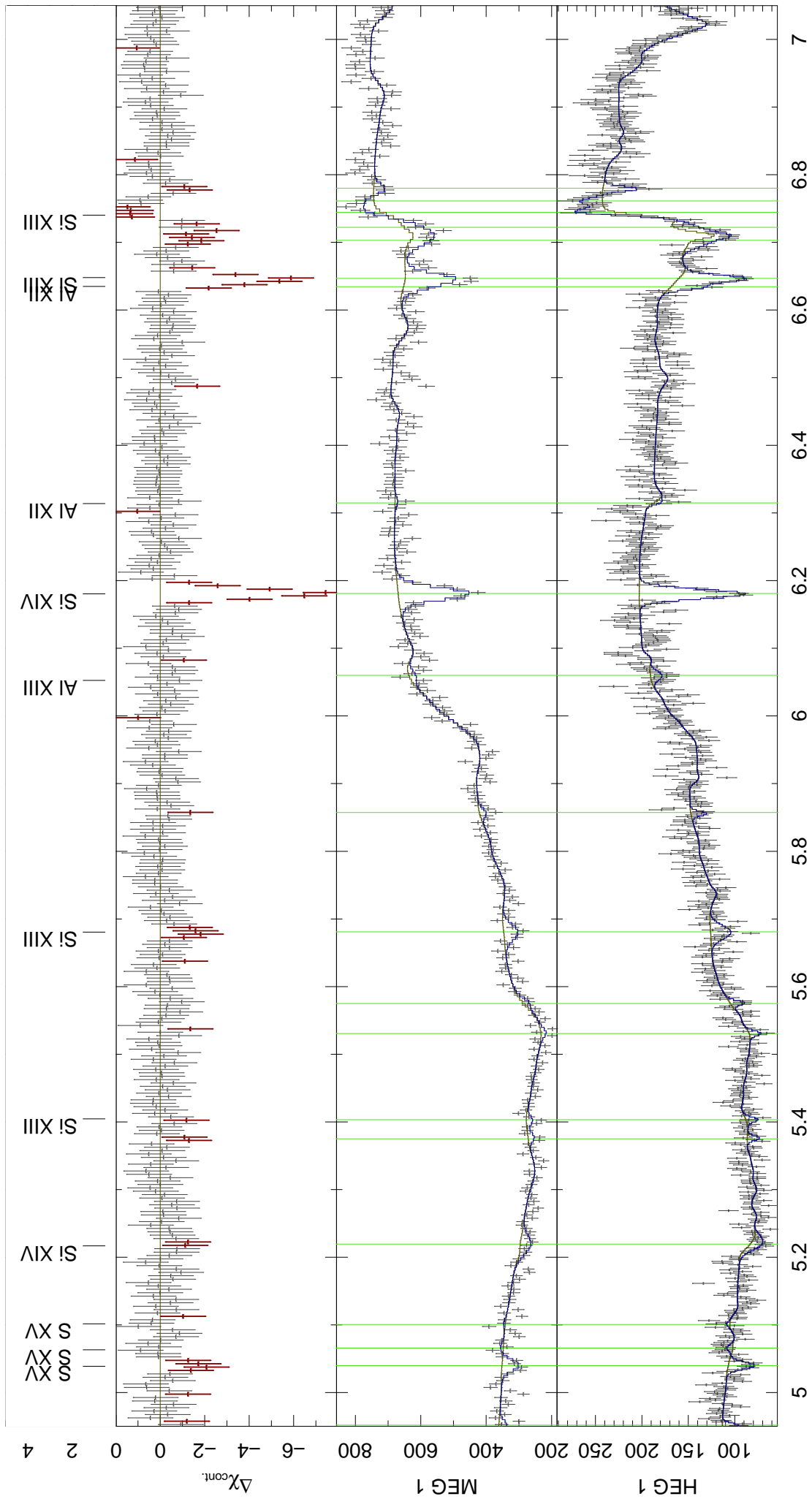
λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition <i>i j</i>	λ_0 [Å][10^{12}s^{-1}]	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
12.3150 ^{+0.0050} _{-0.0000}	0.00 ^{+10.08} _{-0.00}	-2.24 ^{+1.41} _{-1.38}	7.2	Fe XXI (Fe XX)	$1s^2 2s 2p^3$ $2s^2 2p^3$	$1s^2 2p 2p^2 3p$ $2s 2p 2p^2 3p$	12.3182 12.3244	3.37 2.34		-78 ⁺¹²² ₋₀ -229 ⁺¹²² ₋₀
12.4757 ^{+0.0081} _{-0.0091}	25.87 ^{+14.56} _{-10.57}	6.15 ^{+2.90} _{-2.68}	16.0	(Fe XXI Fe XXI Fe XXI Fe XXI)	$1s^2 2s 2p^3$ $1s^2 2p^4$ $1s^2 2s 2p^3$ $1s^2 2s 2p^3$	$1s^2 2s 2p^2 3d$ $1s^2 2p^3 3d$ $1s^2 2s 2p^2 3d$ $1s^2 2p 2p^2 3p$	12.4650 12.463 12.472 12.467	26.9 14.6 9.00 5.82		242 ⁺¹⁹⁴ ₋₂₁₉ 302 ⁺¹⁹⁴ ₋₂₁₉ 74 ⁺¹⁹⁴ ₋₂₁₈ 196 ⁺¹⁹⁴ ₋₂₁₉
12.5661 ^{+0.0044} _{-0.0046}	3.97 ^{+19.79} _{-3.97}	-3.56 ^{+1.46} _{-2.27}	15.9	(Fe XX Fe XX)	$2s^2 2p^3$ $2s^2 2p^3$	$2s 2p 2p^2 3p$ $2s 2p^3 3p$	12.5760 12.5760	4.39 4.44		-237 ⁺¹⁰⁴ ₋₁₁₁ -237 ⁺¹⁰⁴ ₋₁₁₁
12.5810 ^{+0.0033} _{-0.0042}	11.62 ^{+12.40} _{-11.62}	-5.68 ^{+2.07} _{-2.15}	31.6	(Fe XX Fe XX)	$2s^2 2p^3$ $2s^2 2p^3$	$2s 2p 2p^2 3p$ $2s 2p^3 3p$	12.5760 12.5760	4.39 4.44	←	120 ⁺⁷⁹ ₋₉₉ 120 ⁺⁷⁹ ₋₉₉
12.6233 ^{+0.0045} _{-0.0033}	0.04 ^{+17.14} _{-0.04}	3.44 ^{+1.98} _{-1.94}	8.3	(Ca XVIII Ca XVIII)	$1s^2 2s$ $1s^2 2s$	$1s^2 5p$ $1s^2 5p$	12.6360 12.6360	0.54 0.54		-301 ⁺¹⁰⁶ ₋₇₈ -301 ⁺¹⁰⁶ ₋₇₈
12.6834 ^{+0.0124} _{-0.0133}	43.24 ^{+0.04} _{-43.24}	7.78 ^{+3.35} _{-3.32}	14.9	Fe XXI Fe XXI Fe XX	$1s^2 2p^4$ $1s^2 2p^4$ $2s 2p^4$	$1s^2 2p 2p^2 3d$ $1s^2 2p 2p^2 3d$ $2s 2p^3 3d$	12.6730 12.6890 12.688	3.97 2.00 1.34		245 ⁺²⁹⁴ ₋₃₁₄ -147 ⁺²⁹⁴ ₋₃₁₄ -121 ⁺²⁹⁴ ₋₃₁₄
12.8099 ^{+0.0001} _{-0.0050}	0.02 ^{+14.55} _{-0.02}	-5.44 ^{+1.28} _{-2.67}	58.1	(Ni XX Fe XX)	$2s^2 2p^5$ $2s 2p^4$	$2s^2 2p^4 3s$ $2s 2p^2 2p 3d$	12.8122 12.8084	1.10 0.83		-55 ⁺² ₋₁₁₆ 35 ⁺² ₋₁₁₆
12.8281 ^{+0.0012} _{-0.0016}	18.73 ^{+4.33} _{-3.22}	-17.49 ^{+2.14} _{-2.24}	393.4	(Fe XX Fe XX)	$2s^2 2p^3$ $2s^2 2p^3$	$2s^2 2p 2p 3d$ $2s^2 2p 2p 3d$	12.8240 12.8270	17.1 4.90	←	95 ⁺²⁹ ₋₃₇ 25 ⁺²⁹ ₋₃₇
12.8500 ^{+0.0016} _{-0.0021}	0.36 ^{+13.32} _{-0.35}	-7.08 ^{+1.38} _{-1.24}	79.7	(Fe XX Fe XX)	$2s^2 2p^3$ $2s^2 2p^3$	$2s^2 2p 2p 3d$ $2s^2 2p 2p 3d$	12.8460 12.8640	19.2 12.1	←	92 ⁺³⁸ ₋₄₈ -327 ⁺³⁸ ₋₄₈
12.9023 ^{+0.0050} _{-0.0033}	17.65 ^{+17.25} _{-9.52}	-8.60 ^{+2.17} _{-3.70}	55.7	Fe XX Fe XX	$2s 2p^4$ $2s 2p^4$	$2s 2p 2p^2 3d$ $2s 2p 2p^2 3d$	12.9010 12.9030	11.4 7.58		31 ⁺¹¹⁷ ₋₇₇ -15 ⁺¹¹⁷ ₋₇₇
12.9200 ^{+0.0025} _{-0.0000}	0.01 ^{+21.12} _{-0.01}	-4.74 ^{+1.42} _{-1.43}	30.6	(Fe XX Fe XX)	$2s^2 2p^3$ $2s^2 2p^3$	$2s^2 2p 2p 3d$ $2s^2 2p 2p 3d$	12.9120 12.9211	4.92 0.74	←	186 ⁺⁵⁹ ₋₁ -25 ⁺⁵⁹ ₋₁
12.9538 ^{+0.0025} _{-0.0026}	14.50 ^{+8.23} _{-5.82}	-9.91 ^{+2.29} _{-2.52}	80.3	(Fe XIX Fe XX Fe XX)	$2s^2 2p^4$ $2s^2 2p^3$ $2s^2 2p^3$	$2s 2p 2p^3 3p$ $2s^2 2p 2p 3d$ $2s^2 2p^2 3d$	12.9450 12.9650 12.9654	3.11 3.46 0.66	←	204 ⁺⁵⁸ ₋₆₁ -259 ⁺⁵⁸ ₋₆₁ -268 ⁺⁵⁸ ₋₆₁
13.0287 ^{+0.0034} _{-0.0043}	14.32 ^{+11.36} _{-14.32}	-6.46 ^{+2.41} _{-2.05}	27.9	(Fe XX Fe XX Fe XX)	$2p^5$ $2s 2p^4$ $2s 2p^4$	$2p^4 3d$ $2s 2p 2p^2 3d$ $2s 2p^3 3d$	13.0238 13.0328 13.0281	14.1 6.60 3.49		113 ⁺⁷⁸ ₋₉₈ -94 ⁺⁷⁸ ₋₉₈ 14 ⁺⁷⁸ ₋₉₈
13.0522 ^{+0.0051} _{-0.0031}	11.16 ^{+13.73} _{-11.16}	-5.76 ^{+1.90} _{-2.52}	25.0	(Fe XX)	$2s^2 2p^3$	$2s^2 2p^2 3d$	13.0610	2.62	←	-203 ⁺¹¹⁶ ₋₇₁
13.0725 ^{+0.0299} _{-0.0299}	0.00 ^{+50.42} _{-0.00}	1.77 ^{+2.22} _{-1.77}	2.2	(Fe XX Fe XX)	$2s 2p^4$ $2s^2 2p^3$	$2s 2p 2p^2 3d$ $2s^2 2p 2p 3d$	12.920 12.921	0.78 0.74		3529 ⁺⁶⁹³ ₋₆₉₃ 3513 ⁺⁶⁹³ ₋₆₉₃
13.1152 ^{+0.0211} _{-0.0211}	34.47 ^{+15.95} _{-34.47}	4.51 ^{+3.75} _{-3.94}	8.6	(Fe XX Fe XX Fe XX Fe XX Fe XX Fe XX Fe XX Fe XX Fe XX)	$2s 2p^4$ $2s^2 2p^3$ $2s^2 2p^3$ $2s 2p^4$ $2s^2 2p^3$ $2s^2 2p^3$ $2p^5$ $2s 2p^4$ $2p^5$ $2s^2 2p^3$	$2s 2p 2p^2 3d$ $2s^2 2p^2 3d$ $2s^2 2p^2 3d$ $2s 2p^3 3d$ $2s^2 2p^2 3d$ $2s^2 2p^2 3d$ $2p 2p^3 3d$ $2s 2p^3 3d$ $2p 2p^3 3d$ $2s^2 2p 2p 3d$	13.084 13.087 13.088 13.095 13.093 13.100 13.114 13.124 13.129 13.137	24.9 15.1 13.9 12.4 8.09 6.76 5.35 4.38 3.08 2.80		711 ⁺⁴⁸⁴ ₋₄₈₄ 642 ⁺⁴⁸⁴ ₋₄₈₄ 615 ⁺⁴⁸⁴ ₋₄₈₄ 454 ⁺⁴⁸⁴ ₋₄₈₄ 505 ⁺⁴⁸⁴ ₋₄₈₄ 349 ⁺⁴⁸⁴ ₋₄₈₄ 26 ⁺⁴⁸³ ₋₄₈₃ -209 ⁺⁴⁸³ ₋₄₈₃ -333 ⁺⁴⁸³ ₋₄₈₃ -497 ⁺⁴⁸² ₋₄₈₂
13.1383 ^{+0.0044} _{-0.0033}	0.02 ^{+17.29} _{-0.02}	-3.19 ^{+1.95} _{-2.03}	-1.1	Fe XX Fe XVIII (Fe XX)	$2s^2 2p^3$ $2s 2p^6$ $2s 2p^4$	$2s^2 2p 2p 3d$ $2p^2 2p^4 3p$ $2s 2p 2p^2 3d$	13.1370 13.1427 13.1458	2.80 1.95 1.82		29 ⁺¹⁰¹ ₋₇₅ -101 ⁺¹⁰¹ ₋₇₄ -172 ⁺¹⁰¹ ₋₇₄
13.1500 ^{+0.0000} _{-0.0050}	0.00 ^{+11.56} _{-0.00}	7.94 ^{+2.32} _{-2.27}	28.1	(Fe XVIII Fe XX)	$2s 2p^6$ $2s 2p^4$	$2p^2 2p^4 3p$ $2s 2p 2p^2 3d$	13.142 13.145	1.95 1.82		167 ⁺⁰ ₋₁₁₄ 96 ⁺⁰ ₋₁₁₄
13.1550 ^{+0.0001} _{-0.0050}	0.00 ^{+12.15} _{-0.00}	-5.43 ^{+1.91} _{-2.13}	31.6	Fe XX Fe XX Fe XX Fe XX	$2s 2p^4$ $2s 2p^4$ $2s^2 2p^3$ $2s^2 2p^3$	$2s 2p^3 3d$ $2s 2p 2p^2 3d$ $2s^2 2p 2p 3d$ $2s^2 2p 2p 3d$	13.1521 13.1547 13.1530 13.1530	6.12 3.95 3.82 2.12		65 ⁺¹ ₋₁₁₃ 6 ⁺¹ ₋₁₁₃ 45 ⁺¹ ₋₁₁₃ 45 ⁺¹ ₋₁₁₃
13.4202 ^{+0.0043} _{-0.0118}	9.72 ^{+16.28} _{-9.72}	-4.08 ^{+1.57} _{-2.96}	25.2	Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 3d$	13.4230	5.01	←	-61 ⁺⁹⁷ ₋₂₆₃
13.4403 ^{+0.0020} _{-0.0021}	19.29 ^{+4.08} _{-4.44}	-18.54 ^{+2.47} _{-2.28}	301.1	(Ne IX)	$1s^2$	$1s 2p$	13.4473	8.87	←	-155 ⁺⁴⁵ ₋₄₆
13.4647 ^{+0.0030} _{-0.0035}	20.44 ^{+8.49} _{-9.09}	-11.81 ^{+3.60} _{-2.00}	68.4	Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 3d$	13.4620	14.1	←	61 ⁺⁶⁶ ₋₇₉
13.5026 ^{+0.0019} _{-0.0016}	17.55 ^{+4.32} _{-4.83}	-17.72 ^{+2.57} _{-2.18}	241.5	(Fe XIX)	$2s^2 2p^4$	$2s^2 2p 2p^2 3d$	13.4970	12.9	←	125 ⁺⁴¹ ₋₃₆
13.5251 ^{+0.0014} _{-0.0030}	10.44 ^{+7.45} _{-6.78}	-10.90 ^{+2.47} _{-1.82}	88.5	(Fe XIX Fe XIX)	$2s^2 2p^4$ $2s^2 2p^4$	$2s^2 2p^3 3d$ $2s^2 2p 2p^2 3d$	13.5180 13.5146	18.7 1.21	←	156 ⁺³¹ ₋₆₇ 232 ⁺³¹ ₋₆₇
13.6279 ^{+0.0055} _{-0.0062}	5.84 ^{+22.03} _{-5.84}	-5.06 ^{+2.45} _{-3.41}	14.5							
13.6501 ^{+0.0061} _{-0.0055}	20.29 ^{+19.29} _{-10.10}	-8.11 ^{+3.14} _{-3.48}	22.3	Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 3d$	13.6450	2.43	←	112 ⁺¹³⁵ ₋₁₂₂
13.6657 ^{+0.0069} _{-0.0057}	0.23 ^{+47.54} _{-0.23}	-3.38 ^{+2.04} _{-2.72}	7.3							

Table A III.1: List of lines in the ‘non-dip’ spectrum – sorted by wavelength (continued)

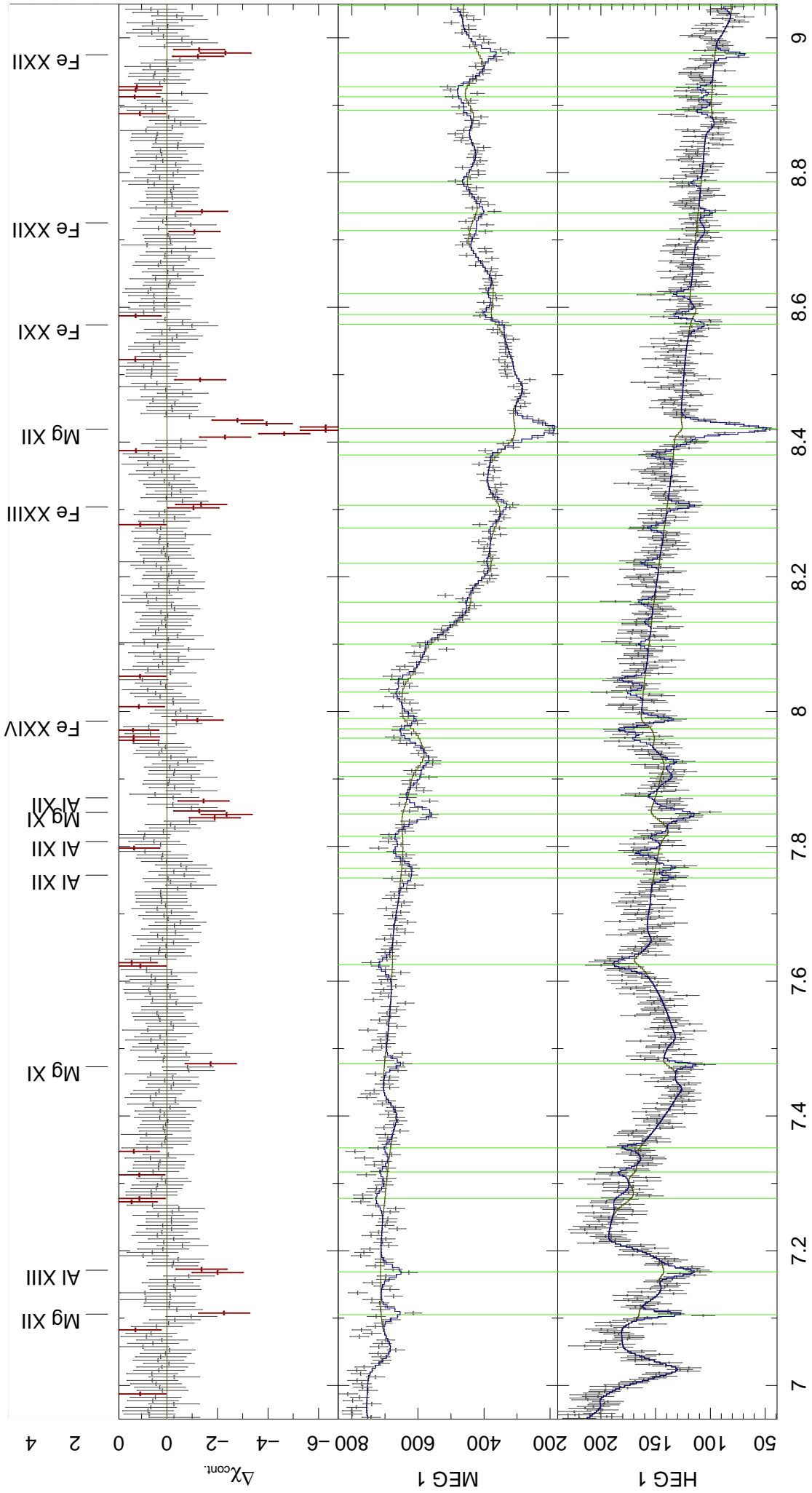
λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å][10^{12}s^{-1}]	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
13.7551 ^{+0.0099} _{-0.0049}	0.02 ^{+47.63} _{-0.02}	-4.74 ^{+2.67} _{-2.62}	8.4	(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 3s$	13.7670	1.02) ← -259 ⁺²¹⁵ ₋₁₀₈
13.7914 ^{+0.0116} _{-0.0058}	26.00 ^{+24.00} _{-19.82}	-9.25 ^{+4.05} _{-4.96}	17.8	Fe XIX	$2s^2 2p^4$	$2s^2 2p 2p^2 3d$	13.7950	5.35	← -78 ⁺²⁵² ₋₁₂₆
13.8318 ^{+0.0088} _{-0.0085}	28.25 ^{+17.58} _{-20.33}	-9.17 ^{+5.40} _{-3.95}	15.6	(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 3s$	13.8390	1.75	← -155 ⁺¹⁹⁰ ₋₁₈₅
				(Fe XX	$2s^2 2p^3$	$2s^2 2p^2 3s$	13.8430	1.00) -242 ⁺¹⁹⁰ ₋₁₈₅
				Fe XVII	$2s^2 2p^6$	$2s 2p^6 3p$	13.8250	3.40	148 ⁺¹⁹⁰ ₋₁₈₅
14.2010 ^{+0.0029} _{-0.0034}	26.41 ^{+8.47} _{-7.78}	-21.96 ^{+4.77} _{-4.27}	142.4	(Fe XVIII	$2s^2 2p^5$	$2s^2 2p 2p^3 3d$	14.2080	17.9) ← -148 ⁺⁶² ₋₇₁
				(Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3d$	14.2080	19.4) -148 ⁺⁶² ₋₇₁
				Fe XVIII	$2s 2p^6$	$2s 2p^2 2p^3 3d$	14.2007	13.5	6 ⁺⁶² ₋₇₁
14.2525 ^{+0.0043} _{-0.0031}	10.00 ^{+13.77} _{-10.00}	-11.44 ^{+2.84} _{-4.96}	41.6	Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3d$	14.2560	12.9	← -74 ⁺⁹¹ ₋₆₅
				Fe XVIII	$2s^2 2p^5$	$2s^2 2p 2p^3 3d$	14.2560	1.29	-74 ⁺⁹¹ ₋₆₅
14.3700 ^{+0.0100} _{-0.0050}	0.00 ^{+26.25} _{-0.00}	-4.49 ^{+2.94} _{-3.95}	6.3	Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3d$	14.3730	6.75	← -62 ⁺²⁰⁹ ₋₁₀₄
14.5301 ^{+0.0099} _{-0.0001}	0.00 ^{+75.00} _{-0.00}	-5.97 ^{+3.07} _{-3.02}	10.1	Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3d$	14.5340	4.05	← -81 ⁺²⁰⁴ ₋₂₀
14.6240 ^{+0.0048} _{-0.0035}	5.02 ^{+18.37} _{-5.02}	-8.80 ^{+2.78} _{-3.38}	27.1	(Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3d$	14.6160	0.95) 164 ⁺⁹⁸ ₋₇₂
				(Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 3s$	14.6359	0.11) -244 ⁺⁹⁸ ₋₇₂
14.7150 ^{+0.0035} _{-0.0043}	0.85 ^{+19.30} _{-0.85}	10.14 ^{+4.76} _{-4.15}	15.2	(Fe XVIII	$2s 2p^6$	$2s 2p 2p^3 3d$	14.7260	11.8) -224 ⁺⁷¹ ₋₈₈
				(Fe XIX	$2s 2p^5$	$2s 2p 2p^3 3s$	14.7200	1.30) -114 ⁺⁷¹ ₋₈₈
				Fe XX	$2s 2p^4$	$2s^2 2p 2p 3p$	14.7110	0.015	73 ⁺⁷¹ ₋₈₈
15.0033 ^{+0.0058} _{-0.0026}	13.76 ^{+9.96} _{-13.75}	-13.12 ^{+3.16} _{-4.82}	42.4	(Fe XVII	$2s^2 2p^6$	$2s^2 2p^5 3d$	15.0140	27.0) ← -214 ⁺¹¹⁶ ₋₅₁
15.1721 ^{+0.0082} _{-0.0072}	0.03 ^{+40.38} _{-0.03}	-4.71 ^{+3.24} _{-5.68}	5.7	O VIII	$1s$	$4p$	15.1760	0.27	← -77 ⁺¹⁶¹ ₋₁₄₂
				O VIII	$1s$	$4p$	15.1765	0.27	-86 ⁺¹⁶¹ ₋₁₄₂
15.2551 ^{+0.0259} _{-0.0259}	0.00 ^{+26.76} _{-0.00}	-3.28 ^{+3.28} _{-2.97}	2.5	Fe XVII	$2s^2 2p^6$	$2s^2 2p^5 3d$	15.2610	5.87	← -117 ⁺⁵⁰⁹ ₋₅₀₉
15.6200 ^{+0.0200} _{-0.0051}	0.31 ^{+74.69} _{-0.31}	-7.54 ^{+3.31} _{-3.86}	12.9	Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3s$	15.6250	0.87	-96 ⁺³⁸⁴ ₋₉₈
16.0002 ^{+0.0051} _{-0.0043}	25.00 ^{+14.56} _{-12.27}	-20.38 ^{+5.40} _{-6.26}	63.1	(O VIII	$1s$	$3p$	16.0055	0.67) ← -99 ⁺⁹⁶ ₋₈₀
				(O VIII	$1s$	$3p$	16.0067	0.67) -120 ⁺⁹⁶ ₋₈₀
				Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 3s$	16.0040	1.36	-70 ⁺⁹⁶ ₋₈₀
16.2411 ^{+0.0100} _{-0.0050}	15.00 ^{+15.00} _{-15.00}	-7.95 ^{+5.09} _{-7.71}	7.3	(Ca XX	$2p$	$3d$	16.2294	8.48) 217 ⁺¹⁸⁵ ₋₉₂
				(Ca XX	$2s$	$3p$	16.2343	3.53) 127 ⁺¹⁸⁵ ₋₉₂
16.3893 ^{+0.0002} _{-0.0198}	75.00 ^{+0.00} _{-19.45}	-24.67 ^{+8.07} _{-10.59}	21.2	Ca XX	$2p$	$3d$	16.3716	10.2	324 ⁺³ ₋₃₆₃
				Ca XX	$2p$	$3d$	16.3872	1.70	39 ⁺³ ₋₃₆₂
16.7700 ^{+0.0100} _{-0.0049}	0.00 ^{+23.84} _{-0.00}	-7.34 ^{+5.71} _{-4.33}	4.8	Fe XVII	$2s^2 2p^6$	$2s^2 2p^5 3s$	16.7800	0.90	← -179 ⁺¹⁷⁹ ₋₈₈
17.7495 ^{+0.0200} _{-0.0200}	16.42 ^{+58.58} _{-16.42}	-5.83 ^{+5.83} _{-11.86}	1.5	Ar XVI	$1s^2 2s$	$1s^2 4p$	17.7320	0.65	← 297 ⁺³³⁸ ₋₃₃₈
				Ar XVI	$1s^2 2s$	$1s^2 4p$	17.7420	0.65	127 ⁺³³⁸ ₋₃₃₈
18.6277 ^{+0.0162} _{-0.0177}	50.00 ^{+0.00} _{-14.23}	-33.47 ^{+14.91} _{-11.47}	16.0	O VII	$1s^2$	$1s 3p$	18.6270	0.93	← 12 ⁺²⁶¹ ₋₂₈₅
18.7387 ^{+0.0074} _{-0.0100}	0.05 ^{+74.95} _{-0.05}	-14.83 ^{+7.61} _{-7.43}	10.3	Ca XVIII	$1s^2 2s$	$1s^2 3p$	18.7320	2.36	← 107 ⁺¹¹⁸ ₋₁₆₀
18.9394 ^{+0.0306} _{-0.0094}	55.92 ^{+90.91} _{-22.72}	-55.11 ^{+18.40} _{-116.40}	38.1	O VIII	$1s$	$2p$	18.9671	2.52	← -438 ⁺⁴⁸³ ₋₁₄₉
				(O VIII	$1s$	$2p$	18.9725	2.52) -523 ⁺⁴⁸³ ₋₁₄₉



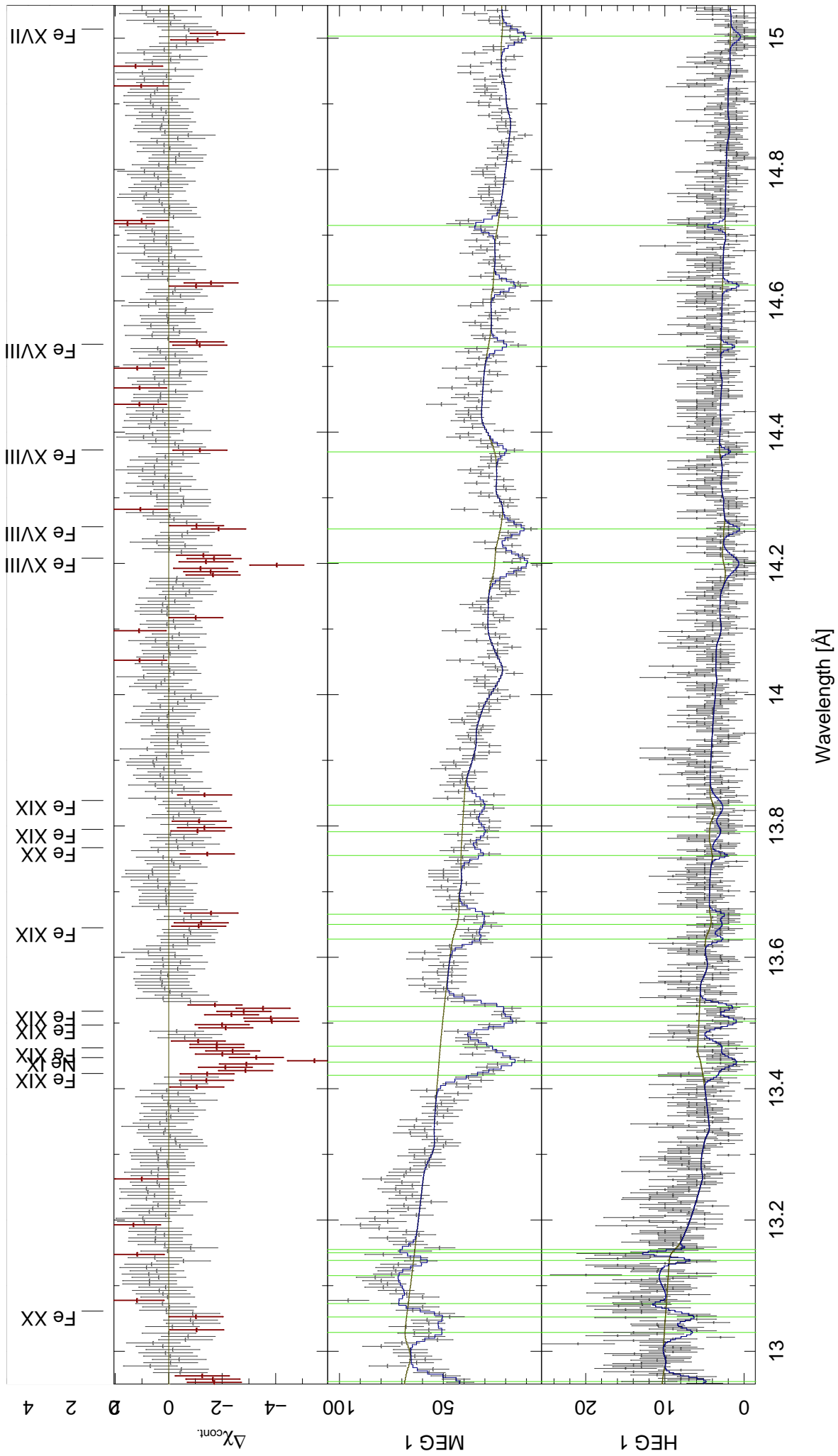
(b) 3...5 Å range of the 'non-dip' Cyg X-1 spectrum



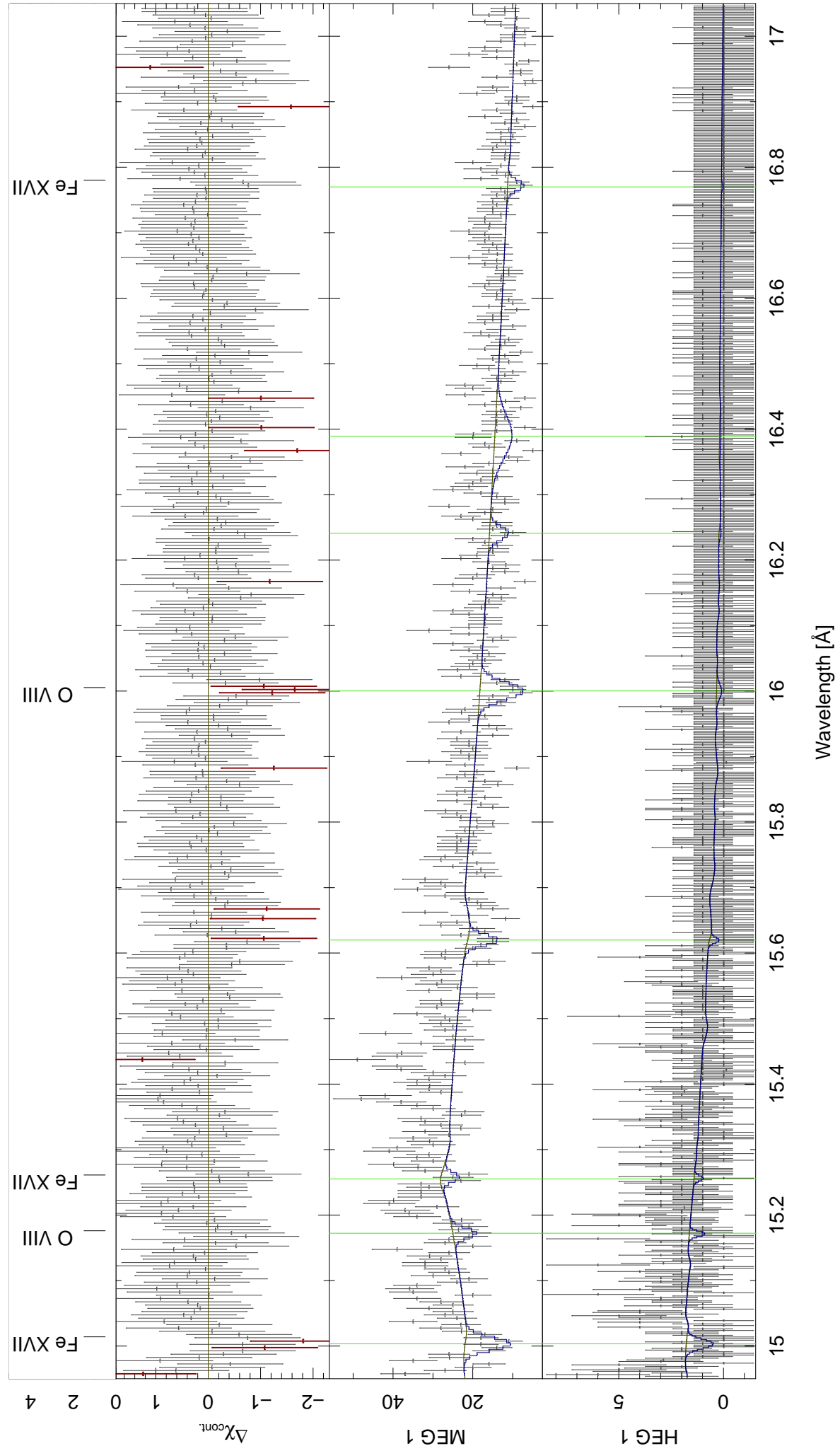
(c) 5...7 Å range of the 'non-dip' Cyg X-1 spectrum



(d) 7...9 Å range of the 'non-dip' Cyg X-1 spectrum



(g) 13...15 Å range of the 'non-dip' Cyg X-1 spectrum



(h) 15...17 \AA range of the 'non-dip' Cyg X-1 spectrum

Figure A III.6: The fitted 'non-dip' Cyg X-1 spectrum, including all lines.

III.3 Dependencies of the continuum-parameters

In Section 4.3.2 (page 88), the dependence of the best fit parameters on **A)** a (fixed) given value of Γ was investigated, as well as the dependence on **B)** the wavelength range used for fitting, in order to justify the introduction of a new model for the continuum, namely a two component (partial covered) photoabsorbed power law with pile-up reduction (model 4.6) for the ‘dip’ spectrum.

The ‘non-dip’ spectrum: A) Γ -dependence

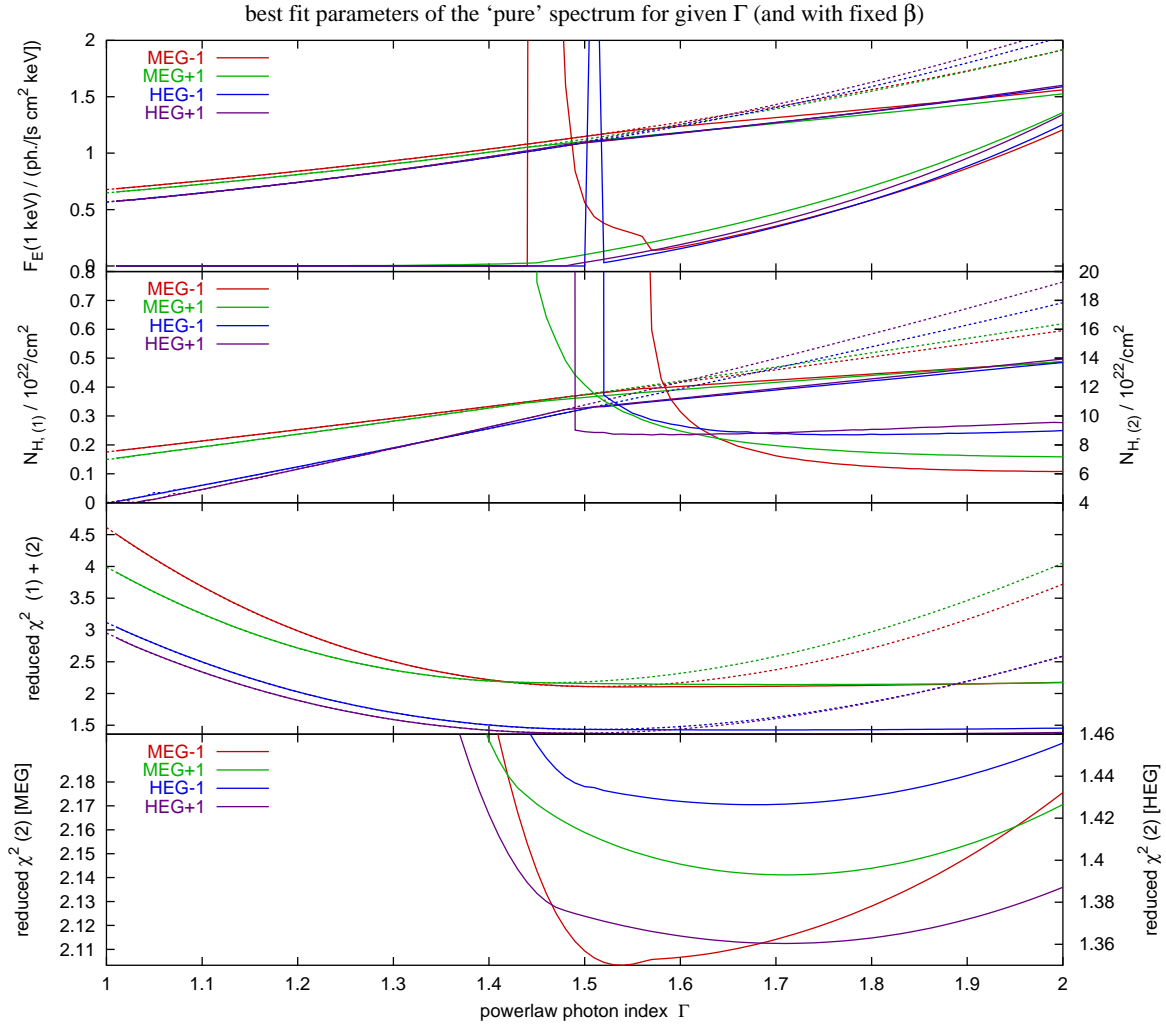


Figure A III.7: The ‘non-dip’ spectrum: Γ -dependence for both models’ fit parameters (rebinned to ≥ 30 counts/bin; [1 Å, 20 Å] range noticed).

The 'non-dip' spectrum: B) λ_2 -dependence (with pile-up scales thawed)

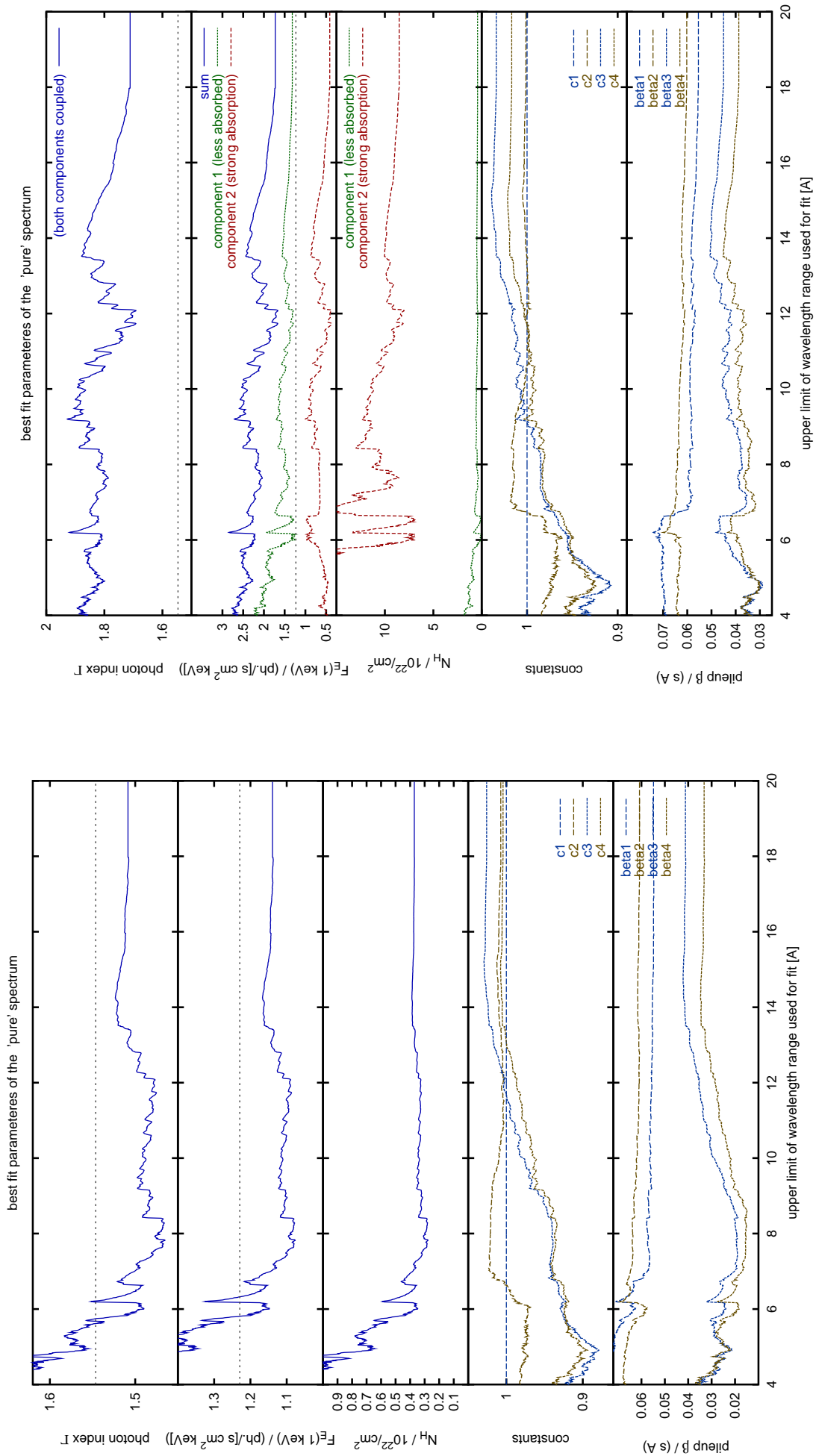


Figure A III.8: λ_2 -dependence of the 1-comp. model's fit parameters.

For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied.

Figure A III.9: λ_2 -dependence of the 2-comp. model's fit parameters.

The ‘dip’ spectrum: A) Γ -dependence

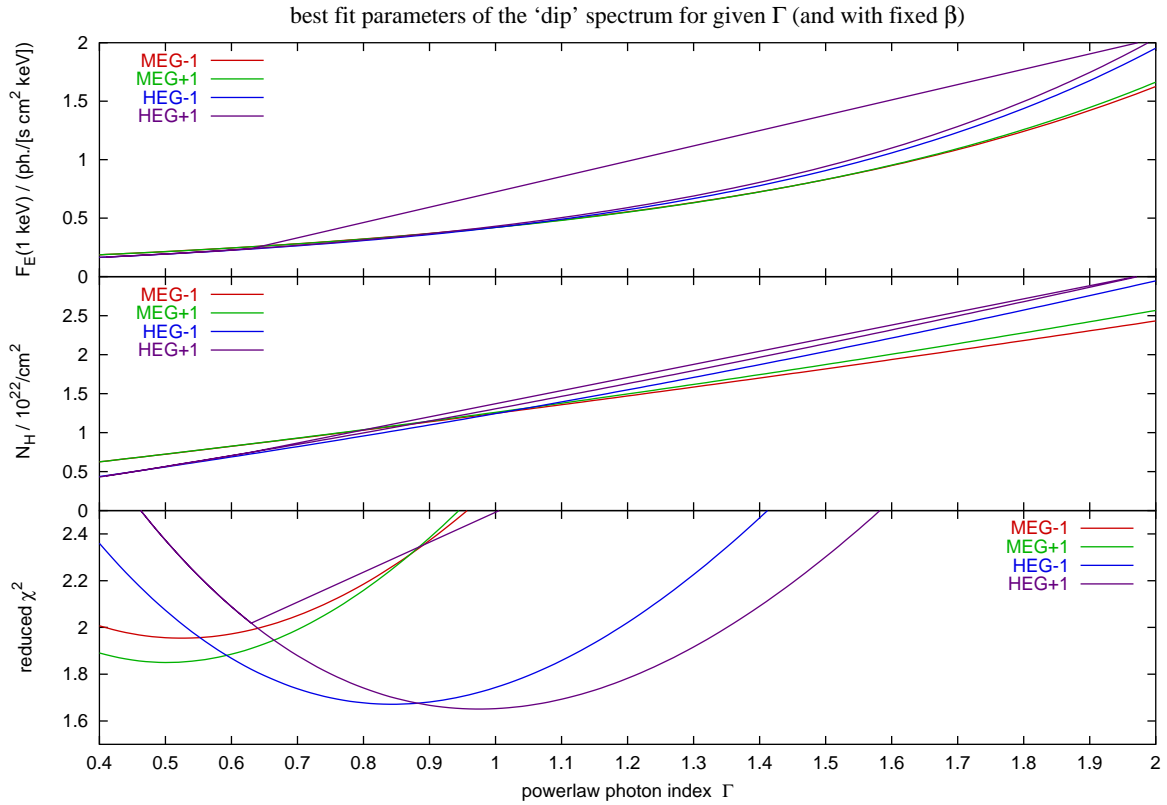


Figure A III.10: The ‘dip’ spectrum: Γ -dependence of the 1-comp. model’s fit parameters (rebinned to ≥ 30 counts/bin; $[1 \text{ \AA}, 20 \text{ \AA}]$ range noticed).

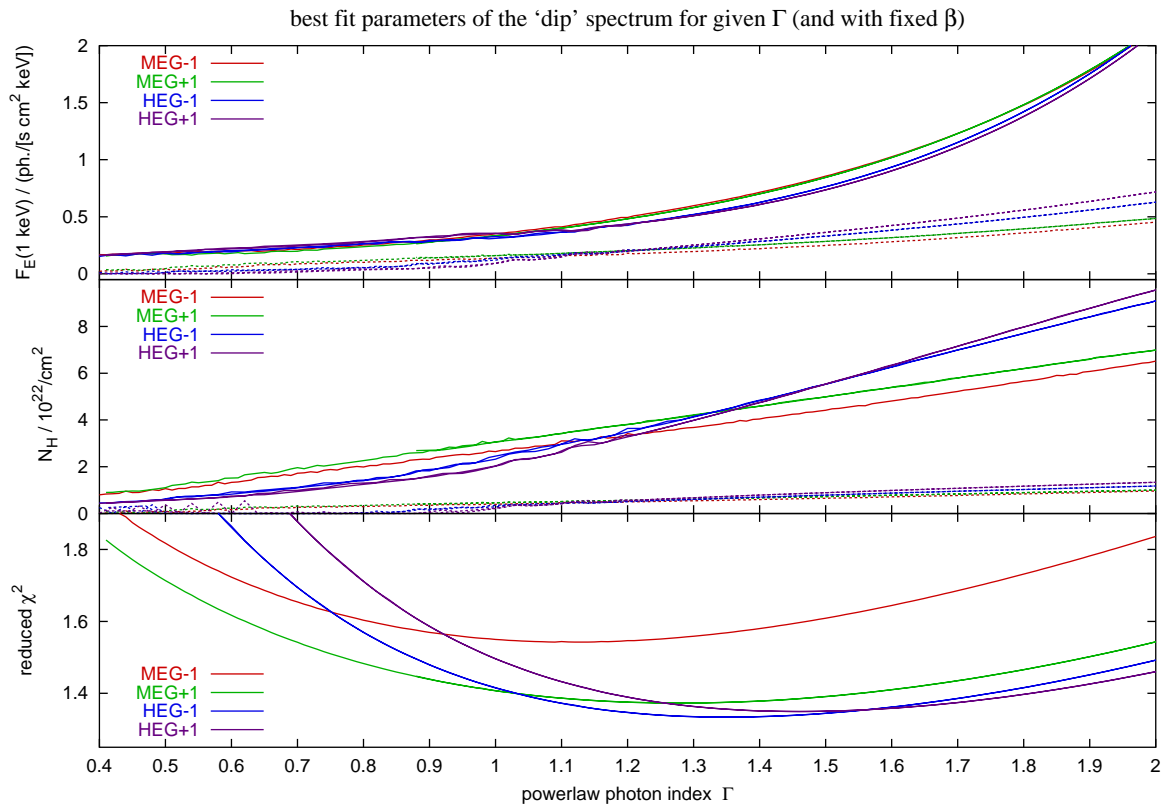


Figure A III.11: The ‘dip’ spectrum: Γ -dependence of the 2-comp. model’s fit parameters (rebinned to ≥ 30 counts/bin; $[1 \text{ \AA}, 20 \text{ \AA}]$ range noticed).

The 'dip' spectrum: B) λ_2 -dependence (with pile-up scales thawed)

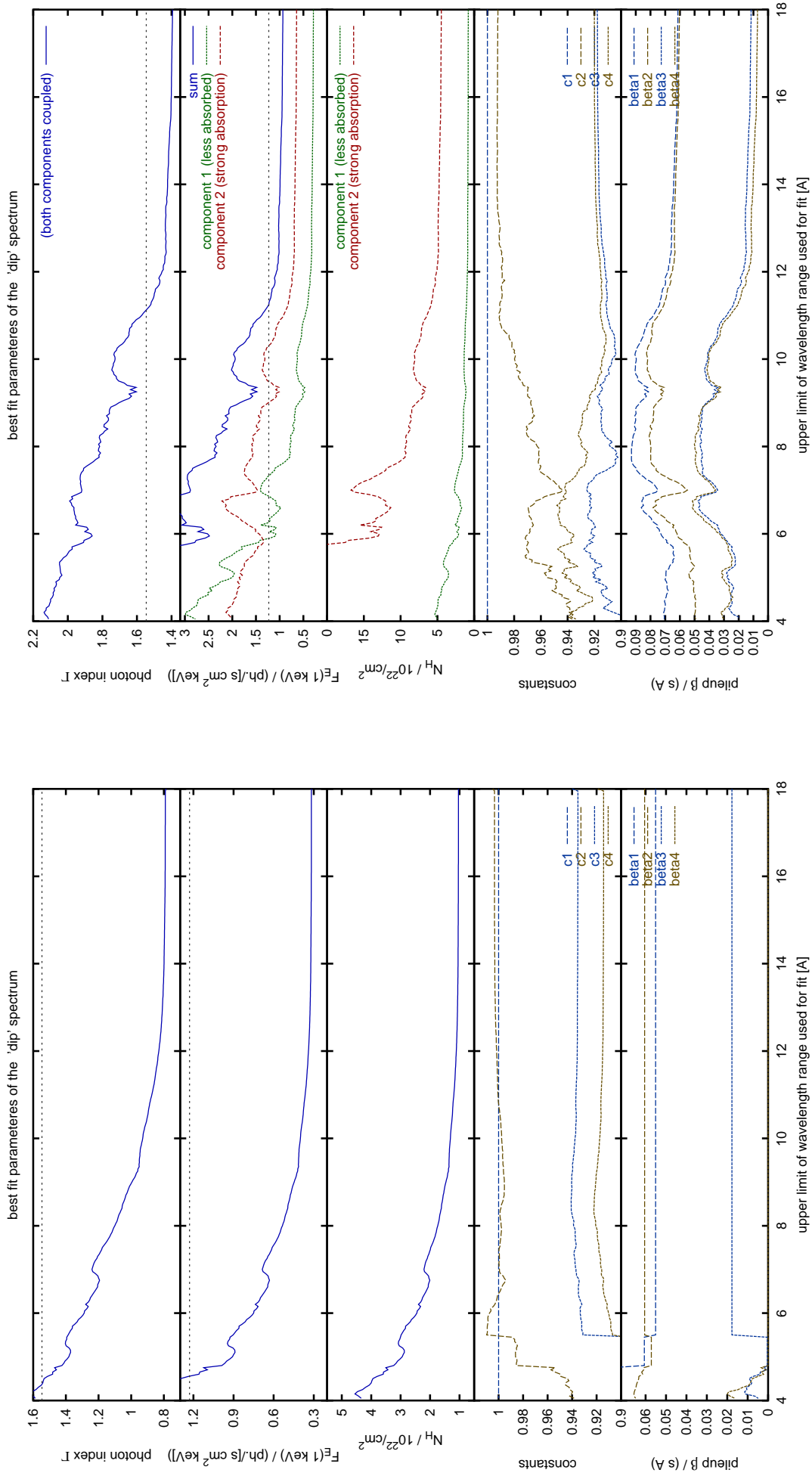


Figure A III.12: λ_2 -dependence of the 1-comp. model's fit parameters.

Figure A III.13: λ_2 -dependence of the 2-comp. model's fit parameters. For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied.

The ‘dip 1’ spectrum: A) Γ -dependence

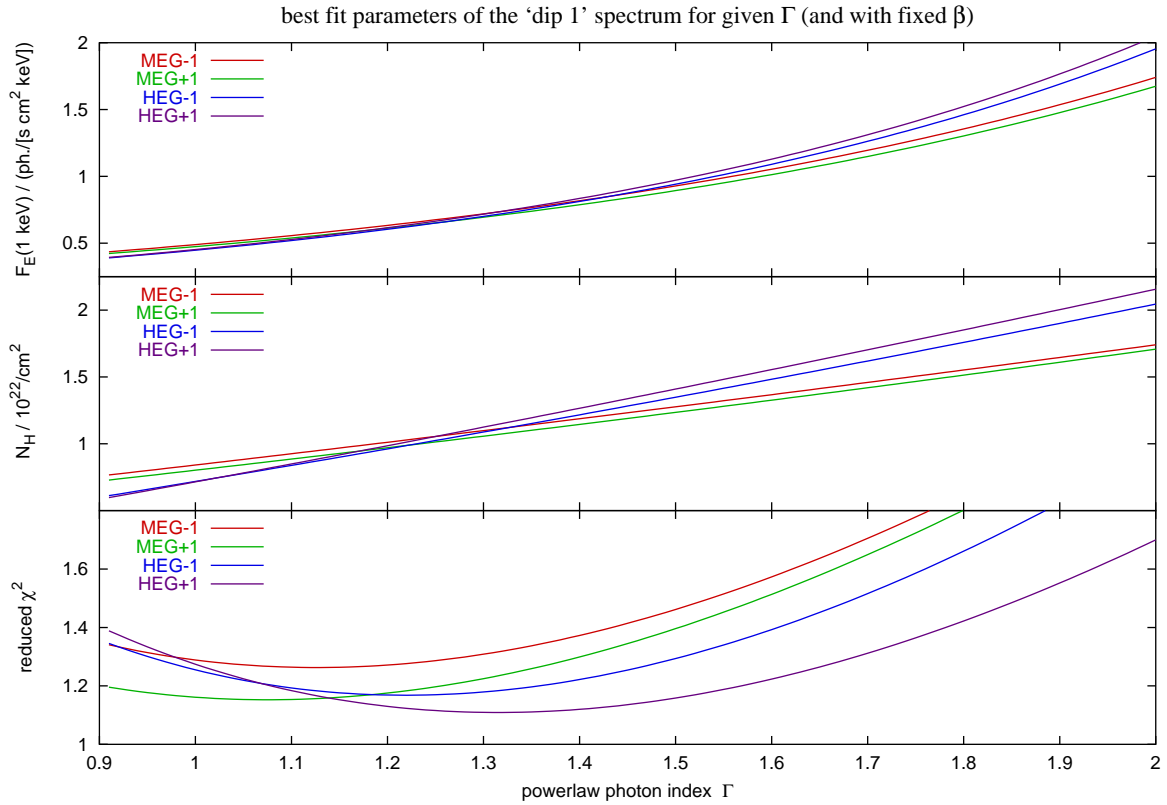


Figure A III.14: The ‘dip 1’ spectrum: Γ -dependence of the 1-comp. model’s fit parameters (rebinned to ≥ 30 counts/bin; $[1 \text{ \AA}, 20 \text{ \AA}]$ range noticed).

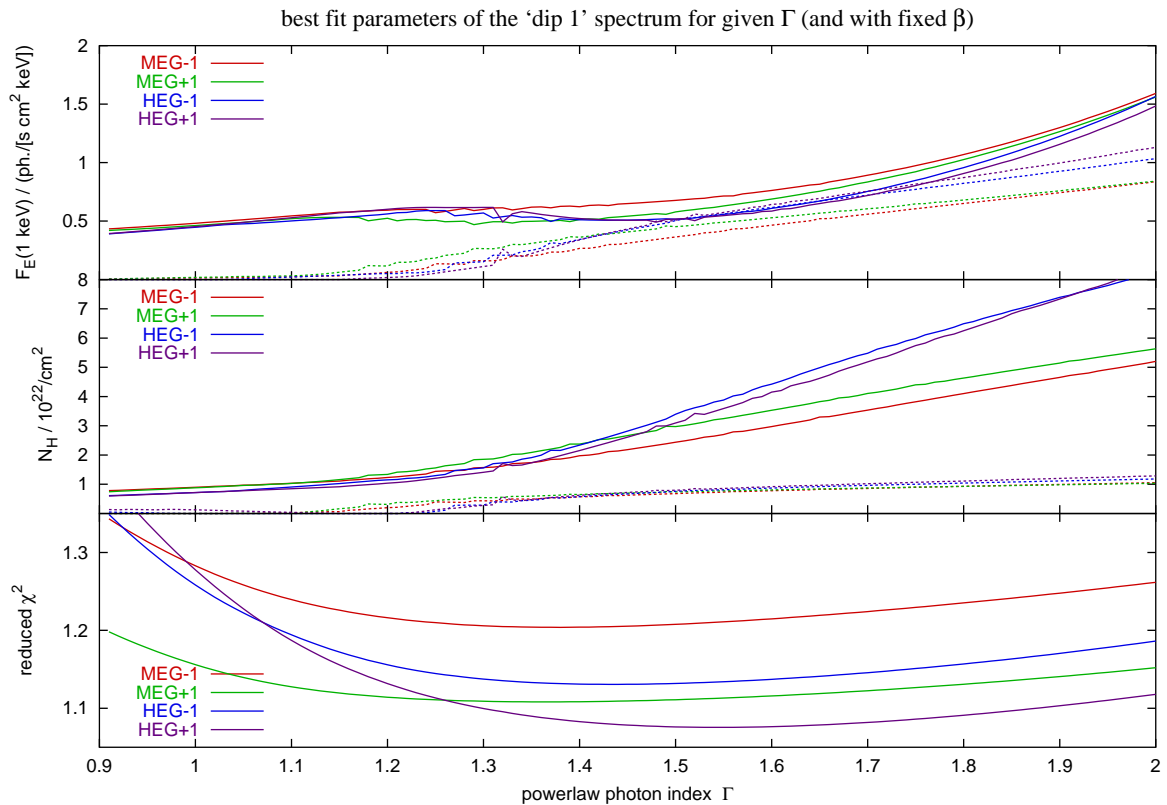


Figure A III.15: The ‘dip 1’ spectrum: Γ -dependence of the 2-comp. model’s fit parameters (rebinned to ≥ 30 counts/bin; $[1 \text{ \AA}, 20 \text{ \AA}]$ range noticed).

The 'dip 1' spectrum: B) λ_2 -dependence (with pile-up scales thawed)

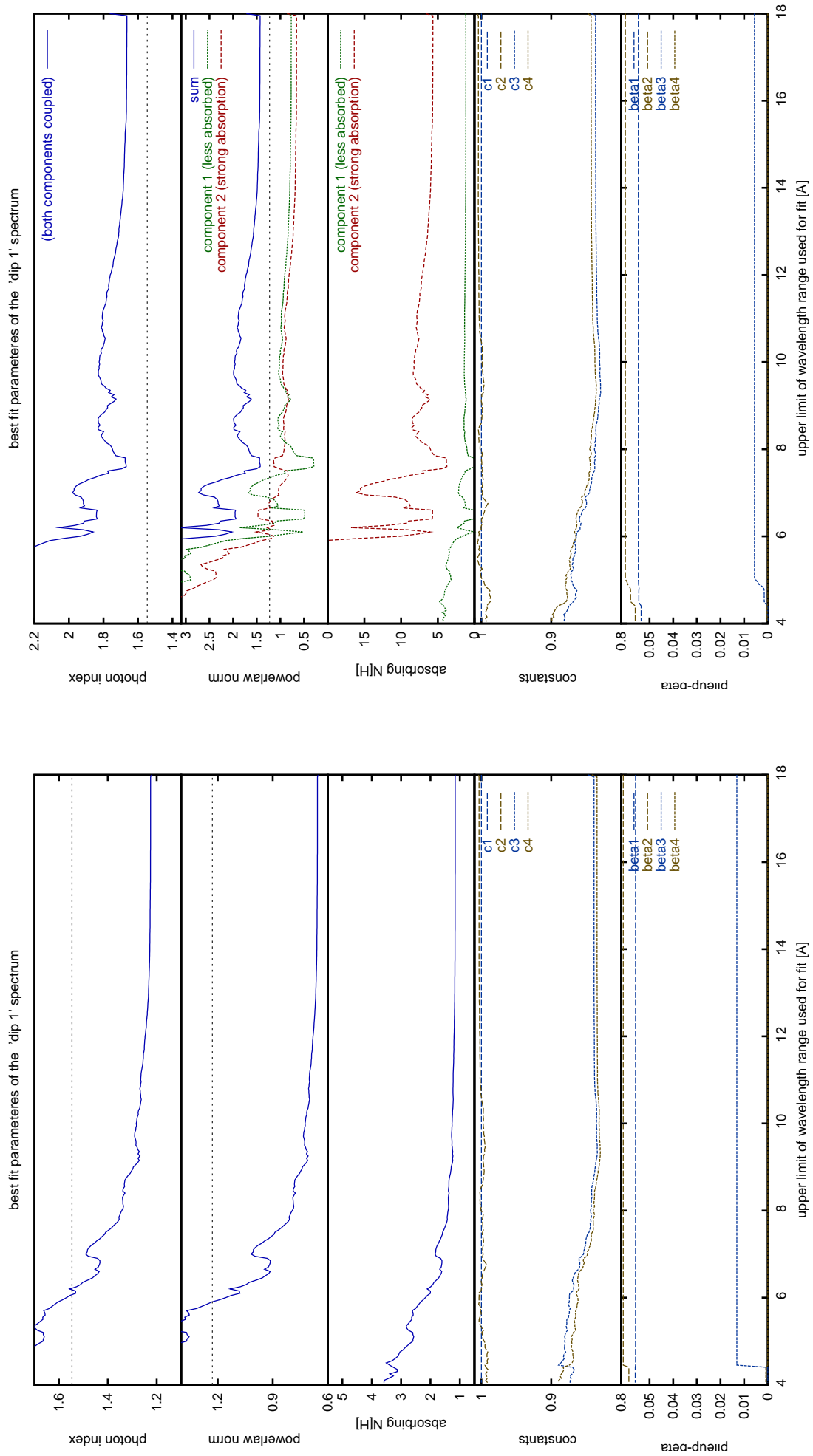


Figure A.III.16: λ_2 -dependence of the 1-comp. model's fit parameters.

Figure A.III.17: λ_2 -dependence of the 2-comp. model's fit parameters.

For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied.

The 'dip 1' spectrum: B') λ_2 -dependence (with fixed pile-up scales)

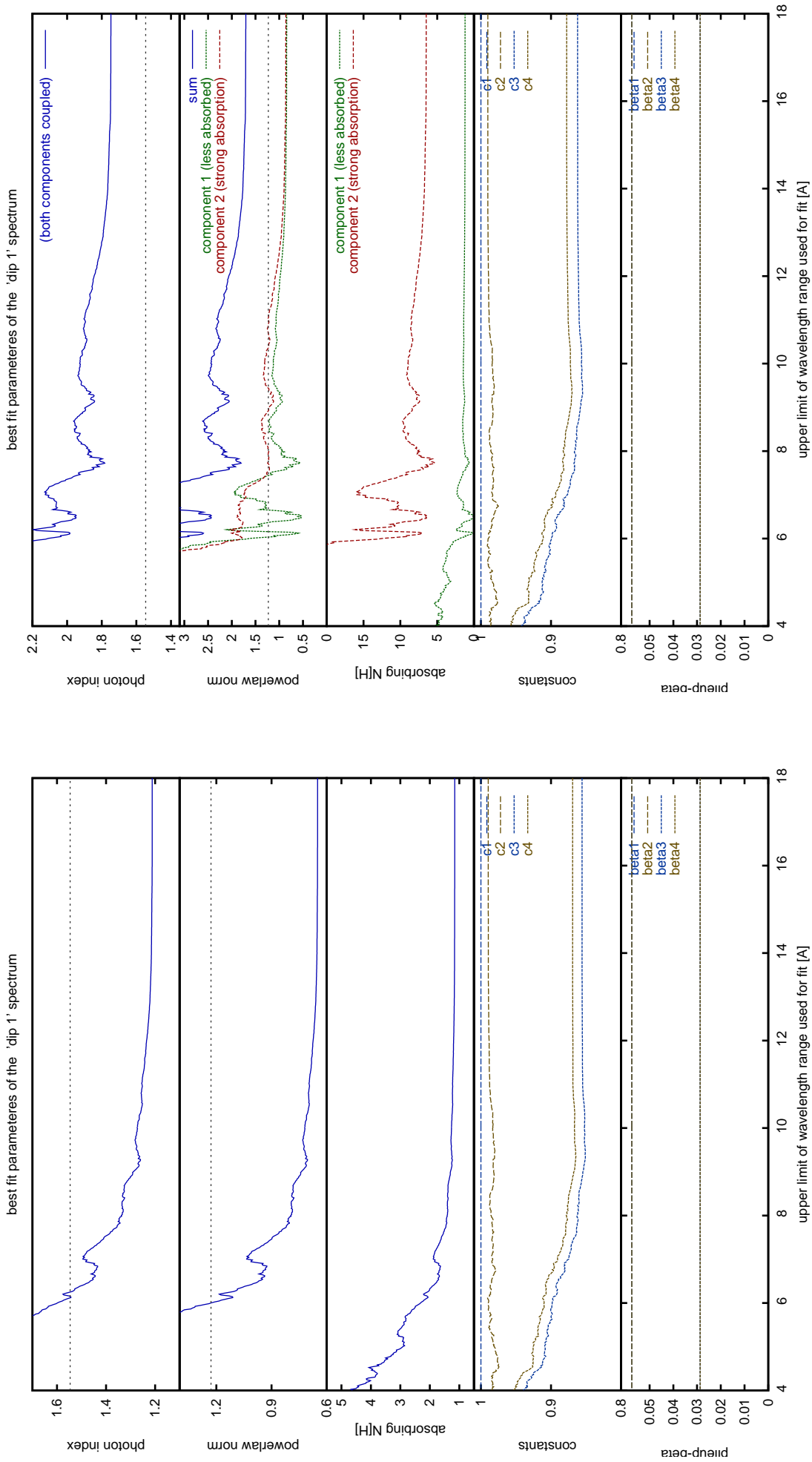


Figure A III.18: λ_2 -dependence of the 1-comp. model's fit par. (β fixed). For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied. The pile-up scales β have been fixed to the canonical values.

Figure A III.19: λ_2 -dependence of the 2-comp. model's fit par. (β fixed). For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied. The pile-up scales β have been fixed to the canonical values.

The ‘dip 2’ spectrum: A) Γ -dependence

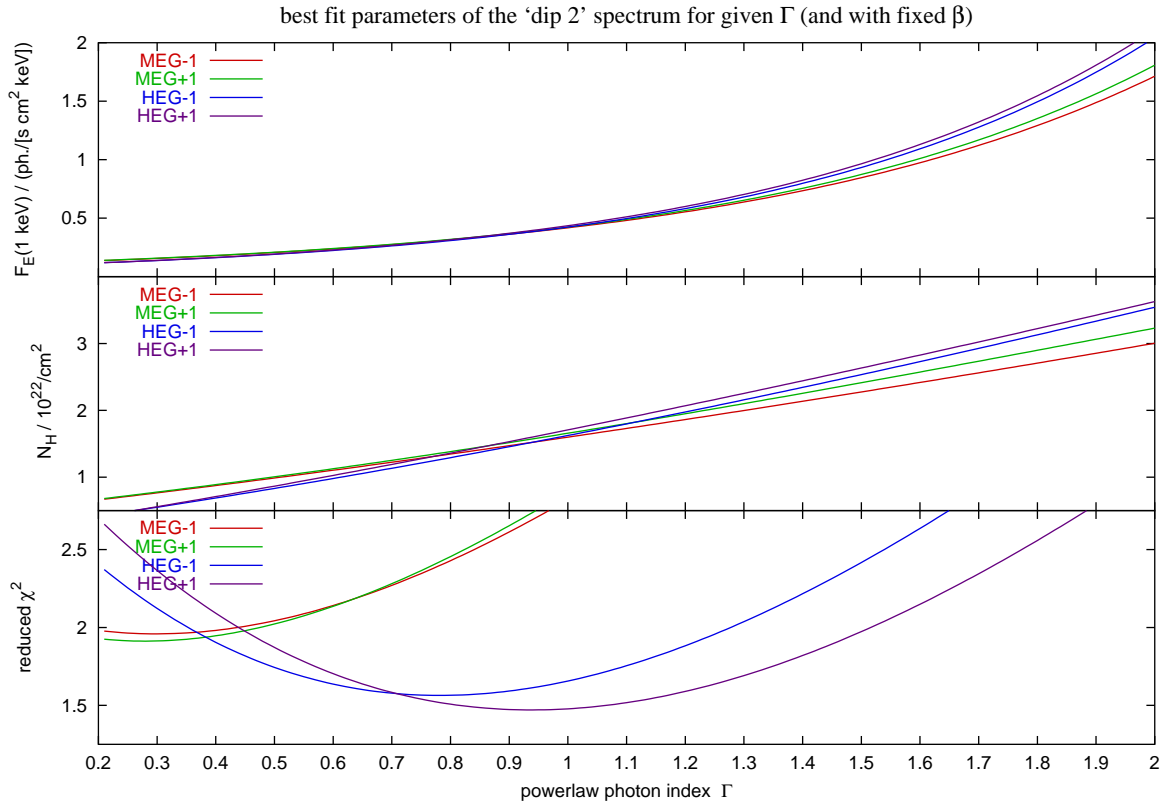


Figure A III.20: The ‘dip 2’ spectrum: Γ -dependence of the 1-comp. model’s fit parameters (rebinned to ≥ 30 counts/bin; $[1 \text{ \AA}, 20 \text{ \AA}]$ range noticed).

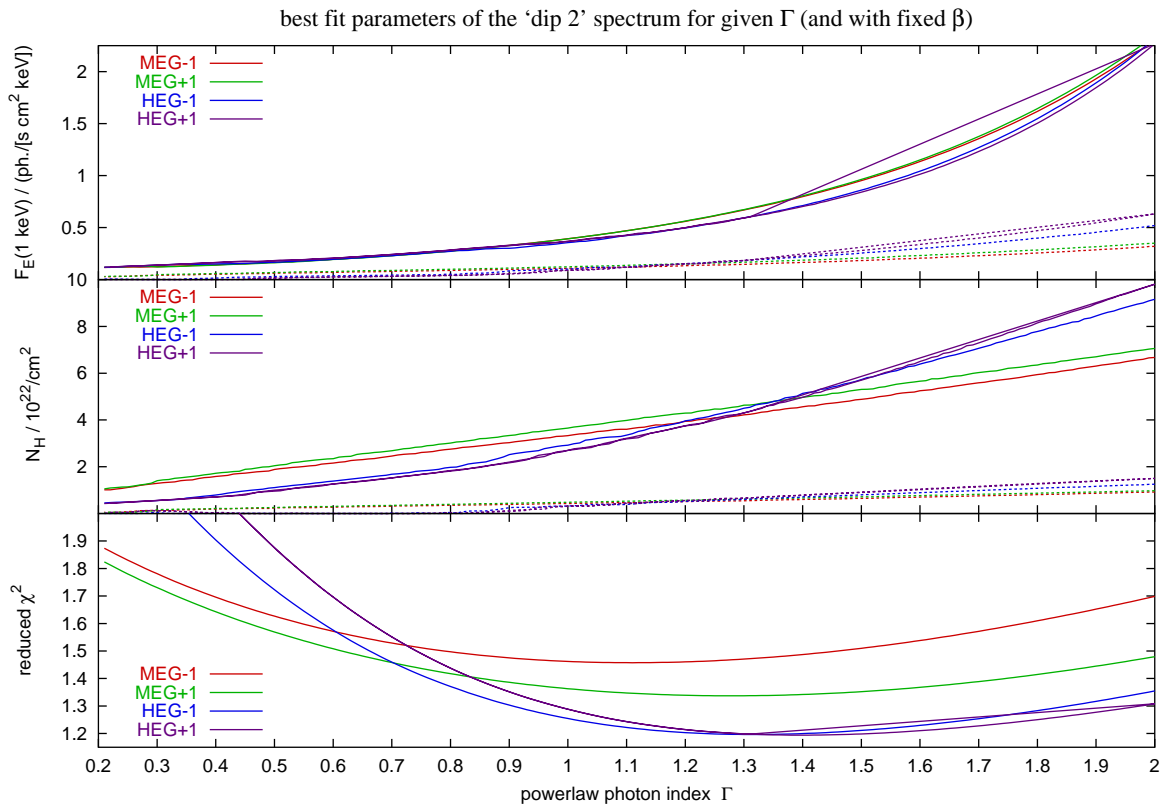


Figure A III.21: The ‘dip 2’ spectrum: Γ -dependence of the 2-comp. model’s fit parameters (rebinned to ≥ 30 counts/bin; $[1 \text{ \AA}, 20 \text{ \AA}]$ range noticed).

The 'dip 2' spectrum: B) λ_2 -dependence (with pile-up scales thawed)

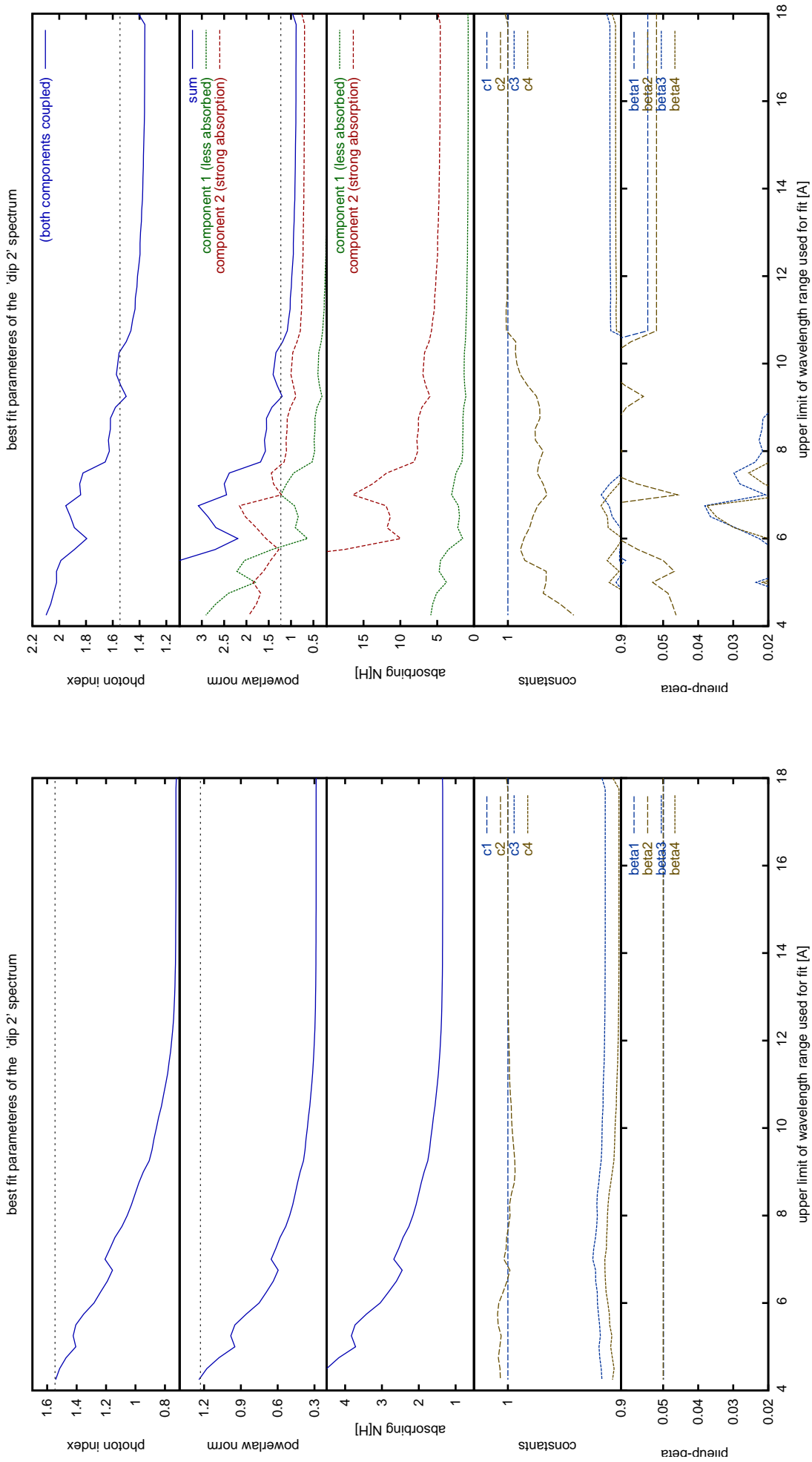


Figure A III.22: λ_2 -dependence of the 1-comp. model's fit parameters.

For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied.

Figure A III.23: λ_2 -dependence of the 2-comp. model's fit parameters.

The 'dip 2' spectrum: B') λ_2 -dependence (with fixed pile-up scales)

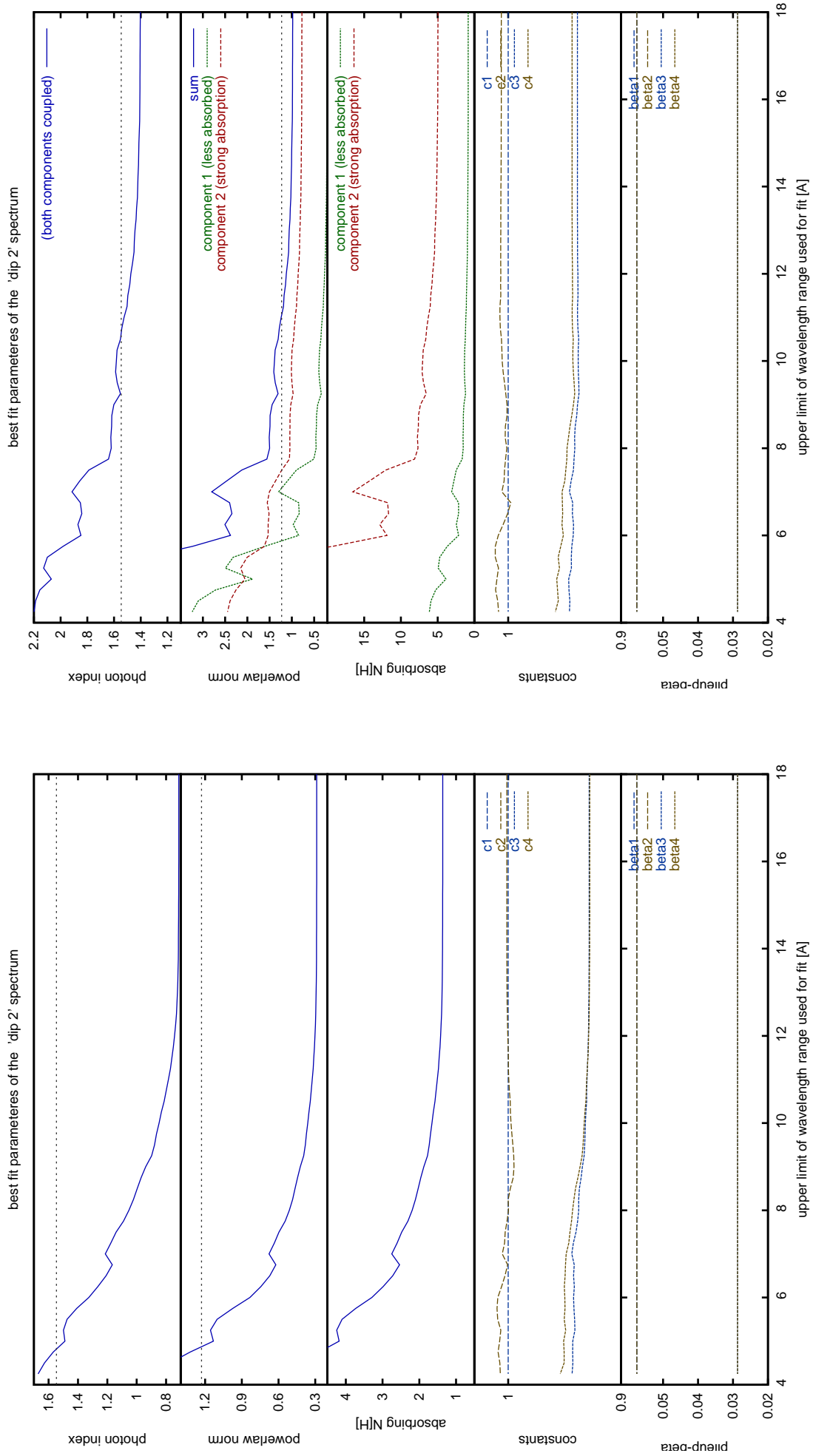


Figure A III.24: λ_2 -dependence of the 1-comp. model's fit par. (β fixed). For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied. The pile-up scales β have been fixed to the canonical values.

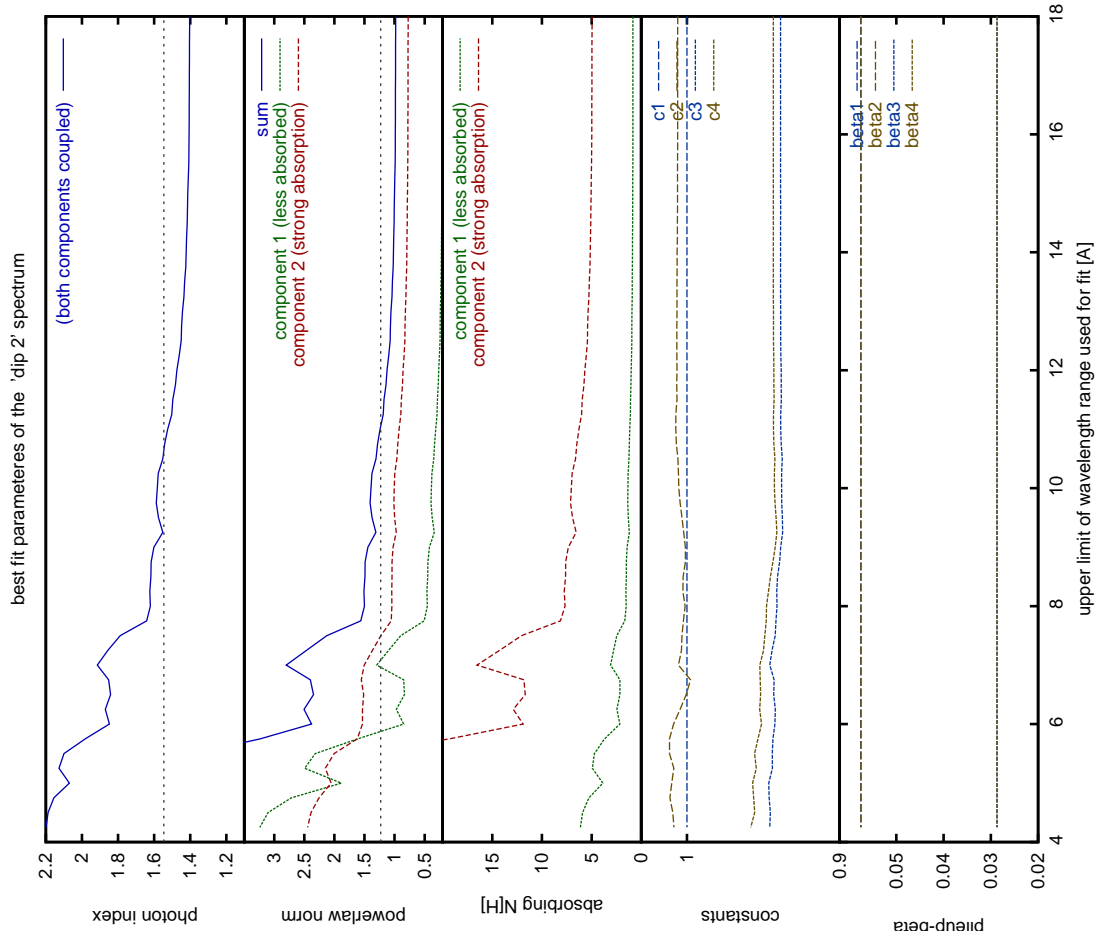


Figure A III.25: λ_2 -dependence of the 2-comp. model's fit par. (β fixed). For these investigations, the wavelength range [$\lambda_1 = 1 \text{ \AA}$, λ_2] that was considered for the fitting was varied. The pile-up scales β have been fixed to the canonical values.

III.4 Spectral analysis of the ‘dip’ spectrum

In section 4.3.3 (page 95), the absorption lines in the dip spectrum were investigated.

In the same way as in appendix III.2, first the list of all fitted lines is given (Table A III.2), and then the plots of the spectra are shown (Figs. A III.26a–A III.26h, pages 159–166). All further details are given on page 126.

Table A III.2: List of lines in the ‘dip’ spectrum – sorted by wavelength

λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å][10 ¹² s ⁻¹]	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
1.4924 ^{+0.0026} _{-0.0032}	0.33 ^{+11.44} _{-0.33}	-3.50 ^{+1.78} _{-2.96}	10.5	(Fe XXVI <u>1s</u> 3p)	1.5024	75.2	←	-1981 ⁺⁵¹¹ ₋₆₄₈	
				(Fe XXVI <u>1s</u> 3p)	1.5035	75.1		-2208 ⁺⁵¹⁰ ₋₆₄₈	
1.5199 ^{+0.0058} _{-0.0070}	15.00 ^{+0.00} _{-4.46}	-6.26 ^{+2.44} _{-2.41}	18.1	(Ni XXVIII <u>1s</u> 2p)	1.5304	379	←	-2049 ⁺¹¹⁴⁴ ₋₁₃₇₇	
				(Ni XXVIII <u>1s</u> 2p)	1.5356	378		-3055 ⁺¹¹⁴⁰ ₋₁₃₇₂	
1.8521 ^{+0.0052} _{-0.0046}	9.01 ^{+11.42} _{-9.01}	-2.36 ^{+1.24} _{-1.30}	13.9	Fe xxv	<u>1s</u> ² 1s2p	1.8504	503	←	280 ⁺⁸³⁹ ₋₇₄₈
1.9405 ^{+0.0049} _{-0.0032}	23.96 ^{+17.37} _{-8.89}	7.67 ^{+2.18} _{-1.81}	69.3	(Fe K α)	1.937		←	536 ⁺⁷⁵⁵ ₋₄₉₂	
3.0461 ^{+0.0063} _{-0.0194}	4.40 ^{+41.32} _{-4.40}	-0.92 ^{+0.75} _{-1.04}	4.1	(Ca XX <u>1s</u> 2p)	3.0185	98.6		2738 ⁺⁶²⁸ ₋₁₉₂₄	
				(Ca XX <u>1s</u> 2p)	3.0239	98.5		2196 ⁺⁶²⁷ ₋₁₉₂₀	
3.3077 ^{+0.0068} _{-0.0156}	10.02 ^{+40.03} _{-10.02}	-1.46 ^{+0.92} _{-1.76}	7.8						
3.3545 ^{+0.0055} _{-0.0145}	0.00 ^{+76.27} _{-0.00}	-0.80 ^{+0.74} _{-0.69}	3.2	(Ar XVII <u>1s</u> ² 1s3p)	3.3650	30.0	←	-939 ⁺⁴⁹³ ₋₁₂₉₀	
3.7350 ^{+0.0072} _{-0.0056}	8.89 ^{+17.25} _{-8.89}	-1.46 ^{+0.90} _{-1.07}	7.8	Ar XVIII <u>1s</u> 2p	3.7311	64.7	←	313 ⁺⁵⁷⁸ ₋₄₅₀	
				Ar XVIII <u>1s</u> 2p	3.7365	64.6		-121 ⁺⁵⁷⁸ ₋₄₅₀	
3.9566 ^{+0.0051} _{-0.0035}	5.91 ^{+19.24} _{-5.91}	-2.42 ^{+0.96} _{-1.68}	24.0	(Ar XVII <u>1s</u> ² 1s2p)	3.9491	109	←	569 ⁺³⁸⁹ ₋₂₆₆	
3.9940 ^{+0.0000} _{-0.0042}	0.37 ^{+7.84} _{-0.37}	-1.83 ^{+0.82} _{-0.76}	13.3	S XVI <u>1s</u> 3p	3.9908	10.8	←	239 ⁺¹ ₋₃₁₆	
				S XVI <u>1s</u> 3p	3.9920	10.8		153 ⁺¹ ₋₃₁₆	
4.3063 ^{+0.0200} _{-0.0200}	0.66 ^{+74.97} _{-0.66}	-1.41 ^{+0.94} _{-1.26}	6.0	S XV <u>1s</u> ² 1s3p	4.2990	18.3	←	511 ⁺¹³⁹⁵ ₋₁₃₉₅	
4.7240 ^{+0.0057} _{-0.0061}	25.65 ^{+14.98} _{-12.44}	-4.91 ^{+1.74} _{-1.86}	29.0	S XVI <u>1s</u> 2p	4.7274	40.4	←	-215 ⁺³⁶⁰ ₋₃₈₆	
				(S XVI <u>1s</u> 2p)	4.7328	40.3		-557 ⁺³⁶⁰ ₋₃₈₆	
5.0375 ^{+0.0050} _{-0.0025}	0.00 ^{+20.22} _{-0.00}	-1.99 ^{+0.89} _{-1.16}	11.5	S XV <u>1s</u> ² 1s2p	5.0387	66.7	←	-73 ⁺²⁹⁷ ₋₁₄₉	
5.1325 ^{+0.0025} _{-0.0050}	0.00 ^{+23.18} _{-0.00}	-2.51 ^{+1.05} _{-2.07}	14.4	S XV <u>1s</u> ² 2p (autoion.)	5.1293			187 ⁺¹⁴⁷ ₋₂₉₂	
				(S XV <u>1s</u> ² 1s2s)	5.1015	0.000		1822 ⁺¹⁴⁸ ₋₂₉₄	
5.1786 ^{+0.0052} _{-0.0051}	26.07 ^{+13.21} _{-8.46}	-6.38 ^{+1.89} _{-2.08}	36.6	(S XV <u>1s</u> ² 2p (autoion.))	5.2090			-1751 ⁺²⁹⁹ ₋₂₉₂	
5.2297 ^{+0.0020} _{-0.0020}	19.38 ^{+5.13} _{-4.34}	-10.98 ^{+1.74} _{-1.82}	153.3	(Si XIV <u>1s</u> 3p)	5.2168	6.32	←	740 ⁺¹¹⁷ ₋₁₁₇	
				(Si XIV <u>1s</u> 3p)	5.2180	6.31		674 ⁺¹¹⁷ ₋₁₁₇	
5.2747 ^{+0.0198} _{-0.0037}	18.51 ^{+35.57} _{-9.81}	-6.24 ^{+1.93} _{-6.39}	40.3	Si XIII <u>1s</u> ² 1s5p	5.2850	2.17		-587 ⁺¹¹²⁵ ₋₂₁₂	
5.3180 ^{+0.0031} _{-0.0023}	1.30 ^{+14.16} _{-1.30}	-3.08 ^{+1.20} _{-1.53}	0.0						
5.4075 ^{+0.0053} _{-0.0052}	0.05 ^{+47.10} _{-0.05}	-1.58 ^{+1.21} _{-1.88}	4.5	Si XIII <u>1s</u> ² 1s4p	5.4045	4.30	←	167 ⁺²⁹² ₋₂₈₈	
5.6875 ^{+0.0054} _{-0.0065}	20.66 ^{+22.18} _{-18.27}	-4.83 ^{+1.91} _{-2.43}	24.7	(Si XIII <u>1s</u> ² 1s3p)	5.6805	10.4	←	369 ⁺²⁸⁵ ₋₃₄₃	
				(Ni XXV <u>1s</u> ² 2s ² 1s ² 2p7d)	5.700	0.040		233 ⁺¹⁴⁹ ₋₁₁₆	
5.7050 ^{+0.0028} _{-0.0022}	0.36 ^{+12.83} _{-0.36}	4.53 ^{+1.51} _{-1.34}	29.8	(Ni XXV <u>1s</u> ² 2s ² 1s ² 2p7s)	5.709	0.000		-207 ⁺¹⁴⁹ ₋₁₁₆	
				(Ni XXVI <u>1s</u> ² 2s 1s ² 6p)	5.800	1.33		-354 ⁺²¹⁶ ₋₂₀₅	
				Ni XXV <u>1s</u> ² 2s2p 1s ² 2p7p	5.789	0.14		194 ⁺²¹⁶ ₋₂₀₅	
				Ni XXV <u>1s</u> ² 2s2p 1s ² 2p7p	5.793	0.11		-12 ⁺²¹⁶ ₋₂₀₅	
5.7935 ^{+0.0042} _{-0.0040}	1.66 ^{+16.02} _{-1.66}	2.08 ^{+1.28} _{-1.30}	7.2						
				Ni XXVII <u>1s</u> 2p 1s5d	5.8177	2.4e+04		-122 ⁺⁵²¹ ₋₆₄₄	
5.8153 ^{+0.0101} _{-0.0125}	18.84 ^{+27.86} _{-18.84}	-2.38 ^{+1.61} _{-1.90}	6.3	Ni XXVI <u>1s</u> ² 2p 1s ² 7d	5.8177	1.49		-122 ⁺⁵²¹ ₋₆₄₄	
				Ni XXVI <u>1s</u> ² 2p 1s ² 7d	5.8181	0.24		-141 ⁺⁵²¹ ₋₆₄₄	
				Ni XXVII <u>1s</u> 2p 1s5s	5.9064	0.40		170 ⁺¹⁴ ₋₃₆₇	
				(Ni XXVII <u>1s</u> 2p 1s5d)	5.8914	4.1e+04		932 ⁺¹⁴ ₋₃₆₈	
5.9097 ^{+0.0003} _{-0.0072}	0.00 ^{+16.16} _{-0.00}	-1.40 ^{+1.07} _{-1.04}	4.6	(Ni XXVII <u>1s</u> 2p 1s5d)	5.8914	3.3e+03		932 ⁺¹⁴ ₋₃₆₈	
				(Ni XXVII <u>1s</u> 2p 1s5d)	5.8944	1.1e+03		779 ⁺¹⁴ ₋₃₆₈	
				(Ni XXVII <u>1s</u> 2p 1s5d)	5.8944	7.1e+03		779 ⁺¹⁴ ₋₃₆₈	
6.0442 ^{+0.0048} _{-0.0045}	14.82 ^{+10.15} _{-8.86}	-3.22 ^{+1.24} _{-1.36}	19.9	(Al XIII <u>1s</u> 3p)	6.0526	4.70	←	-417 ⁺²⁴⁰ ₋₂₂₅	
				(Al XIII <u>1s</u> 3p)	6.0537	4.69		-471 ⁺²⁴⁰ ₋₂₂₅	
6.1814 ^{+0.0027} _{-0.0026}	24.59 ^{+6.87} _{-6.05}	-8.83 ^{+1.52} _{-1.59}	132.5	Si XIV <u>1s</u> 2p	6.1804	23.7	←	46 ⁺¹³⁰ ₋₁₂₇	
				(Si XIV <u>1s</u> 2p)	6.1858	23.6		-216 ⁺¹²⁹ ₋₁₂₇	
6.3148 ^{+0.0070} _{-0.0061}	14.19 ^{+13.54} _{-14.19}	-2.35 ^{+1.23} _{-1.36}	10.3	Al XII <u>1s</u> ² 1s4p	6.3140	3.14		38 ⁺³³² ₋₂₉₁	
				(Ni XXV <u>1s</u> ² 2p ² 1s ² 2p6d)	6.3239	3.64		-430 ⁺³³² ₋₂₉₁	
				Ni XXV <u>1s</u> ² 2p ² 1s ² 2p6d	6.3175	1.91		-127 ⁺³³² ₋₂₉₁	
6.4552 ^{+0.0039} _{-0.0052}	0.49 ^{+16.17} _{-0.49}	2.45 ^{+1.38} _{-1.17}	11.9	Ni XXV <u>1s</u> ² 2s2p 1s ² 2p5p	6.453	1.88		93 ⁺¹⁸³ ₋₂₄₂	
				Ni XXV <u>1s</u> ² 2s2p 1s ² 2p5p	6.458	1.39		-163 ⁺¹⁸³ ₋₂₄₁	
6.6461 ^{+0.0026} _{-0.0027}	11.50 ^{+7.78} _{-5.60}	-5.52 ^{+1.36} _{-1.58}	65.2	Si XIII <u>1s</u> ² 1s2p	6.6479	37.7	←	-81 ⁺¹¹⁹ ₋₁₂₀	

Table A III.2: List of lines in the ‘dip’ spectrum – sorted by wavelength (continued)

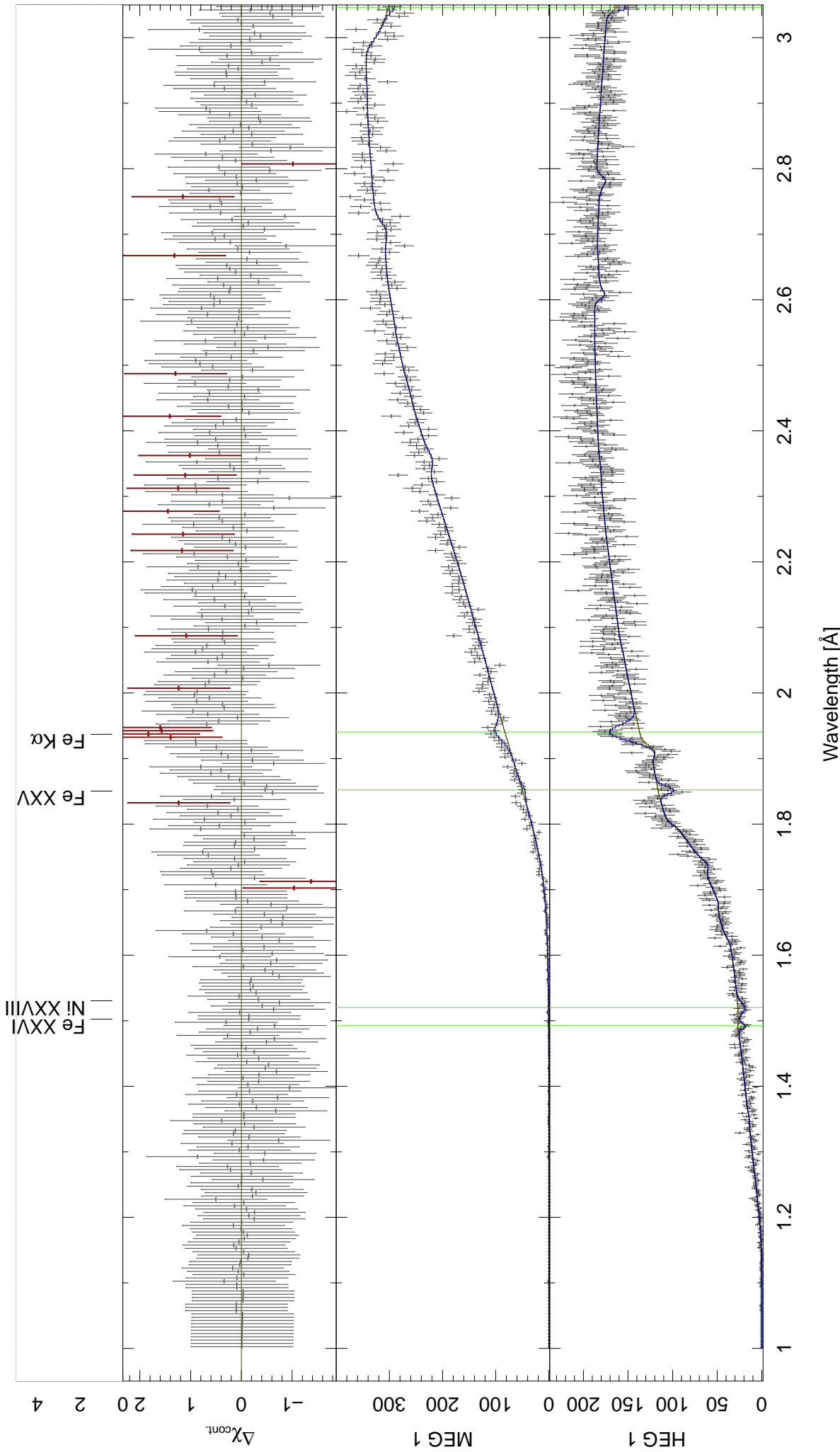
λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å][10 ¹² s ⁻¹]	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]	
6.7243 ^{+0.0009} _{-0.0045}	0.09 ^{+21.23} _{-0.09}	-1.96 ^{+0.98} _{-1.18}	10.6	(Mg XII	<u>1s</u> 4p	6.7378	1.39) ←	-598 ⁺⁴¹ ₋₂₀₂
				(Mg XII	<u>1s</u> 4p	6.7382	1.39) ←	-617 ⁺⁴¹ ₋₂₀₂
6.7426 ^{+0.0024} _{-0.0010}	1.02 ^{+6.56} _{-1.02}	6.88 ^{+1.35} _{-1.44}	79.0	(Si XIII	<u>1s²</u> 1s2s	6.740	0.000) ←	103 ⁺¹⁰⁶ ₋₄₅
6.7867 ^{+0.0020} _{-0.0021}	17.89 ^{+5.97} _{-4.72}	-7.64 ^{+1.31} _{-1.41}	134.3	(Si XI	K α	6.8130) ←	-1158 ⁺⁸⁹ ₋₉₂
6.8536 ^{+0.0016} _{-0.0013}	9.96 ^{+4.04} _{-9.96}	-6.16 ^{+1.73} _{-1.00}	154.5	(Si X	K α	6.8820) ←	-1239 ⁺⁶⁸ ₋₅₆
6.8659 ^{+0.0016} _{-0.0016}	15.57 ^{+4.95} _{-3.60}	-7.81 ^{+1.17} _{-1.24}	178.4	(Si X	K α	6.8820) ←	-702 ⁺⁷⁰ ₋₇₁
6.9236 ^{+0.0014} _{-0.0013}	16.74 ^{+3.67} _{-3.74}	-10.09 ^{+1.28} _{-1.22}	274.7	(Si IX	K α	6.9470) ←	-1010 ⁺⁶⁰ ₋₅₈
6.9400 ^{+0.0025} _{-0.0001}	0.02 ^{+8.26} _{-0.02}	-4.70 ^{+0.74} _{-0.67}	107.9	(Si IX	K α	6.9470) ←	-302 ⁺¹⁰⁶ ₋₂
6.9981 ^{+0.0013} _{-0.0012}	10.87 ^{+4.53} _{-4.30}	-8.26 ^{+1.26} _{-1.35}	207.6	(Si VIII	K α	7.0070) ←	-380 ⁺⁵⁸ ₋₅₂
7.0550 ^{+0.0003} _{-0.0029}	0.07 ^{+7.53} _{-0.07}	-3.27 ^{+0.99} _{-0.90}	32.7	(Si VII	K α	7.0630) ←	-341 ⁺¹³ ₋₁₂₂
7.1080 ^{+0.0047} _{-0.0030}	0.21 ^{+20.25} _{-0.21}	-2.05 ^{+0.99} _{-1.18}	11.3	Mg XII	<u>1s</u> 3p	7.1058	3.41 ←	95 ⁺¹⁹⁶ ₋₁₂₆
				Mg XII	<u>1s</u> 3p	7.1069	3.41 ←	47 ⁺¹⁹⁶ ₋₁₂₆
7.1561 ^{+0.0240} _{-0.0260}	0.01 ^{+50.42} _{-0.01}	-0.76 ^{+0.76} _{-1.46}	1.3	Fe XXIV	<u>1s²2s</u> 1s ² 5p	7.1690	1.71	-539 ⁺¹⁰⁰⁵ ₋₁₀₈₉
				Fe XXIV	<u>1s²2s</u> 1s ² 5p	7.1690	1.69	-539 ⁺¹⁰⁰⁵ ₋₁₀₈₉
				Al XIII	<u>1s</u> 2p	7.1710	17.6	-623 ⁺¹⁰⁰⁵ ₋₁₀₈₉
				Al XIII	<u>1s</u> 2p	7.1764	17.6	-848 ⁺¹⁰⁰⁴ ₋₁₀₈₈
				Fe XXVI	2p 4d	7.1712	9.27	-631 ⁺¹⁰⁰⁵ ₋₁₀₈₉
				Fe XXVI	2p 4d	7.1748	1.54	-781 ⁺¹⁰⁰⁵ ₋₁₀₈₈
7.1762 ^{+0.0038} _{-0.0062}	0.01 ^{+17.29} _{-0.01}	-1.40 ^{+1.03} _{-1.17}	4.9	Al XIII	<u>1s</u> 2p	7.1710	17.6 ←	218 ⁺¹⁵⁷ ₋₂₅₉
				Al XIII	<u>1s</u> 2p	7.1764	17.6	-8 ⁺¹⁵⁷ ₋₂₅₉
7.4700 ^{+0.0025} _{-0.0000}	0.00 ^{+6.59} _{-0.00}	-3.23 ^{+0.97} _{-0.95}	28.2	(Mg XI	<u>1s²</u> 1s4p	7.4730	2.24) ←	-120 ⁺¹⁰⁰ ₋₀
7.7750 ^{+0.0171} _{-0.0097}	0.00 ^{+23.68} _{-19.88}	-2.08 ^{+2.45} _{-2.90}	0.0	(Al XII	<u>1s²</u> 1s2p	7.7573	27.5) ←	684 ⁺⁶⁶² ₋₃₇₃
7.8024 ^{+0.0057} _{-0.0056}	25.11 ^{+13.65} _{-11.32}	6.32 ^{+2.28} _{-2.19}	27.4	Al XII	<u>1s²</u> 1s2p	7.8070	0.082 ←	-175 ⁺²¹⁷ ₋₂₁₄
				Al XII	<u>1s²</u> 1s2p	7.8030	0.000	-55 ⁺²¹⁷ ₋₂₁₄
7.8550 ^{+0.0040} _{-0.0038}	17.84 ^{+11.35} _{-8.58}	-5.79 ^{+1.73} _{-1.79}	40.5	(Mg XI	<u>1s²</u> 1s3p	7.8503	5.43) ←	180 ⁺¹⁵¹ ₋₁₄₆
7.8776 ^{+0.0224} _{-0.0176}	0.02 ^{+49.98} _{-0.02}	1.07 ^{+1.36} _{-1.07}	1.7	Al XII	<u>1s²</u> 1s2s	7.8720	0.000 ←	210 ⁺⁸⁵² ₋₆₇₂
8.0257 ^{+0.0136} _{-0.0155}	46.64 ^{+8.76} _{-38.22}	-5.00 ^{+3.13} _{-2.78}	13.0	(Ni XXII	2s2p ⁴ 2s2p ³ 5d	8.0494	4.98)	-884 ⁺⁵⁰⁵ ₋₅₇₉
				(Ni XXIV	1s ² 2s2p ² 1s ² 2s2p4d	7.9965	4.47)	1094 ⁺⁵⁰⁸ ₋₅₈₃
				(Ni XXIII	1s ² 2s ² 2p ² 1s ² 2s2p2p4p	8.0400	4.11)	-533 ⁺⁵⁰⁵ ₋₅₇₉
				(Ni XXIV	1s ² 2s2p ² 1s ² 2s2p4d	8.0437	4.10)	-674 ⁺⁵⁰⁵ ₋₅₇₉
8.0706 ^{+0.0069} _{-0.0006}	0.00 ^{+17.80} _{-0.00}	2.70 ^{+1.48} _{-1.44}	9.8	(Ni XXIV	1s ² 2s2p ² 1s ² 2s2p4d	8.0820	12.4)	-434 ⁺²⁵⁷ ₋₂₂
8.4220 ^{+0.0151} _{-0.0149}	42.69 ^{+0.00} _{-0.00}	-8.28 ^{+0.00} _{-0.00}	0.0	Mg XII	<u>1s</u> 2p	8.4192	12.8 ←	99 ⁺⁵³⁷ ₋₅₃₂
				Mg XII	<u>1s</u> 2p	8.4246	12.8	-93 ⁺⁵³⁷ ₋₅₃₂
				(Ni XXIII	1s ² 2s2p ³ 1s ² 2s2p2p4d	8.4499	4.37)	-991 ⁺⁵³⁵ ₋₅₃₀
8.6551 ^{+0.0158} _{-0.0106}	74.76 ^{+0.59} _{-66.28}	-11.31 ^{+2.62} _{-3.80}	24.5	Ni XX	2s ² 2p ⁵ 2s ² 2p ⁴ 5d	8.6556	2.81	-19 ⁺⁵⁴⁶ ₋₃₆₈
				Ni XXIII	1s ² 2p ⁴ 1s ² 2p ² 2p4d	8.6564	3.33	-47 ⁺⁵⁴⁵ ₋₃₆₈
				Fe XXI	1s ² 2s2p ³ 1s ² 2s2p ² 5d	8.6582	2.84	-106 ⁺⁵⁴⁵ ₋₃₆₈
				(Ni XXVII	1s2p 1s3d	8.7069	1.7e+05)	-1785 ⁺⁵⁴² ₋₃₆₆
				(Ni XXVII	1s2p 1s3d	8.7135	1.5e+04)	-2008 ⁺⁵⁴² ₋₃₆₆
				(Ni XXVII	1s2s 1s3p	8.6102	11.9)	1563 ⁺⁵⁴⁸ ₋₃₇₀
9.2311 ^{+0.0026} _{-0.0016}	7.16 ^{+5.39} _{-7.16}	13.39 ^{+3.29} _{-2.25}	72.8	Mg XI	<u>1s²</u> 1s2p	9.2310	0.034 ←	-3 ⁺⁸⁶ ₋₅₂
9.2860 ^{+0.0037} _{-0.0034}	21.55 ^{+10.54} _{-6.45}	-11.18 ^{+2.62} _{-3.12}	63.7	(Ni XX	2s2p ⁶ 2s2p2p ⁴ 4d	9.2618	7.19)	784 ⁺¹²⁰ ₋₁₁₀
				(Fe XXII	1s ² 2s2p ² 1s ² 2s2p4d	9.2630	5.69)	746 ⁺¹²⁰ ₋₁₁₀
				(Ni XX	2s2p ⁶ 2s2p ² 2p ³ 4d	9.2644	4.87)	698 ⁺¹²⁰ ₋₁₁₀
				(Ni XXV	1s ² 2s2p 1s ² 2p3p	9.2681	8.79)	580 ⁺¹²⁰ ₋₁₁₀
9.3155 ^{+0.0029} _{-0.0029}	16.62 ^{+7.32} _{-6.37}	13.46 ^{+3.39} _{-3.25}	54.3	Mg XI	<u>1s²</u> 1s2s	9.3140	0.000	36 ⁺⁹⁴ ₋₉₃
9.3819 ^{+0.0019} _{-0.0021}	15.16 ^{+8.42} _{-4.97}	-13.22 ^{+2.29} _{-2.98}	140.6	Fe XXII	1s ² 2s2p ² 1s ² 2s2p4d	9.3824	5.12	-15 ⁺⁶² ₋₆₆
				(Fe XX	2s2p ⁴ 2p ² 2p ² 4p	9.3797	0.60)	71 ⁺⁶² ₋₆₆
				Fe XX	2s2p ⁴ 2s2p2p ² 5d	9.3833	0.44	-44 ⁺⁶² ₋₆₆
				(Fe XX	2s2p ⁴ 2s2p2p ² 5d	9.3797	0.25)	72 ⁺⁶² ₋₆₆
9.4794 ^{+0.0043} _{-0.0131}	18.05 ^{+11.98} _{-8.94}	-8.69 ^{+2.62} _{-2.80}	39.0	Fe XXI	1s ² 2s ² 2p ² 1s ² 2s ² 2p4d	9.4797	6.12 ←	-11 ⁺¹³⁸ ₋₄₁₃
				Ne X	<u>1s</u> 5p	9.4807	0.34	-42 ⁺¹³⁸ ₋₄₁₃
				Ne X	<u>1s</u> 5p	9.4809	0.34	-48 ⁺¹³⁸ ₋₄₁₃
9.5006 ^{+0.0018} _{-0.0032}	8.34 ^{+8.00} _{-6.69}	-7.82 ^{+2.28} _{-2.21}	53.1	(Ni XX	2s ² 2p ⁵ 2s ² 2p2p ³ 4d	9.4966	10.3)	124 ⁺⁵⁸ ₋₁₀₀
				(Ni XX	2s ² 2p ⁵ 2s ² 2p ⁴ 4d	9.4966	6.58)	124 ⁺⁵⁸ ₋₁₀₀
				(Fe XXI	1s ² 2s2p ³ 1s ² 2s2p ² 4d	9.4973	1.95)	103 ⁺⁵⁸ ₋₁₀₀

Table A III.2: List of lines in the ‘dip’ spectrum – sorted by wavelength (continued)

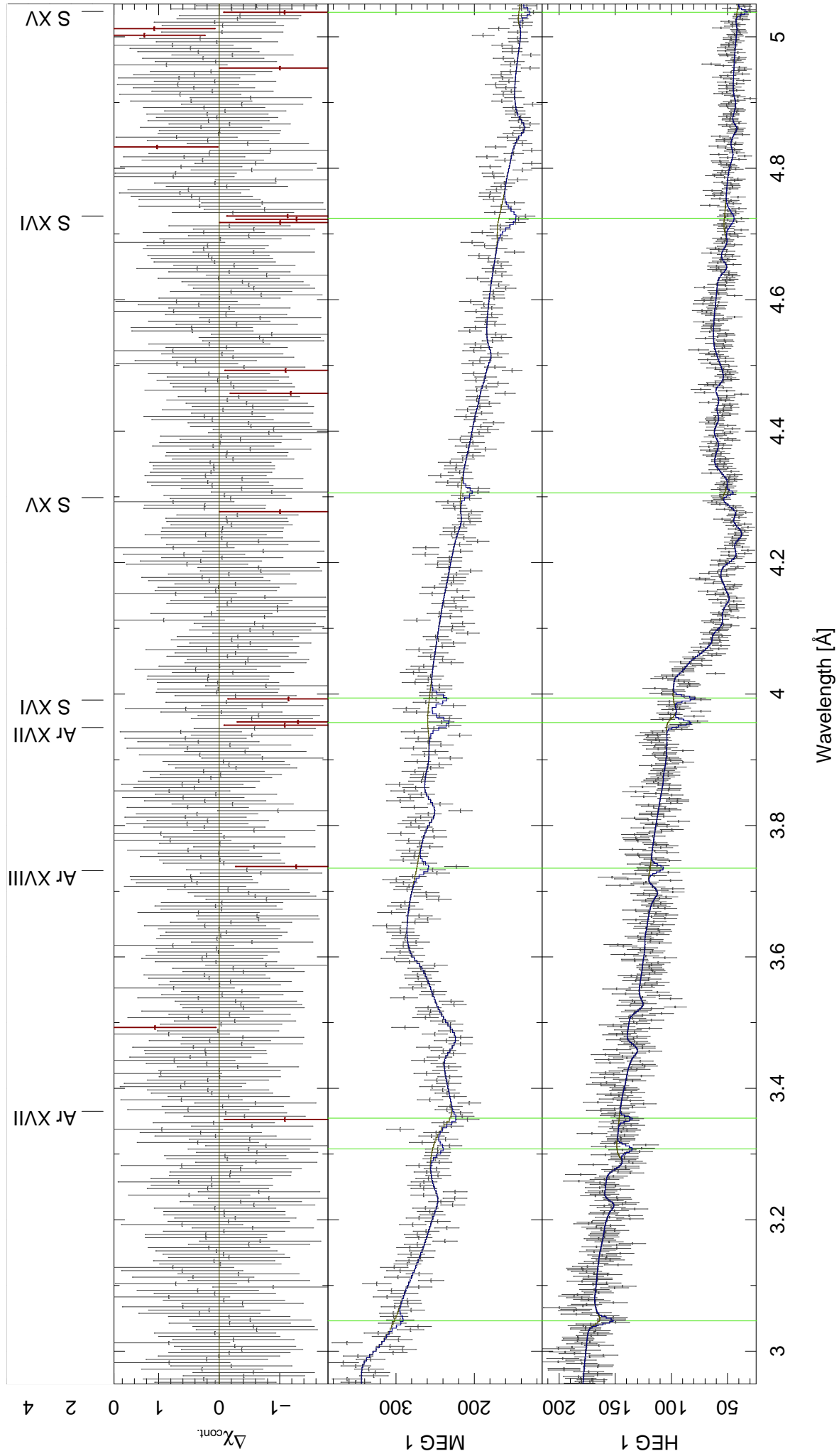
λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition λ_0 [Å] $[10^{12}\text{s}^{-1}]$	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]		
9.5167 ^{+0.0035} _{-0.0028}	11.68 ^{+8.57} _{-6.83}	-7.20 ^{+2.14} _{-2.45}	36.5	Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p 2p 4d$	9.5178	4.39	-36 ⁺¹¹¹ ₋₈₈
				(Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p 2p 4d$	9.5120	4.02	145 ⁺¹¹¹ ₋₈₈
				(Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p 2p 4d$	9.5213	3.29	-147 ⁺¹¹¹ ₋₈₈
				Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p 2p 4d$	9.5146	2.45	65 ⁺¹¹¹ ₋₈₈
				Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p 2p 4d$	9.5140	2.33	83 ⁺¹¹¹ ₋₈₈
				(Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 4d$	9.5231	2.09	-201 ⁺¹¹¹ ₋₈₈
				Ni XX	$2s 2p^6$	$2s 2p 2p^4 4d$	9.5196	6.21	-93 ⁺¹¹¹ ₋₈₈
9.5963 ^{+0.0054} _{-0.0053}	26.43 ^{+13.15} _{-10.46}	-9.94 ^{+3.12} _{-3.35}	35.4	Ni XXV	$1s^2 2s 2p$	$1s^2 2s 3d$	9.6010	17.3	-147 ⁺¹⁷⁰ ₋₁₆₄
				(Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 4d$	9.6059	8.91	-300 ⁺¹⁷⁰ ₋₁₆₄
				Fe XXI	$1s^2 2p^4$	$1s^2 2p^3 4d$	9.5926	4.75	115 ⁺¹⁷⁰ ₋₁₆₄
				Fe XXI	$1s^2 2s^2 2p^2$	$1s^2 2s^2 2p 4d$	9.5917	3.77	144 ⁺¹⁷⁰ ₋₁₆₄
9.6276 ^{+0.0024} _{-0.0002}	0.01 ^{+15.92} _{-0.01}	-5.43 ^{+1.64} _{-1.62}	29.2	Ni XXV	$1s^2 2s 2p$	$1s^2 2s 3d$	9.6300	12.6	-76 ⁺⁷⁶ ₋₆
				Ni XX	$2s 2p^6$	$2s 2p^2 2p^3 4d$	9.6291	2.37	-46 ⁺⁷⁶ ₋₆
9.6489 ^{+0.0041} _{-0.0039}	5.39 ^{+9.63} _{-5.39}	-5.06 ^{+1.88} _{-2.33}	19.1	(Fe XXI	$1s^2 2p^4$	$1s^2 2p^3 4d$	9.6582	6.64	-289 ⁺¹²⁸ ₋₁₂₁
				(Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 4d$	9.6421	2.71	212 ⁺¹²⁸ ₋₁₂₁
				Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p 2p 4d$	9.6500	2.59	-33 ⁺¹²⁸ ₋₁₂₁
				(Fe XXI	$1s^2 2p^4$	$1s^2 2p^3 4d$	9.6567	2.35	-242 ⁺¹²⁸ ₋₁₂₁
9.6770 ^{+0.0103} _{-0.0102}	26.93 ^{+23.07} _{-18.05}	-5.89 ^{+3.20} _{-3.58}	10.4	Fe XXVI	$2p$	$3d$	9.6745	29.1	77 ⁺³¹⁹ ₋₃₁₆
				(Ni XXV	$1s^2 2p^2$	$1s^2 2p 3d$	9.6887	20.5	-362 ⁺³¹⁸ ₋₃₁₆
				(Ni XXV	$1s^2 2p^2$	$1s^2 2p 3d$	9.6913	15.2	-443 ⁺³¹⁸ ₋₃₁₆
9.7116 ^{+0.0026} _{-0.0018}	0.10 ^{+12.34} _{-0.10}	-4.55 ^{+1.69} _{-1.90}	19.3	(Fe XIX	$2s^2 2p^4$	$2s 2p^2 2p^2 4p$	9.7061	0.015	170 ⁺⁷⁹ ₋₅₅
				(Ne x	$1s$	$4p$	9.7080	0.67	109 ⁺⁷⁹ ₋₅₅
				(Ne x	$1s$	$4p$	9.7085	0.67	96 ⁺⁷⁹ ₋₅₅
9.7275 ^{+0.0064} _{-0.0060}	19.64 ^{+19.64} _{-11.76}	-6.60 ^{+2.82} _{-3.19}	18.2	Fe XIX	$2s 2p^5$	$2s 2p 2p^3 5d$	9.7326	2.73	-159 ⁺¹⁹⁷ ₋₁₈₅
				Fe XX	$2s^2 2p^3$	$2s 2p 2p^2 4p$	9.7242	2.47	99 ⁺¹⁹⁷ ₋₁₈₅
				Fe XX	$2s^2 2p^3$	$2s 2p^2 2p 4p$	9.7269	2.42	16 ⁺¹⁹⁷ ₋₁₈₅
				Fe XX	$2s^2 2p^3$	$2s 2p^2 2p 4p$	9.7269	2.42	16 ⁺¹⁹⁷ ₋₁₈₅
				Fe XIX	$2s 2p^5$	$2s 2p 2p^3 5d$	9.7313	2.28	-120 ⁺¹⁹⁷ ₋₁₈₅
				Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 5d$	9.7327	1.45	-162 ⁺¹⁹⁷ ₋₁₈₅
				Ni XXV	$1s^2 2p^2$	$1s^2 2p 3d$	9.7230	27.1	138 ⁺¹⁹⁷ ₋₁₈₅
10.3125 ^{+0.0199} _{-0.0125}	41.97 ^{+8.03} _{-25.80}	8.89 ^{+5.95} _{-6.30}	8.0	(Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.349	32.7	-1060 ⁺⁵⁷⁸ ₋₃₆₃
				Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.329	20.8	-488 ⁺⁵⁷⁹ ₋₃₆₄
				(Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.297	19.4	444 ⁺⁵⁸⁰ ₋₃₇₈
				(Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.343	14.5	-903 ⁺⁵⁷⁸ ₋₃₆₃
				Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.331	14.2	-550 ⁺⁵⁷⁹ ₋₃₆₄
				Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.4237	29.8	92 ⁺³⁷⁸ ₋₄₃₀
10.4269 ^{+0.0131} _{-0.0149}	19.75 ^{+34.82} _{-19.75}	-5.14 ^{+3.85} _{-4.00}	5.3	(Ni XXIV	$1s^2 2s^2 2p$	$1s^2 2s^2 3d$	10.4410	24.7	-406 ⁺³⁷⁷ ₋₄₂₉
				(Ni XXIV	$1s^2 2s 2p^2$	$1s^2 2s 2p 3d$	10.4116	20.6	440 ⁺³⁷⁹ ₋₄₃₀
				(Fe XVIII	$2s^2 2p^5$	$2s^2 2p^4 5d$	10.5364	2.60	-396 ⁺²¹⁴ ₋₁₄₄
10.5225 ^{+0.0075} _{-0.0051}	0.01 ^{+28.98} _{-0.01}	-4.12 ^{+2.53} _{-1.93}	6.8	(Fe XVIII	$2s^2 2p^5$	$2s^2 2p^2 2p^2 5d$	10.5382	2.25	-448 ⁺²¹⁴ ₋₁₄₄
				(Fe XVIII	$2s^2 2p^5$	$2s^2 2p^2 2p^2 5d$	10.5442	1.22	-618 ⁺²¹⁴ ₋₁₄₄
				(Fe XVIII	$2s^2 2p^5$	$2s 2p^2 2p^3 4p$	10.5640	1.58	-326 ⁺²¹³ ₋₀
10.5525 ^{+0.0075} _{-0.0000}	0.00 ^{+14.78} _{-0.00}	-4.18 ^{+2.60} _{-2.77}	7.0	(Fe XVIII	$2s^2 2p^5$	$2s 2p^2 2p^3 4p$	10.5672	1.39	-417 ⁺²¹³ ₋₀
				(Fe XVIII	$2s^2 2p^5$	$2s 2p^2 2p^3 4p$	10.5672	1.39	266 ⁺¹⁸⁰ ₋₁₇₀
10.5734 ^{+0.0064} _{-0.0060}	0.45 ^{+24.33} _{-0.45}	-5.06 ^{+2.72} _{-2.49}	9.6	(Fe XVIII	$2s^2 2p^5$	$2s 2p^2 2p^3 4p$	10.5640	1.58	175 ⁺¹⁸⁰ ₋₁₇₀
				(Fe XVIII	$2s^2 2p^5$	$2s 2p^2 2p^3 4p$	10.5672	1.39	-34 ⁺¹⁵² ₋₁₀₅
10.6402 ^{+0.0054} _{-0.0037}	4.39 ^{+16.07} _{-4.39}	-6.04 ^{+2.70} _{-3.61}	13.6	Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.6414	5.20	-250 ⁺¹⁵² ₋₁₀₅
				(Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.6491	3.74	-14 ⁺¹⁵² ₋₁₀₅
				Fe XIX	$2s^2 2p^4$	$2s^2 2p 2p^2 4d$	10.6407	1.24	302 ⁺¹⁵² ₋₁₀₅
				(Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.6295	4.78	160 ⁺¹⁸ ₋₈₅
10.6897 ^{+0.0007} _{-0.0030}	0.09 ^{+8.30} _{-0.09}	-8.65 ^{+2.30} _{-2.29}	37.4	(Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.6840	2.28	278 ⁺¹⁸ ₋₈₅
				(Fe XIX	$2s^2 2p^4$	$2s^2 2p^3 4d$	10.6798	0.84	-175 ⁺³⁸⁰ ₋₃₄₈
10.9339 ^{+0.0139} _{-0.0127}	44.77 ^{+5.23} _{-44.77}	16.66 ^{+7.84} _{-10.33}	16.3	Ni XXIII	$1s^2 2p^4$	$1s^2 2p 2p^2 3d$	10.940	28.8	136 ⁺³⁸⁰ ₋₃₄₈
				(Ni XXIII	$1s^2 2p^4$	$1s^2 2p 2p^2 3d$	10.928	26.4	376 ⁺³⁸⁰ ₋₃₄₈
				(Ni XXIII	$1s^2 2p^4$	$1s^2 2p 2p^2 3d$	10.920	11.7	-361 ⁺³⁸⁰ ₋₃₄₈
				Ni XXIII	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 3d$	10.947	8.41	296 ⁺³⁸⁰ ₋₃₄₈
				Fe XIX	$2p^6$	$2p 2p^4 4d$	10.923	8.25	
11.0010 ^{+0.0201} _{-0.0205}	0.00 ^{+50.42} _{-0.00}	-3.08 ^{+3.08} _{-3.45}	2.6						

Table A III.2: List of lines in the ‘dip’ spectrum – sorted by wavelength (continued)

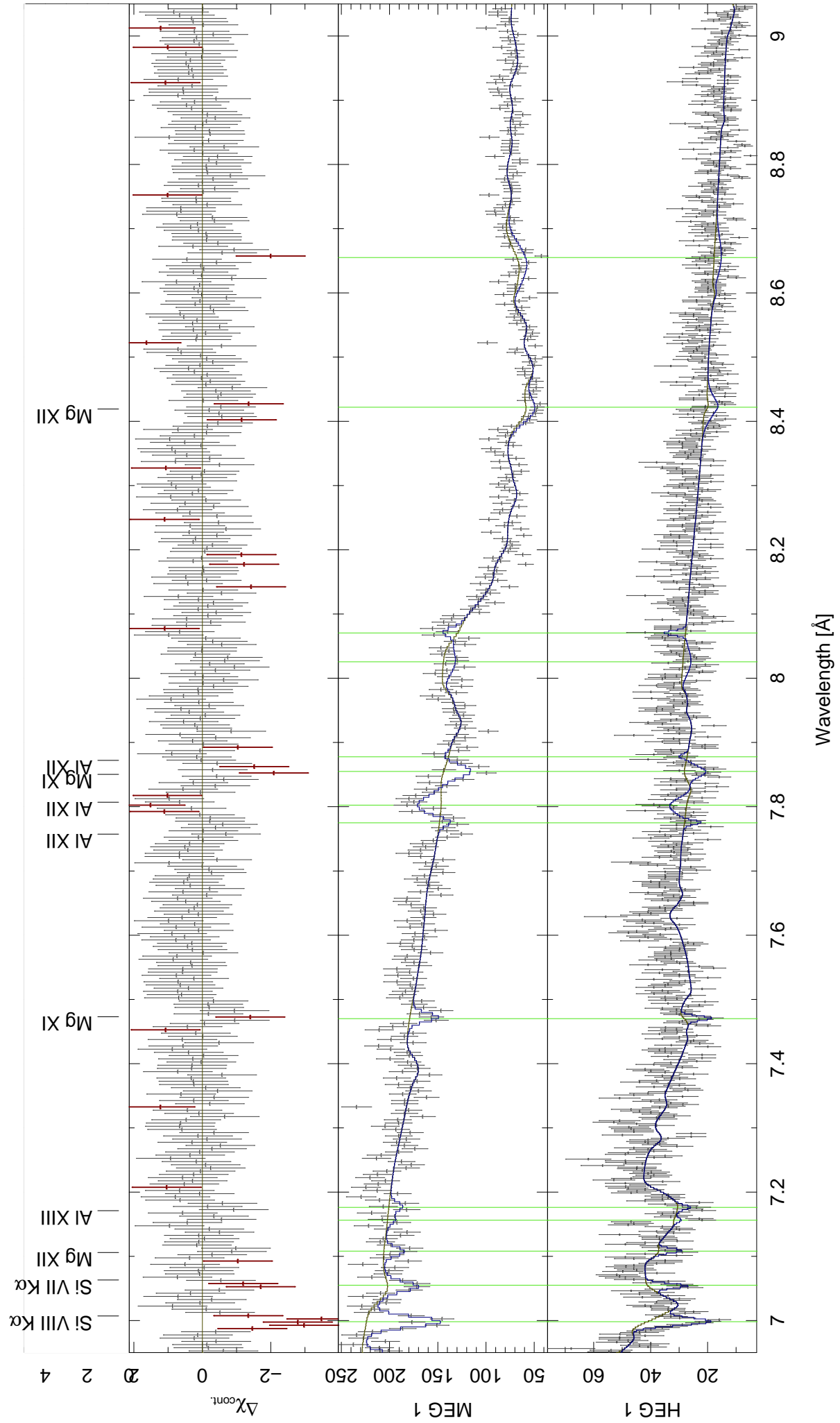
λ [Å]	FWHM [mÅ]	EW [mÅ]	$\Delta\chi^2$	ion <i>i j</i>	transition	λ_0 [Å][10 ¹² s ⁻¹]	A_{ji}	$\Delta\lambda/\lambda \cdot c$ [km/s]	
11.0742 ^{+0.0008} _{-0.0092}	0.01 ^{+24.08} _{-0.01}	-4.77 ^{+3.17} _{-3.15}	6.1	(Ni XXIII	$1s^2 2p^4$	$1s^2 2p 2p^2 3d$	11.0891	13.3	-403 ⁺²² ₋₂₄₉
				(Ni XXIII	$1s^2 2p^4$	$1s^2 2p 2p^2 3d$	11.0951	12.9	-565 ⁺²² ₋₂₄₉
				Ni XXIII	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 3d$	11.0722	12.6	55 ⁺²² ₋₂₄₉
				(Ni XXIII	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 3d$	11.0894	12.3	-410 ⁺²² ₋₂₄₉
11.0974 ^{+0.0026} _{-0.0073}	24.37 ^{+20.55} _{-12.87}	14.75 ^{+6.90} _{-5.94}	16.7						
11.2051 ^{+0.0049} _{-0.0245}	42.98 ^{+7.02} _{-30.24}	11.34 ^{+7.39} _{-7.36}	6.4						
11.5700 ^{+0.0125} _{-0.0200}	12.13 ^{+38.30} _{-12.13}	-5.64 ^{+4.56} _{-6.27}	4.2	(Ne IX	$1s^2$	$1s 3p$	11.5440	2.48	675 ⁺³²⁴ ₋₅₁₉
				Fe XVIII	$2s^2 2p^5$	$2s^2 2p^2 2p^2 4d$	11.5740	1.53	-104 ⁺³²³ ₋₅₁₈
				Ni XXII	$2s 2p^4$	$2s 2p 2p^2 3d$	11.5824	20.6	-320 ⁺³²³ ₋₅₁₇
				Ni XXII	$2s 2p^4$	$2s 2p^3 3d$	11.5589	15.7	288 ⁺³²⁴ ₋₅₁₉
12.0976 ^{+0.0135} _{-0.0095}	28.00 ^{+19.20} _{-28.00}	-15.63 ^{+8.63} _{-7.98}	10.9	(Ne x	$1s$	$2p$	12.1321	6.16	-853 ⁺³³³ ₋₂₃₅
				(Ne x	$1s$	$2p$	12.1375	6.16	-986 ⁺³³³ ₋₂₃₄
12.4652 ^{+0.0098} _{-0.0005}	0.06 ^{+13.28} _{-0.06}	-14.04 ^{+5.65} _{-5.99}	16.6	Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 3d$	12.4656	26.9	-9 ⁺²³⁵ ₋₁₂
				Fe XXI	$1s^2 2s 2p^3$	$1s^2 2s 2p^2 3d$	12.4726	9.00	-177 ⁺²³⁵ ₋₁₂
				Fe XXI	$1s^2 2s 2p^3$	$1s^2 2p 2p^2 3p$	12.4675	5.82	-54 ⁺²³⁵ ₋₁₂
12.8029 ^{+0.0177} _{-0.0223}	0.15 ^{+49.85} _{-0.15}	-3.94 ^{+3.94} _{-10.45}	0.9	(Fe XX	$2s^2 2p^3$	$2s^2 2p 2p 3d$	12.8460	19.2	-1006 ⁺⁴¹³ ₋₅₂₀
				(Fe XX	$2s^2 2p^3$	$2s^2 2p 2p 3d$	12.8240	17.1	-493 ⁺⁴¹⁴ ₋₅₂₁
				Fe XXI	$1s^2 2p^4$	$1s^2 2p^3 3d$	12.7869	28.2	375 ⁺⁴¹⁵ ₋₅₂₃
13.5398 ^{+0.0058} _{-0.0049}	31.33 ^{+12.61} _{-12.19}	129.81 ^{+36.46} _{-35.78}	39.0	(Ne IX	$1s^2$	$1s 2p$	13.550	0.000	-231 ⁺¹²⁸ ₋₁₀₉
				(Ne IX	$1s^2$	$1s 2p$	13.553	0.006	-294 ⁺¹²⁸ ₋₁₀₉
15.3289 ^{+0.0154} _{-0.0154}	49.66 ^{+0.34} _{-29.20}	138.96 ^{+68.53} _{-73.93}	11.4	(Fe XVII	$2s^2 2p^6$	$2s^2 2p^5 3d$	15.261	5.87	1334 ⁺³⁰² ₋₃₀₃
				Fe XIX	$2p^6$	$2p^2 2p^3 3s$	15.334	0.89	-99 ⁺³⁰¹ ₋₃₀₁
				(Fe XIX	$2s 2p^5$	$2s 2p 2p^3 3s$	15.347	0.42	-355 ⁺³⁰¹ ₋₃₀₁
				(Fe XIX	$2s 2p^5$	$2s 2p^2 2p^2 3s$	15.350	0.33	-423 ⁺³⁰¹ ₋₃₀₁



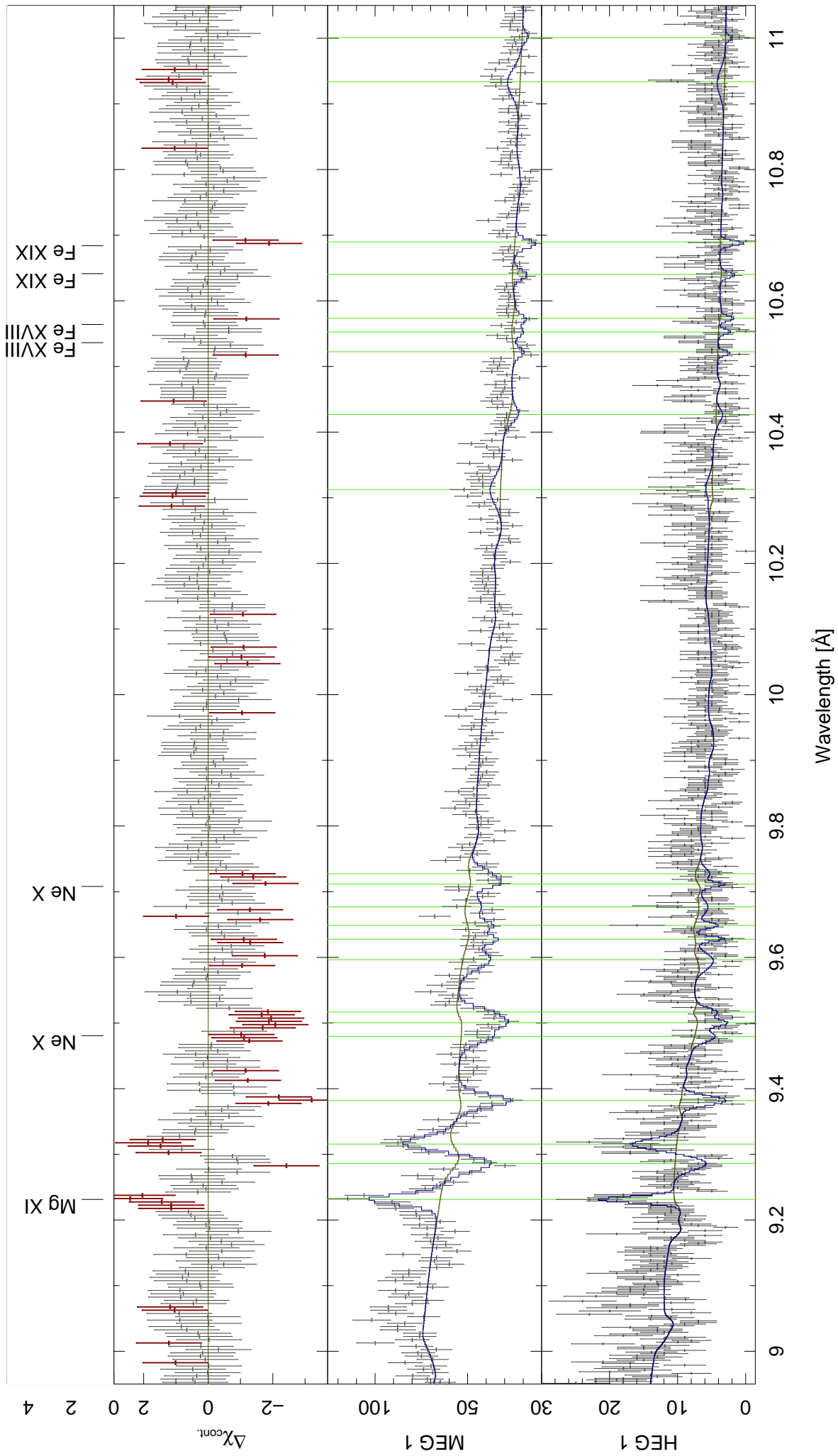
(a) 1...3 Å range of the 'dip' spectrum



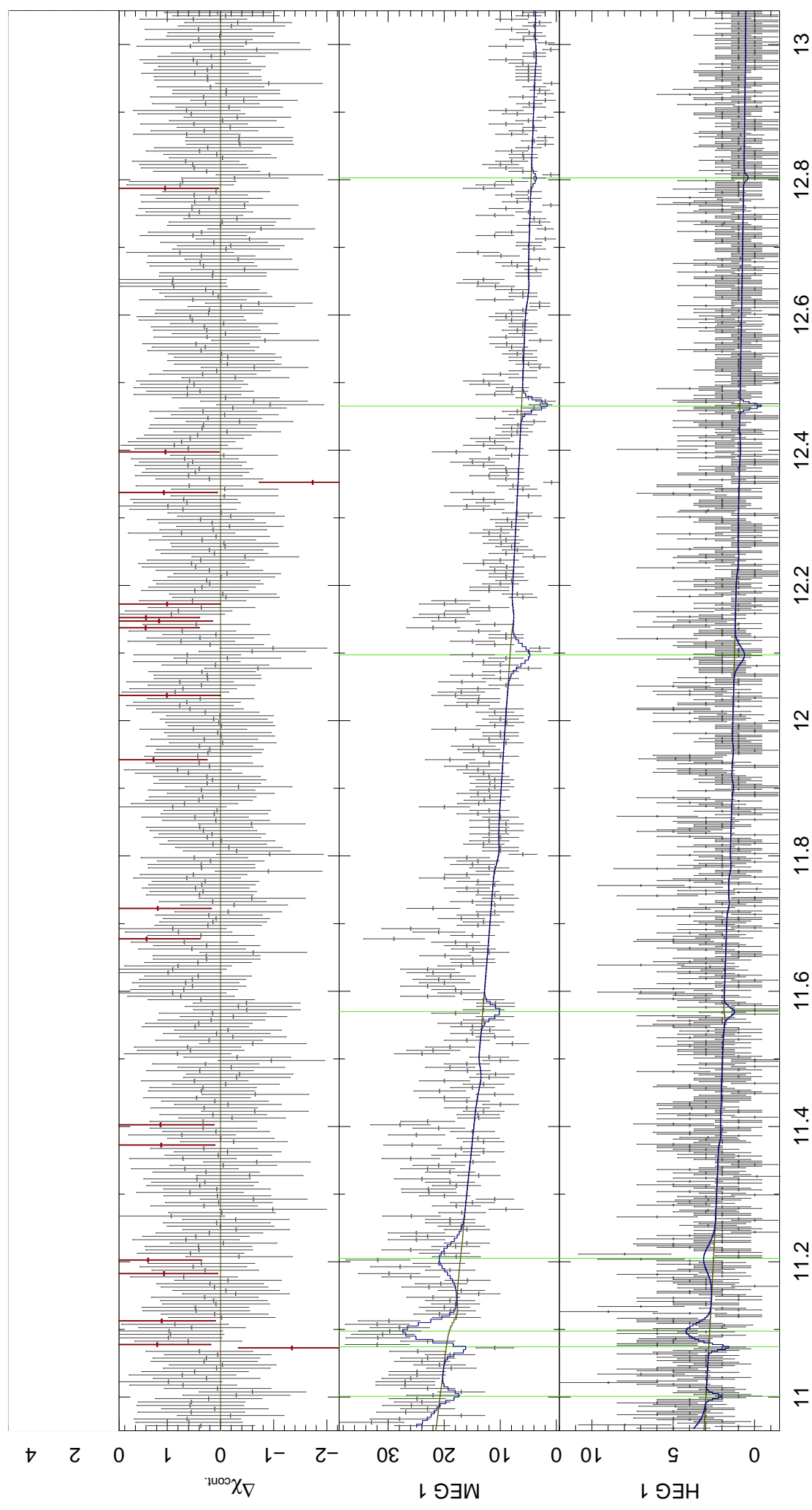
(b) 3...5 Å range of the 'dip' spectrum



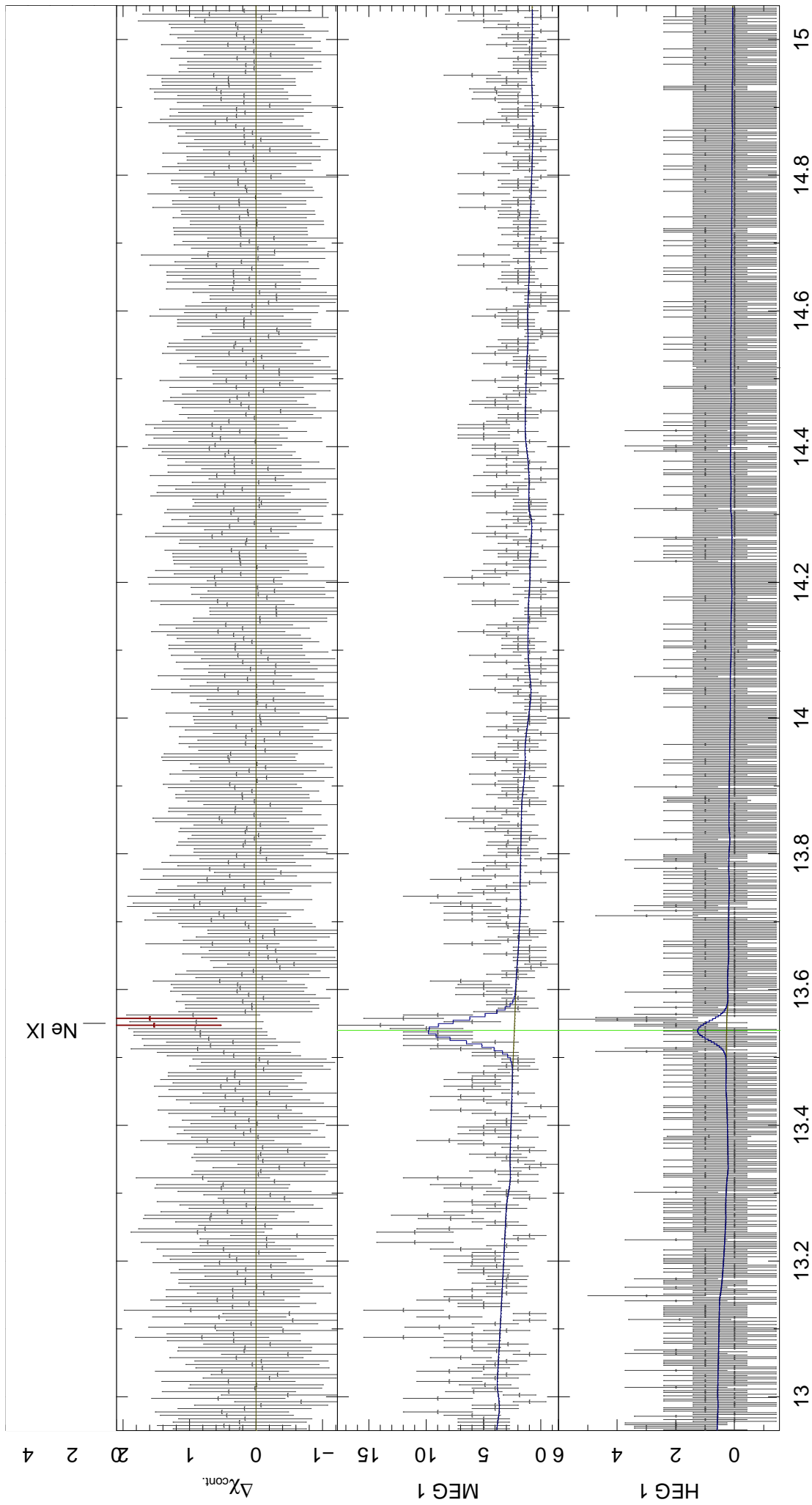
(d) 7...9 Å range of the 'dip' spectrum



(e) 9...11 Å range of the 'dip' spectrum



Wavelength [Å]
 (f) 11...13 Å range of the 'dip' spectrum



(g) 13...15 Å range of the 'dip' spectrum

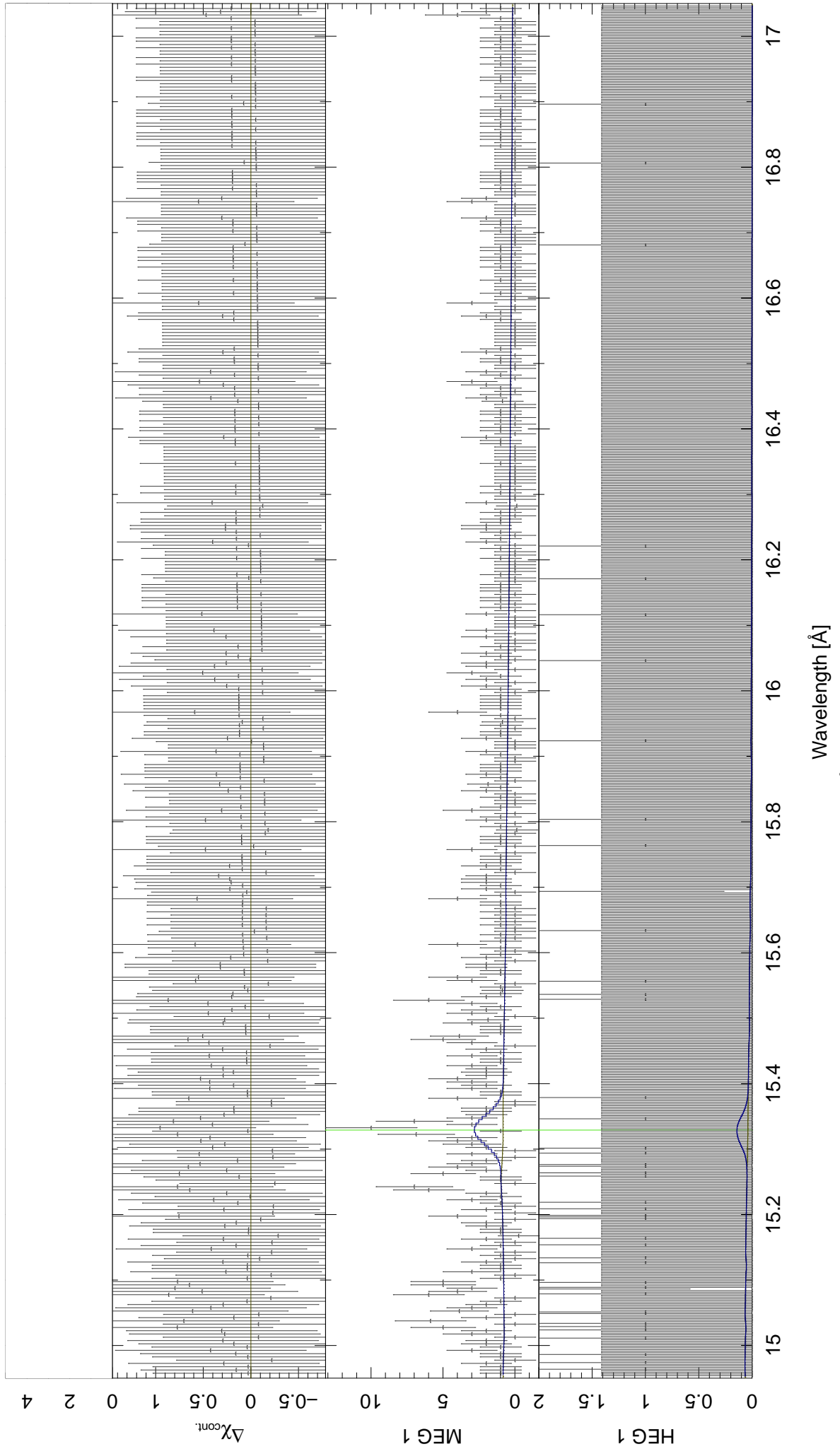


Figure A.III.26: The fitted 'dip' spectrum, including all lines

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DECLARATION

Hereby I declare that I wrote this diploma thesis autonomously and that I have not used other resources than those quoted in this work.

ERKLÄRUNG

Hiermit erkläre ich, dass ich die Diplomarbeit selbständig angefertigt und keine Hilfsmittel außer den in der Arbeit angegebenen benutzt habe.

Bamberg/Madrid, July 2007

(Manfred Hanke)

The references can be found at the end of the main part (before the appendix), on page 114.

Indeed, that's all.
Thanks for reading so far!

| Mh